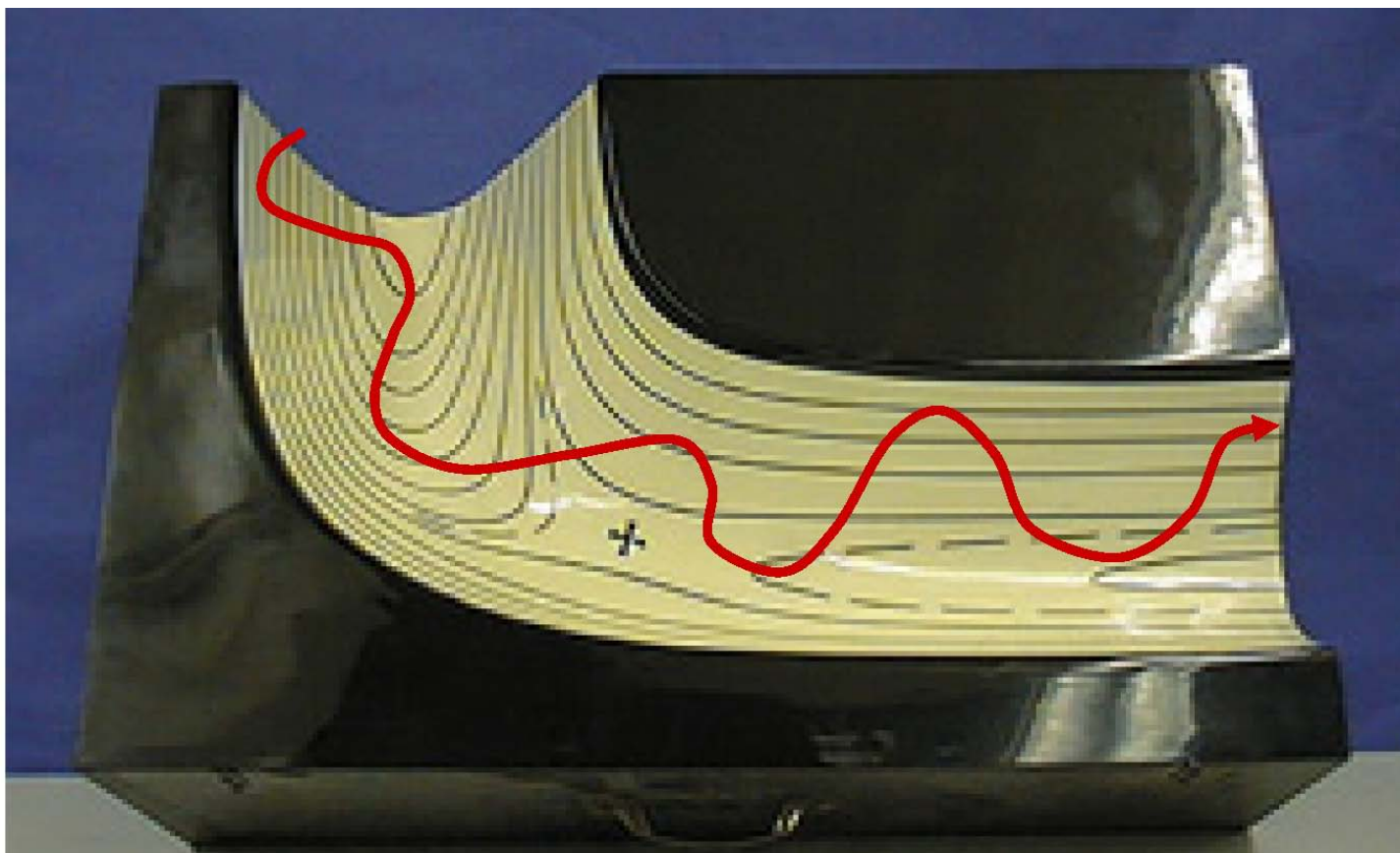


Molecular Dynamics

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1

John

Lance

Natasa

Vinod

Xiaosong

Dufie

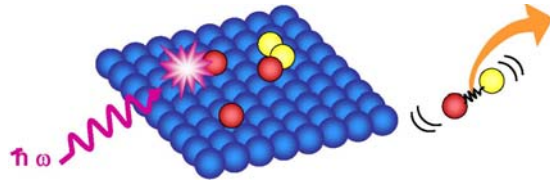
Priya

Sharani

Hongzhi



Tully Group: August, 2004



Prelude: Classical Mechanics

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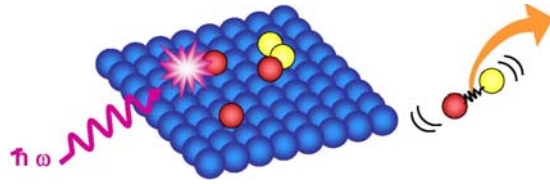
Newton's equations:

$$F = ma = m\ddot{q} = \dot{p}$$

Force is the gradient of the potential:

$$F = -\partial V(q)/\partial q$$

$$\longrightarrow m\ddot{q} = \dot{p} = -\partial V(q)/\partial q \quad \text{Newton's Equations}$$



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Hamilton's equations:

Define total energy = *Hamiltonian*, \mathcal{H} :

$$\dot{q} = \partial\mathcal{H}(p, q)/\partial p$$

$$\dot{p} = -\partial\mathcal{H}(q, p)/\partial q$$

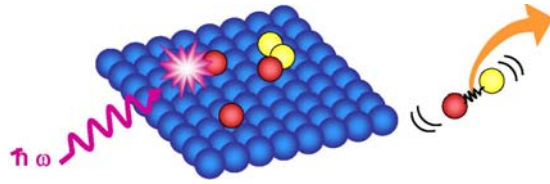
Hamilton's Equations

$$\text{if } \mathcal{H}(p, q) = \mathcal{T}(p) + V(q) = p^2 / 2m + V(q)$$

$$\dot{q} = p / m$$



$$\dot{p} = -\partial V(q)/\partial q$$



Prelude: Classical Mechanics

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Lagrangian Mechanics:

Define *Lagrangian L*:

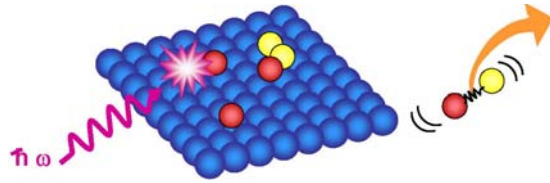
$$\mathcal{L}(q, \dot{q}) = \mathcal{T}(\dot{q}) - V(q)$$

$$\frac{d}{dt} \left(\frac{\partial \mathcal{L}}{\partial \dot{q}} \right) = \frac{\partial \mathcal{L}}{\partial q}$$

Euler-Lagrange Equations

$$\text{if } \mathcal{L}(q, \dot{q}) = \mathcal{T}(\dot{q}) - V(q) = \frac{1}{2} m \dot{q}^2 - V(q)$$

$$\longrightarrow m \ddot{q} = \dot{p} = -\partial V(q) / \partial q$$



Prelude: Classical Mechanics

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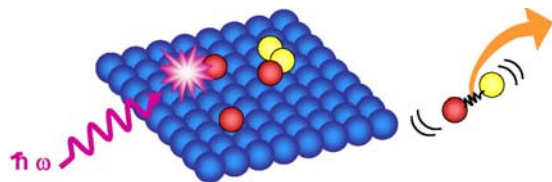
Hamilton-Jacobi Equation:

Define *Hamilton's Principle Function S*:

$$S = \int p dq \quad = \text{“classical action” integral}$$

$$\frac{\partial S}{\partial t} = -\mathcal{H}\left(q, \frac{\partial S}{\partial q}\right) \quad \text{Hamilton-Jacobi Equations}$$

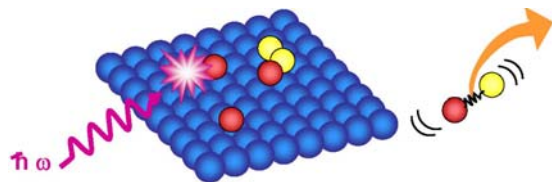
$$\longrightarrow \frac{\partial}{\partial q} \left[\frac{\partial S}{\partial t} + \mathcal{H}\left(q, \frac{\partial S}{\partial q}\right) \right] = \dot{p} + \frac{\partial V(q)}{\partial q} = 0$$



Molecular Dynamics

Park City
June 2005
Tully

- I. The Potential Energy Surface
- II. The Classical Limit via the Bohm Equations
- III. Adiabatic “on-the-fly” Dynamics
- IV. Car-Parrinello Dynamics
- V. Beyond Born Oppenheimer
- VI. Ehrenfest Dynamics
- VII. Surface Hopping
- VIII. Equilibrium in Mixed Quantum-Classical Dynamics
- IX. Mixed Quantum-Classical Nuclear Motion



I. The Potential Energy Surface

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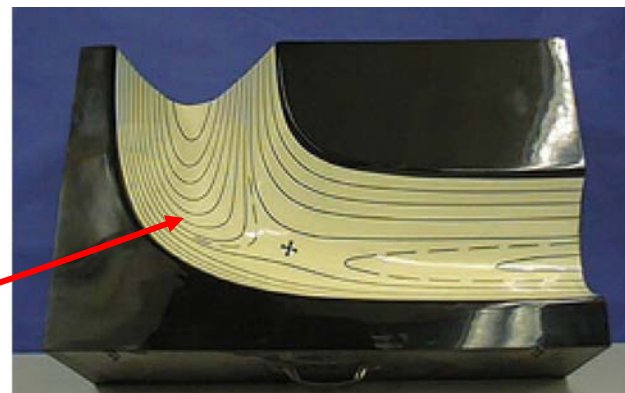
Objective:
$$i \hbar \frac{\partial}{\partial t} \Psi(\mathbf{r}, \mathbf{R}, t) = \mathcal{H}(\mathbf{r}, \mathbf{R}) \Psi(\mathbf{r}, \mathbf{R}, t)$$

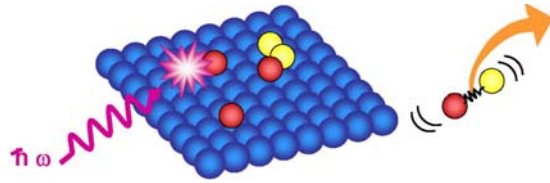
\mathbf{r} = electrons \mathbf{R} = nuclei

$$\mathcal{H} = - \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \nabla_{R_{\alpha}}^2 - \underbrace{\sum_i \frac{\hbar^2}{2m_e} \nabla_r^2 + V(\mathbf{r}, \mathbf{R})}_{\mathcal{H}_{el}(\mathbf{r}; \mathbf{R})} = - \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \nabla_{R_{\alpha}}^2 + \mathcal{H}_{el}(\mathbf{r}; \mathbf{R})$$

$$\mathcal{H}_{el}(\mathbf{r}; \mathbf{R}) \Phi_j(\mathbf{r}; \mathbf{R}) = \mathcal{E}_j(\mathbf{R}) \Phi_j(\mathbf{r}; \mathbf{R})$$

Adiabatic (Born-Oppenheimer)
Potential Energy Surface





I. The Potential Energy Surface

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Born-Oppenheimer Approximation:

$$\Psi(\mathbf{r}, \mathbf{R}, t) \cong \Phi_j(\mathbf{r}; \mathbf{R}) \Omega_j(\mathbf{R}, t)$$

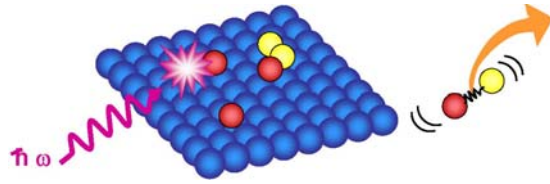
Substitute into TDSE, multiply from left by $\Phi_j^*(\mathbf{r}; \mathbf{R})$, integrate over \mathbf{r} :

$$i\hbar \frac{\partial}{\partial t} \Omega_j(\mathbf{R}, t) = \langle \Phi_j(\mathbf{r}, \mathbf{R}) | \mathcal{H}_{el} | \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \Omega_j(\mathbf{R}, t)$$

$$\left[- \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \langle \Phi_j(\mathbf{r}, \mathbf{R}) | \nabla_{R_{\alpha}}^2 \Phi_j(\mathbf{r}, \mathbf{R}) \Omega_j(\mathbf{R}) \rangle \right]$$

$$= \mathcal{E}_j(\mathbf{R}) \Omega_j(\mathbf{R}, t) - \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \left[\langle \Phi_j(\mathbf{r}, \mathbf{R}) | \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \nabla_{R_{\alpha}}^2$$

$$+ 2 \langle \Phi_j(\mathbf{r}, \mathbf{R}) | \nabla_{R_{\alpha}} \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \nabla_{R_{\alpha}} + \langle \Phi_j(\mathbf{r}, \mathbf{R}) | \nabla_{R_{\alpha}}^2 \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \right] \Omega_j(\mathbf{R}, t)$$



I. The Potential Energy Surface

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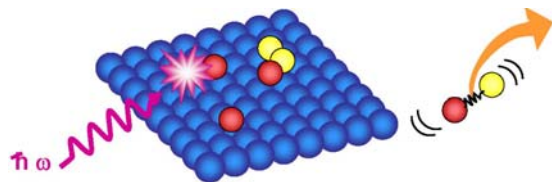
Born-Oppenheimer Approximation:

$$i\hbar \frac{\partial}{\partial t} \Omega_j(\mathbf{R}, t) = \mathcal{E}_j(\mathbf{R}) \Omega_j(\mathbf{R}, t) - \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \left[\langle \Phi_j(\mathbf{r}, \mathbf{R}) | \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \nabla_{R_{\alpha}}^2 + 2 \langle \Phi_j(\mathbf{r}, \mathbf{R}) | \nabla_{R_{\alpha}} \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \nabla_{R_{\alpha}} + \langle \Phi_j(\mathbf{r}, \mathbf{R}) | \nabla_{R_{\alpha}}^2 \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \right] \Omega_j(\mathbf{R}, t)$$

$$\begin{aligned} \nabla_{R_{\alpha}} \langle \Phi_j(\mathbf{r}, \mathbf{R}) | \Phi_j(\mathbf{r}, \mathbf{R}) \rangle &= \nabla_{R_{\alpha}} (1) = 0 \\ &= \langle \Phi_j(\mathbf{r}, \mathbf{R}) \nabla_{R_{\alpha}} \Phi_j(\mathbf{r}, \mathbf{R}) \rangle + \langle \nabla_{R_{\alpha}} \Phi_j(\mathbf{r}, \mathbf{R}) | \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \end{aligned}$$

$$i\hbar \frac{\partial}{\partial t} \Omega_j(\mathbf{R}, t) = \mathcal{E}_j(\mathbf{R}) \Omega_j(\mathbf{R}, t) - \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \nabla_{R_{\alpha}}^2 \Omega_j(\mathbf{R}, t) + \langle \Phi_j(\mathbf{r}, \mathbf{R}) \nabla_{R_{\alpha}}^2 \Phi_j(\mathbf{r}, \mathbf{R}) \rangle \Omega_j(\mathbf{R}, t)$$

???



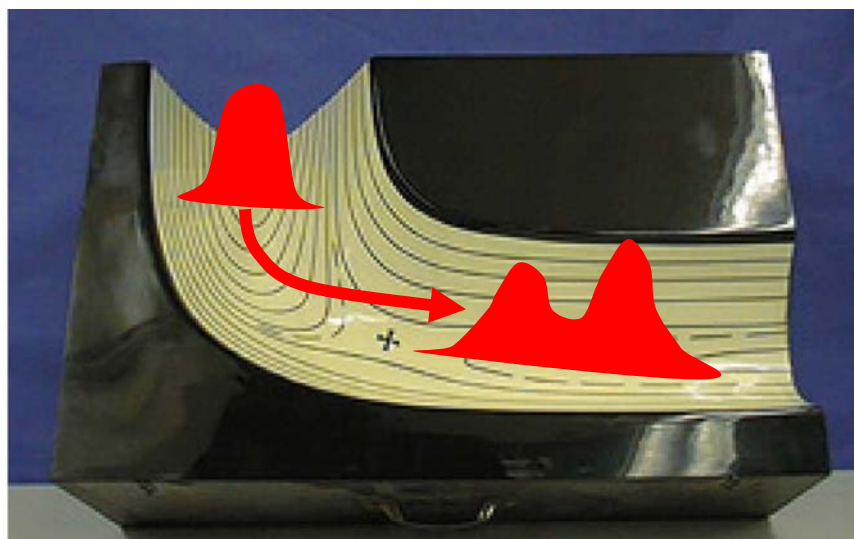
I. The Potential Energy Surface

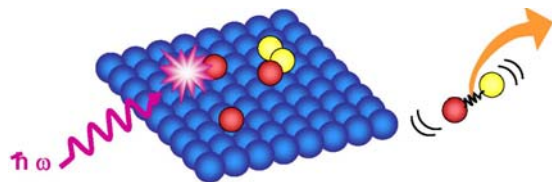
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Born-Oppenheimer Approximation:

$$i\hbar \frac{\partial}{\partial t} \Omega_j(\mathbf{R}, t) = - \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \nabla_{R_{\alpha}}^2 \Omega_j(\mathbf{R}) + \mathcal{E}_j(\mathbf{R}) \Omega_j(\mathbf{R}, t)$$

where $\mathcal{E}_j(\mathbf{R}) = \langle \Phi_j(\mathbf{r}, \mathbf{R}) | \mathcal{H}_{el} | \Phi_j(\mathbf{r}, \mathbf{R}) \rangle$

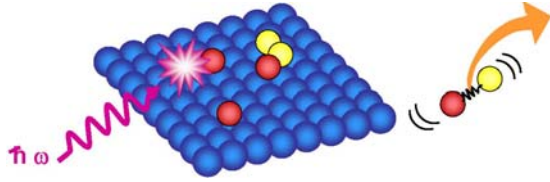




Molecular Dynamics

Park City
June 2005
Tully

- I. The Potential Energy Surface
- II. The Classical Limit via the Bohm Equations**
- III. Adiabatic “on-the-fly” Dynamics
- IV. Car-Parrinello Dynamics
- V. Beyond Born Oppenheimer
- VI. Ehrenfest Dynamics
- VII. Surface Hopping
- VIII. Equilibrium in Mixed Quantum-Classical Dynamics
- IX. Mixed Quantum-Classical Nuclear Motion



II. Classical Limit via Bohm Eqs.

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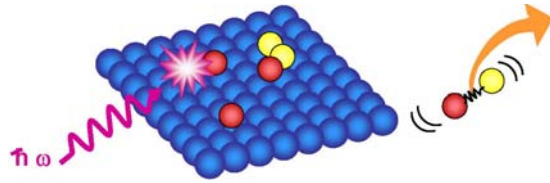
$$i\hbar \frac{\partial}{\partial t} \Omega_j(\mathbf{R}, t) = \left[- \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \nabla_{R_{\alpha}}^2 + \mathcal{E}_j(\mathbf{R}) \right] \Omega_j(\mathbf{R}, t) \quad (1)$$

$$\Omega_j(\mathbf{R}, t) = A_j(\mathbf{R}, t) \exp\left[\frac{i}{\hbar} S_j(\mathbf{R}, t)\right] \quad (2)$$

Substitute (2) into (1) and separate real and imaginary parts:

$$\rightarrow \dot{S}_j = - \sum_{\alpha} \frac{1}{2M_{\alpha}} [\nabla_{R_{\alpha}} S_j]^2 - \mathcal{E}_j(\mathbf{R}) - \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \frac{\nabla_{R_{\alpha}}^2 A_j}{A_j} \quad (3)$$

$$\rightarrow \dot{A}_j = \sum_{\alpha} \frac{1}{2M_{\alpha}} \left\{ 2[\nabla_{R_{\alpha}} A_j] \cdot [\nabla_{R_{\alpha}} S_j] - A_j \nabla_{R_{\alpha}}^2 S_j \right\} \quad (4)$$



II. Classical Limit via Bohm Eqs.

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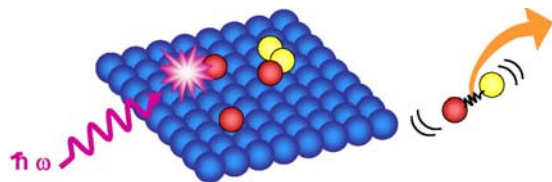
Compare Eq. (3) with Hamilton-Jacobi Equation:

$$\frac{\partial S}{\partial t} = -\mathcal{H}\left(q, \frac{\partial S}{\partial q}\right) \quad \text{where} \quad \frac{\partial S}{\partial q} = p$$

$$\dot{S}_j = -\sum_{\alpha} \frac{1}{2M_{\alpha}} [\nabla_{R_{\alpha}} S_j]^2 - \mathcal{E}_j(\mathbf{R}) - \sum_{\alpha} \frac{\hbar^2}{2M_{\alpha}} \frac{\nabla_{R_{\alpha}}^2 A_j}{A_j} \quad (3)$$

the “quantum potential”

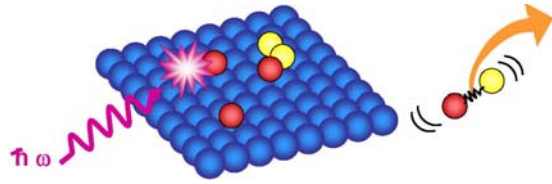
$$\hbar \rightarrow 0: \quad \dot{S}_j = -\sum_{\alpha} \frac{1}{2M_{\alpha}} [\nabla_{R_{\alpha}} S_j]^2 - \mathcal{E}_j(\mathbf{R})$$



Molecular Dynamics

Park City
June 2005
Tully

- I. The Potential Energy Surface
- II. The Classical Limit via the Bohm Equations
- III. Adiabatic “on-the-fly” Dynamics**
- IV. Car-Parrinello Dynamics
- V. Beyond Born Oppenheimer
- VI. Ehrenfest Dynamics
- VII. Surface Hopping
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- IX. Mixed Quantum-Classical Nuclear Motion



III. Adiabatic “on-the-fly” Dynamics

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The Hellman – Feynman Theorem:

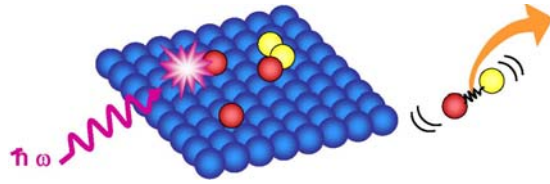
$$\frac{d}{dR} \mathcal{E}_j(R) = \frac{d}{dR} \langle \Phi_j(R) | \mathcal{H}_{el}(R) | \Phi_j(R) \rangle$$

subject to $\langle \Phi_j(R) | \Phi_j(R) \rangle = 1$

and $\mathcal{H}_{el}(R) | \Phi_j(R) \rangle = \mathcal{E}_j(R) | \Phi_j(R) \rangle$



$$\begin{aligned} \frac{d}{dR} \mathcal{E}_j(R) &= \left\langle \Phi_j(R) \left| \frac{d\mathcal{H}_{el}(R)}{dR} \right| \Phi_j(R) \right\rangle \\ &+ \left\langle \frac{d}{dR} \Phi_j(R) \left| \mathcal{H}_{el}(R) \right| \Phi_j(R) \right\rangle + \left\langle \Phi_j(R) \left| \mathcal{H}_{el}(R) \right| \frac{d}{dR} \Phi_j(R) \right\rangle \end{aligned}$$



III. Adiabatic “on-the-fly” Dynamics

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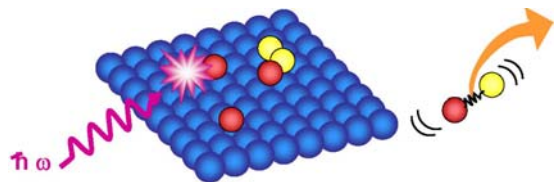
The Hellman – Feynman Theorem:

$$\frac{d}{dR} \mathcal{E}_j(R) = \left\langle \Phi_j(R) \left| \frac{d\mathcal{H}_{el}(R)}{dR} \right| \Phi_j(R) \right\rangle + \left\langle \frac{d}{dR} \Phi_j(R) \left| \mathcal{H}_{el}(R) \right| \Phi_j(R) \right\rangle + \left\langle \Phi_j(R) \left| \mathcal{H}_{el}(R) \right| \frac{d}{dR} \Phi_j(R) \right\rangle$$

$$\boxed{\frac{d}{dR} \mathcal{E}_j(R) = \left\langle \Phi_j(R) \left| \frac{d\mathcal{H}_{el}(R)}{dR} \right| \Phi_j(R) \right\rangle}$$

$$+ \mathcal{E}_j(R) \left[\left\langle \frac{d}{dR} \Phi_j(R) \left| \Phi_j(R) \right\rangle + \left\langle \Phi_j(R) \left| \frac{d}{dR} \Phi_j(R) \right\rangle \right]$$

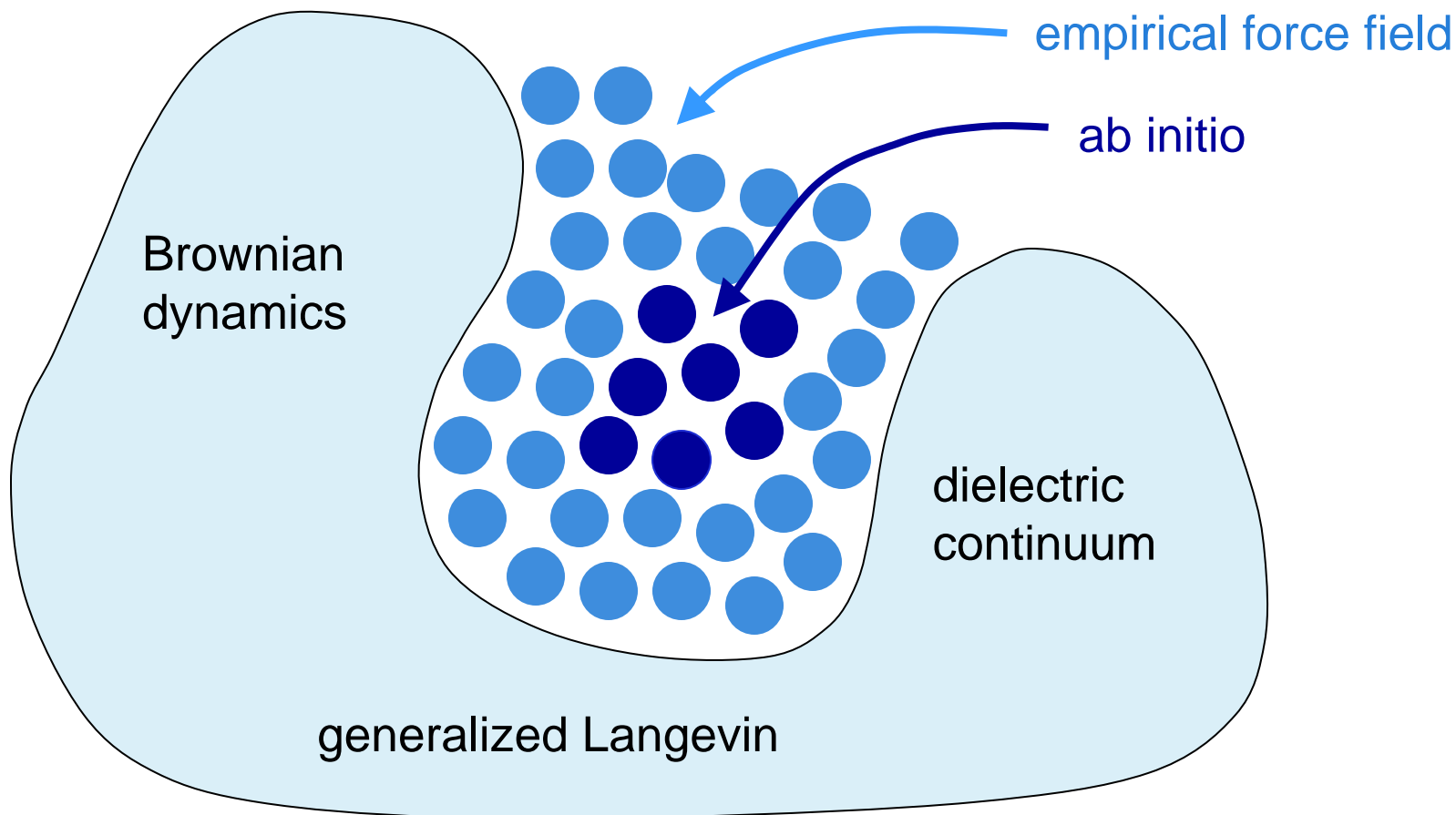
$$= \frac{d}{dR} \left[\langle \Phi_j(R) | \Phi_j(R) \rangle \right] = 0$$

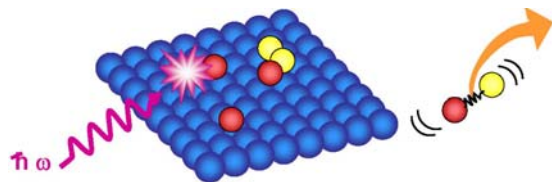


III. Adiabatic “on-the-fly” Dynamics

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QM/MM Approach:

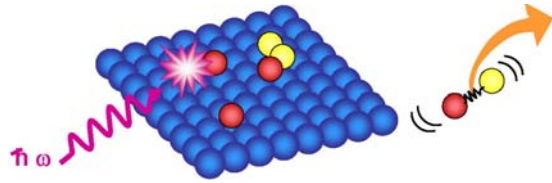




Molecular Dynamics

Park City
June 2005
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- I. The Potential Energy Surface
- II. The Classical Limit via the Bohm Equations
- III. Adiabatic “on-the-fly” Dynamics
- IV. Car-Parrinello Dynamics**
- V. Beyond Born Oppenheimer
- VI. Ehrenfest Dynamics
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- VIII. Equilibrium in Mixed Quantum-Classical Dynamics
- IX. Mixed Quantum-Classical Nuclear Motion



IV. Car-Parrinello Dynamics

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Euler-Lagrange equations:
$$\frac{d}{dt} \frac{d\mathcal{L}}{dq} = \frac{\partial \mathcal{L}}{\partial q}$$

Classical Lagrangian:
$$\mathcal{L} = T - V = \sum_{\alpha} \frac{1}{2} M_{\alpha} \dot{R}^2 - \mathcal{E}_j(\mathbf{R})$$

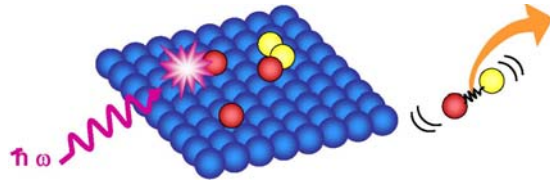
$$\rightarrow M_{\alpha} \ddot{R}_{\alpha} = -\partial \mathcal{E}_j(\mathbf{R}) / \partial R_{\alpha}$$

Car-Parrinello Lagrangian:
$$\mathcal{L} = \sum_{\alpha} \frac{1}{2} M_{\alpha} \dot{R}^2 - \langle \Psi_j | \mathcal{H}_{el} | \Psi_j \rangle$$

$$+ \sum_n \frac{1}{2} \mu_n \langle \dot{\phi}_n | \dot{\phi}_n \rangle + \sum_{nm} \lambda_{nm} [\langle \phi_n | \phi_m \rangle - \delta_{nm}]$$

$$\rightarrow M_{\alpha} \ddot{R}_{\alpha} = -\partial \langle \Psi_j | \mathcal{H}_{el} | \Psi_j \rangle / \partial R_{\alpha}$$

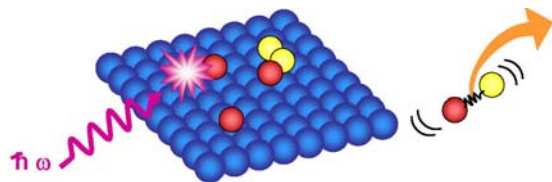
$$\rightarrow \mu_n \ddot{\phi}_n = -\delta \langle \Psi_j | \mathcal{H}_{el} | \Psi_j \rangle / \delta \phi_n^* + \sum_m \lambda_{nm} \phi_m$$



Molecular Dynamics

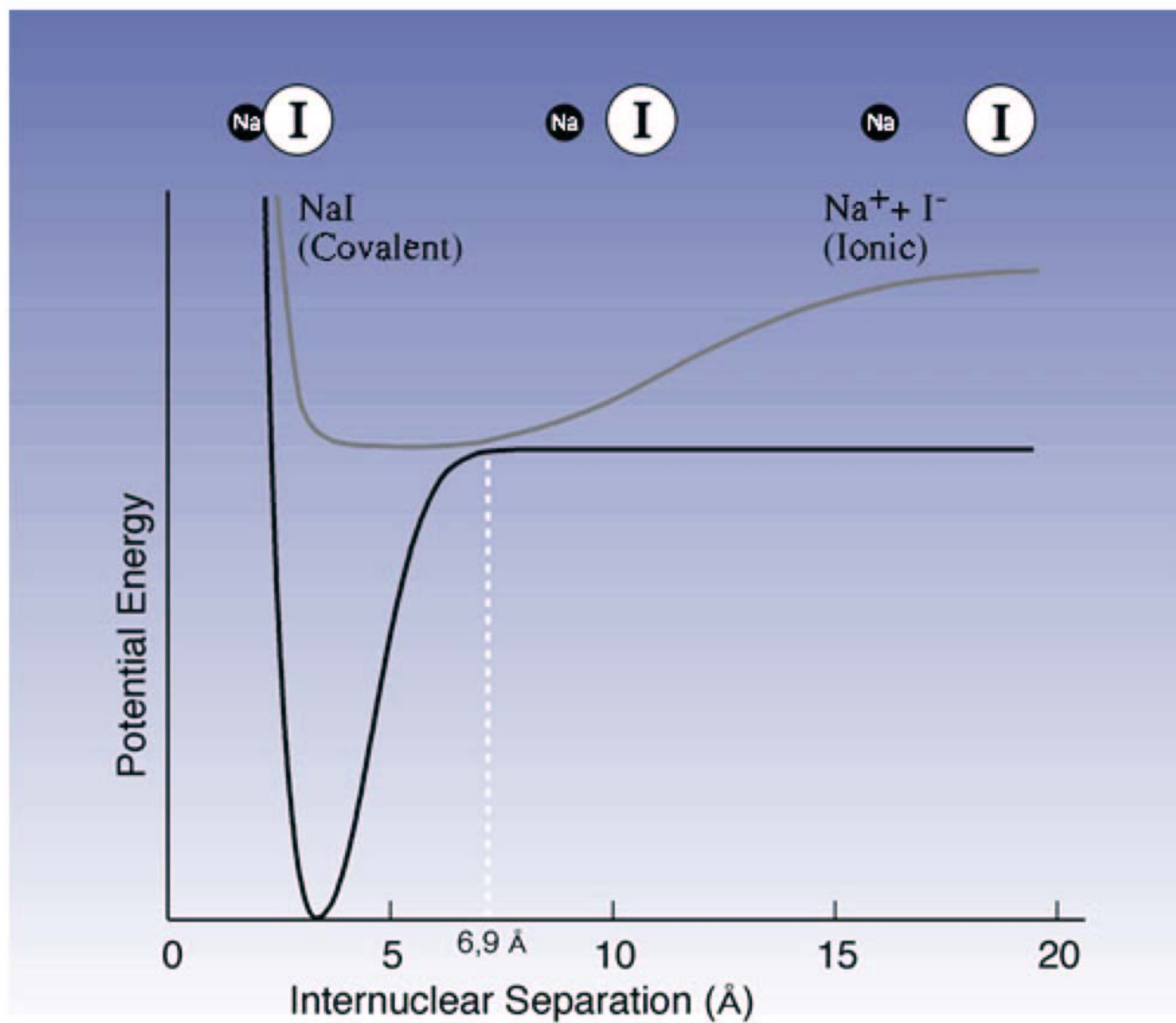
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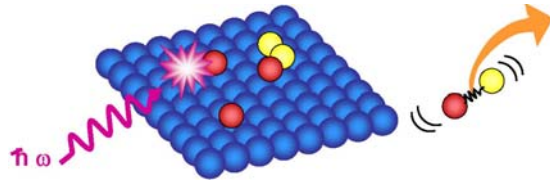
- I. The Potential Energy Surface
- II. The Classical Limit via the Bohm Equations
- III. Adiabatic “on-the-fly” Dynamics
- IV. Car-Parrinello Dynamics
- V. Beyond Born Oppenheimer**
- VI. Ehrenfest Dynamics
- VII. Surface Hopping
- VIII. Equilibrium in Mixed Quantum-Classical Dynamics
- IX. Mixed Quantum-Classical Nuclear Motion



V. Beyond Born-Oppenheimer

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V. Beyond Born-Oppenheimer

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$$\Psi(\mathbf{r}, \mathbf{R}) \approx \Phi_j(\mathbf{r}; \mathbf{R}) \Omega_j(\mathbf{R}) \quad \rightarrow \quad \Psi(\mathbf{r}, \mathbf{R}) = \sum_i \Phi_i(\mathbf{r}; \mathbf{R}) \Omega_i(\mathbf{R})$$

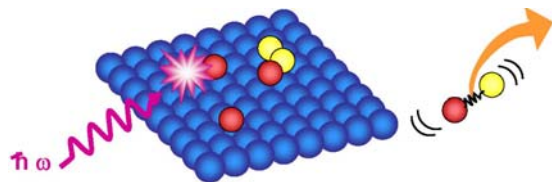
Substitute into TISE, multiply from left by $\Phi_j^*(\mathbf{r}; \mathbf{R})$ integrate over \mathbf{r} :

$$\begin{aligned} -\frac{\hbar^2}{2} \sum_{\alpha} M_{\alpha}^{-1} \nabla_{R_{\alpha}}^2 \Omega_j(\mathbf{R}) + \mathcal{E}_j(\mathbf{R}) \Omega_j(\mathbf{R}) - E \Omega_j(\mathbf{R}) = \\ -\frac{\hbar^2}{2} \sum_i D_{ji}(\mathbf{R}) \Omega_i(\mathbf{R}) + \hbar^2 \sum_{i \neq j} \mathbf{d}_{ji}(\mathbf{R}) \cdot \nabla_{R_{\alpha}} \Omega_i(\mathbf{R}) \end{aligned}$$

where nonadiabatic (derivative) couplings are defined by:

$$\mathbf{d}_{ij}(\mathbf{R}) = -\sum_{\alpha} M_{\alpha}^{-1} \int \left\{ \Phi_i^*(\mathbf{r}, \mathbf{R}) \left[\nabla_{R_{\alpha}} \Phi_j(\mathbf{r}, \mathbf{R}) \right] \right\} d\mathbf{r}$$

$$D_{ij}(\mathbf{R}) = -\sum_{\alpha} M_{\alpha}^{-1} \int \left\{ \Phi_i^*(\mathbf{r}, \mathbf{R}) \left[\nabla_{R_{\alpha}}^2 \Phi_j(\mathbf{r}, \mathbf{R}) \right] \right\} d\mathbf{r}$$

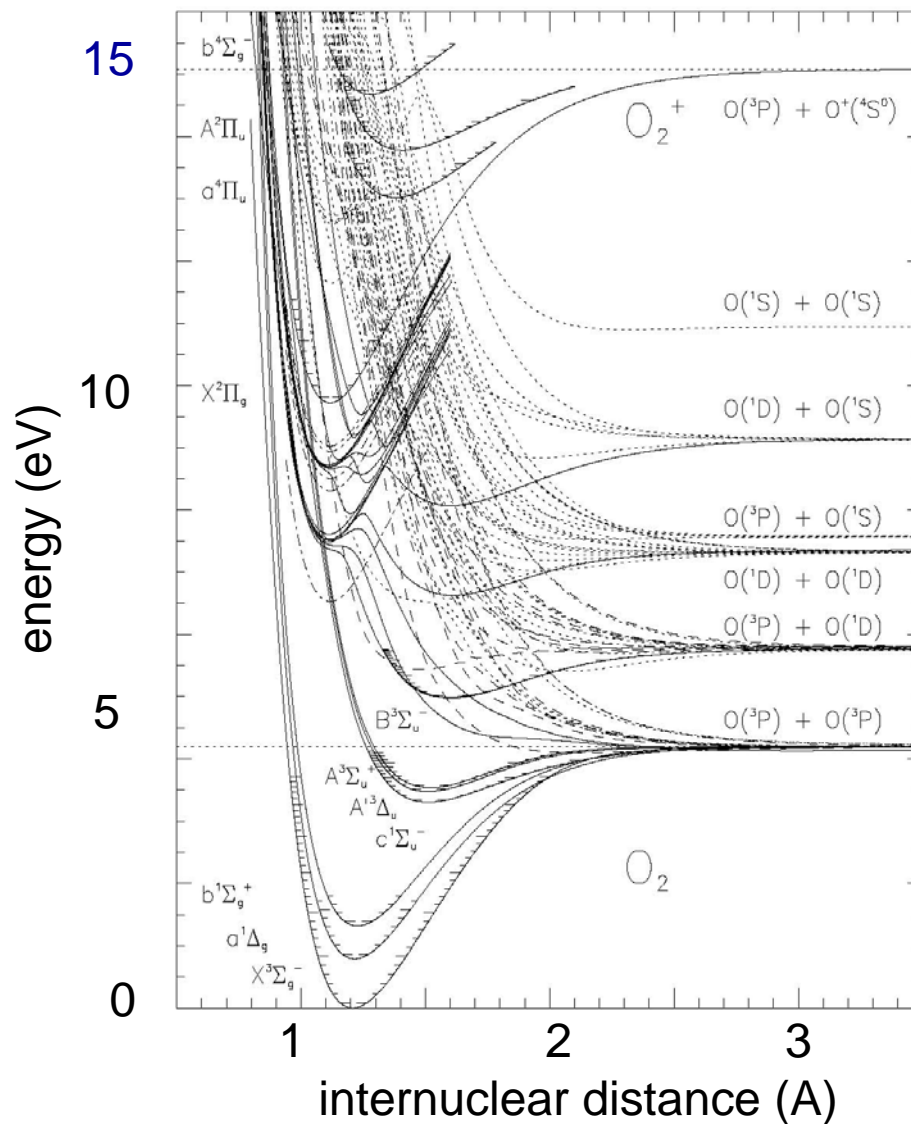


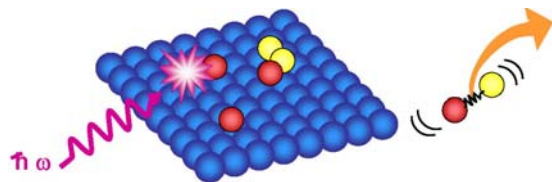
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Potential Energy Curves for the Oxygen Molecule

from
R. P. Saxon and B. Liu,
J. Chem. Phys. 1977

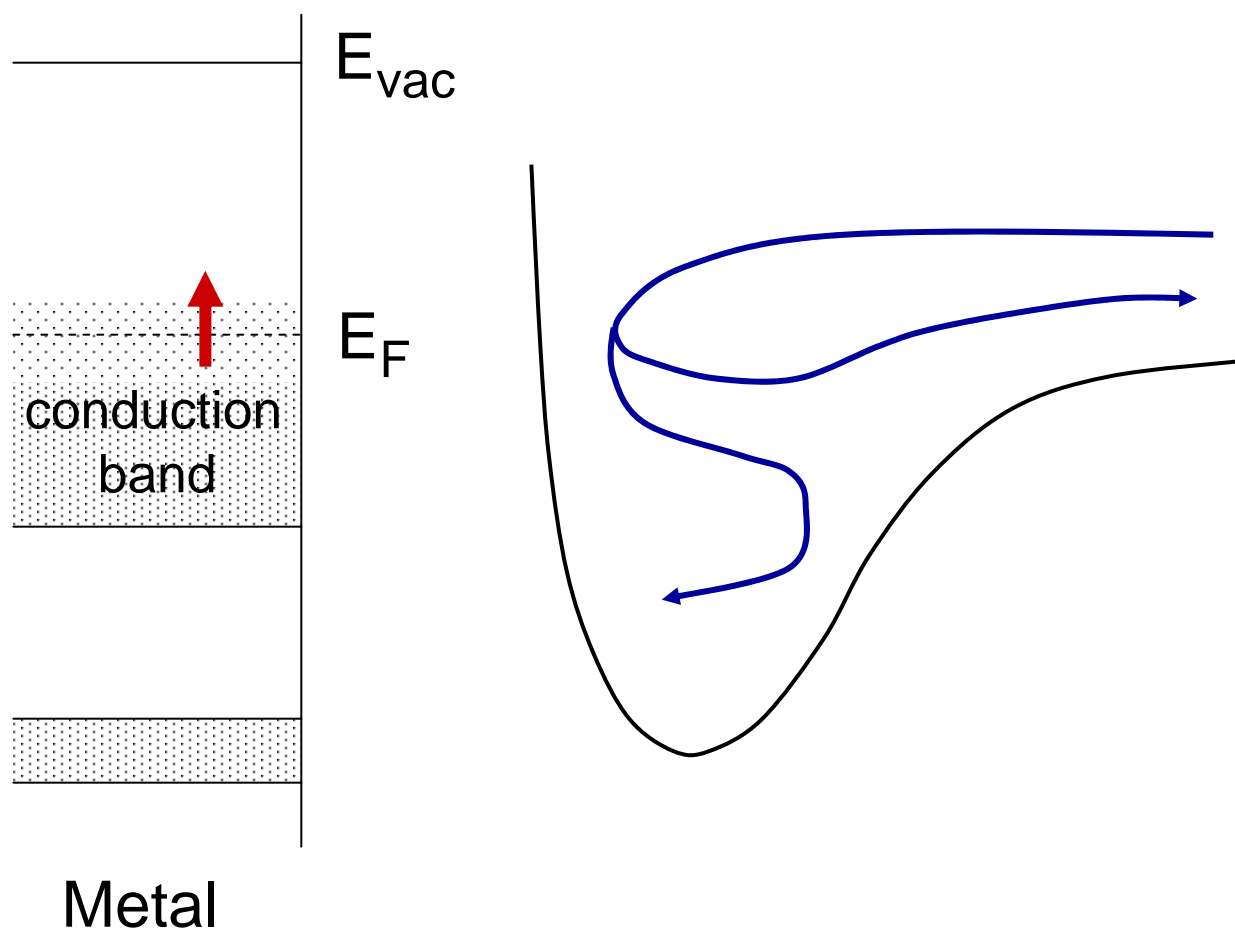


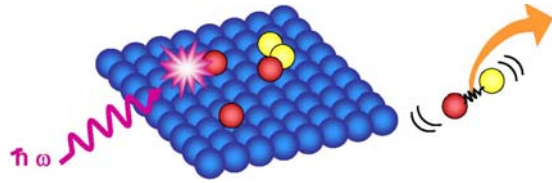


V. Beyond Born-Oppenheimer

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Nonadiabatic Transitions at Metal Surfaces: Electron-Hole Pairs



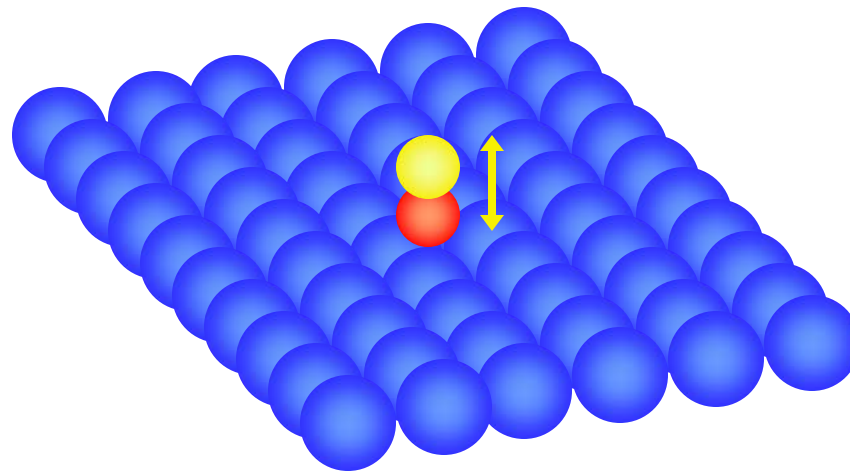


V. Beyond Born-Oppenheimer

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Vibrational Lifetime of CO on Cu(100)

$$\nu = 1 \longrightarrow \nu = 0$$

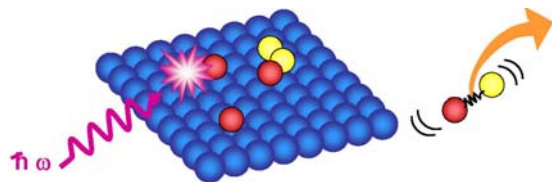


Molecular Dynamics:

$$t \sim 10^{-3} \text{ s.}$$

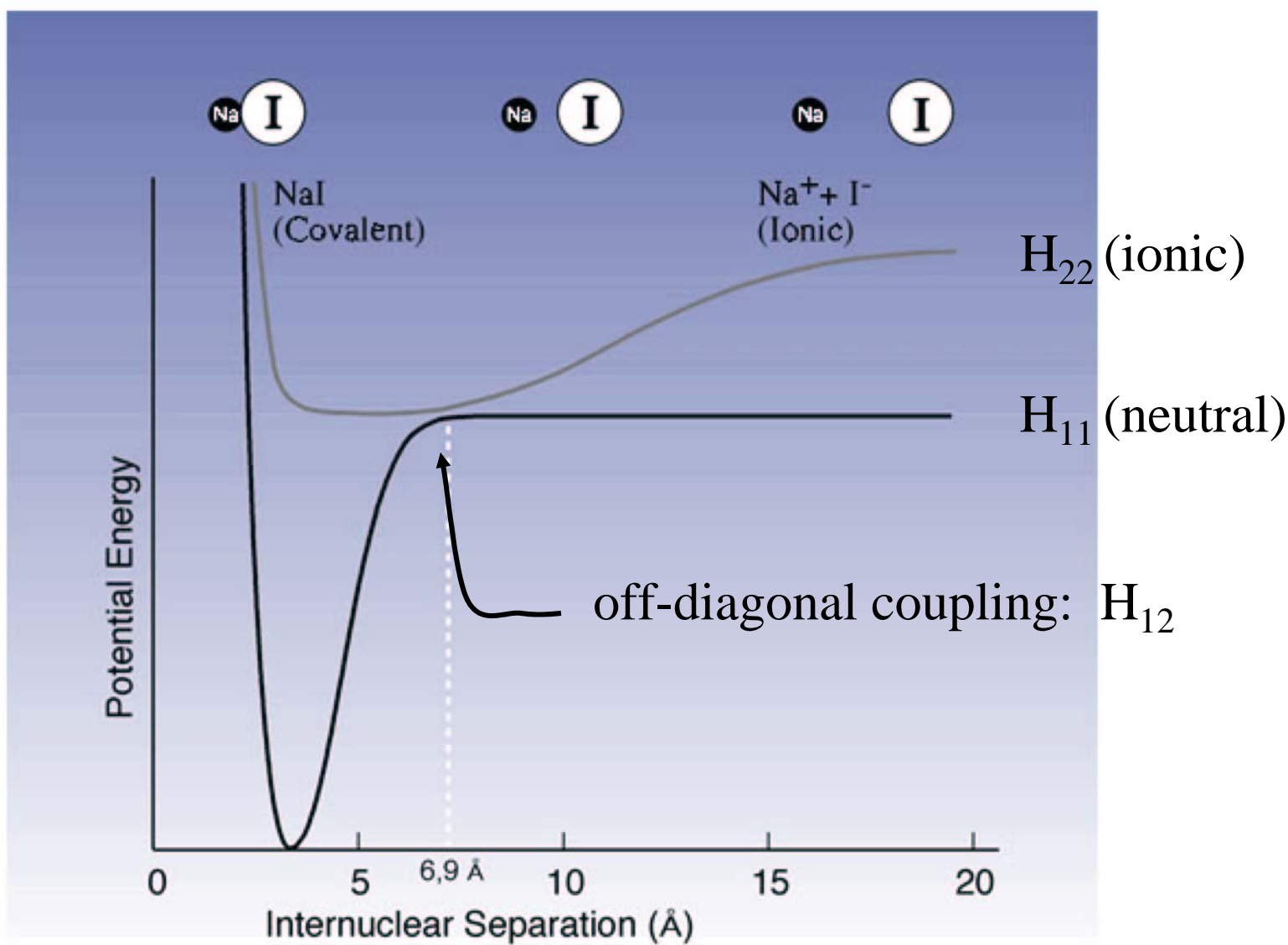
Experiment (A. Harris et al.):

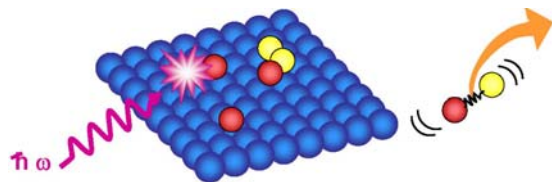
$$t = 2.5 \text{ ps.}$$



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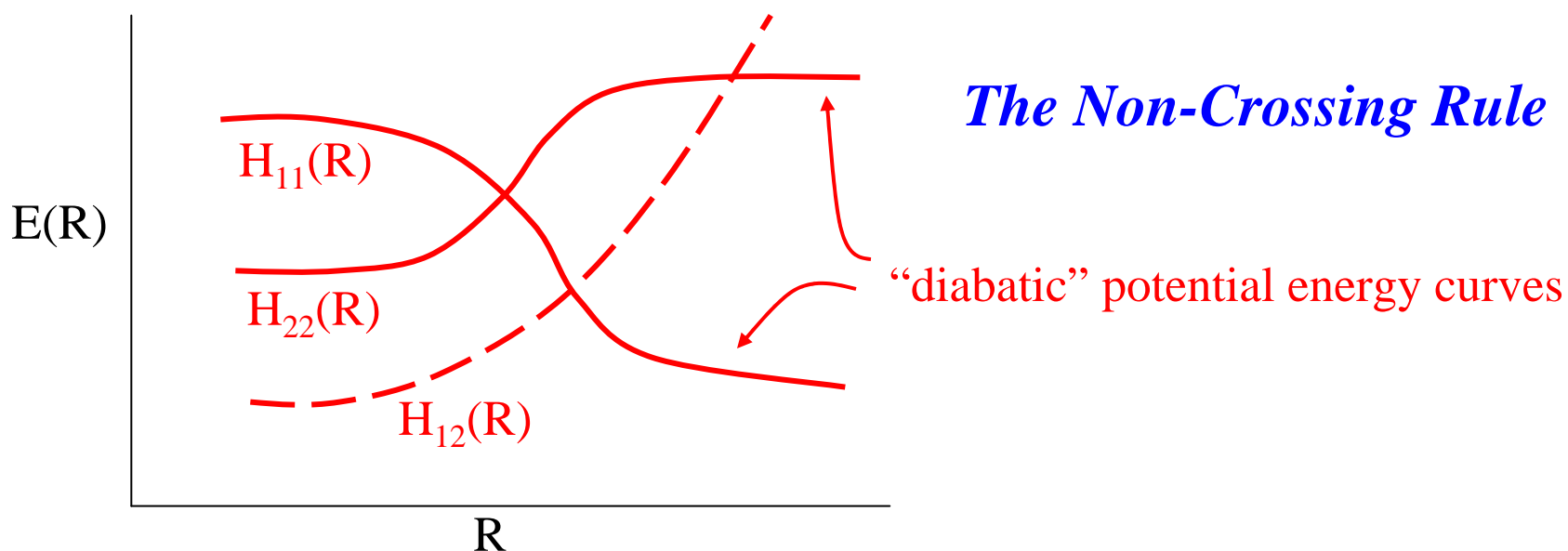




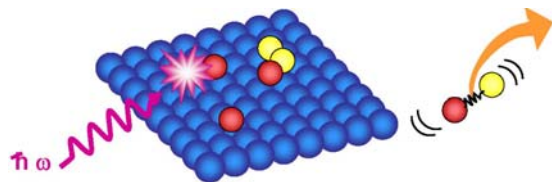
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$$\mathcal{H}(R) = \begin{bmatrix} H_{11}(R) & H_{12}(R) \\ H_{12}(R) & H_{22}(R) \end{bmatrix}$$



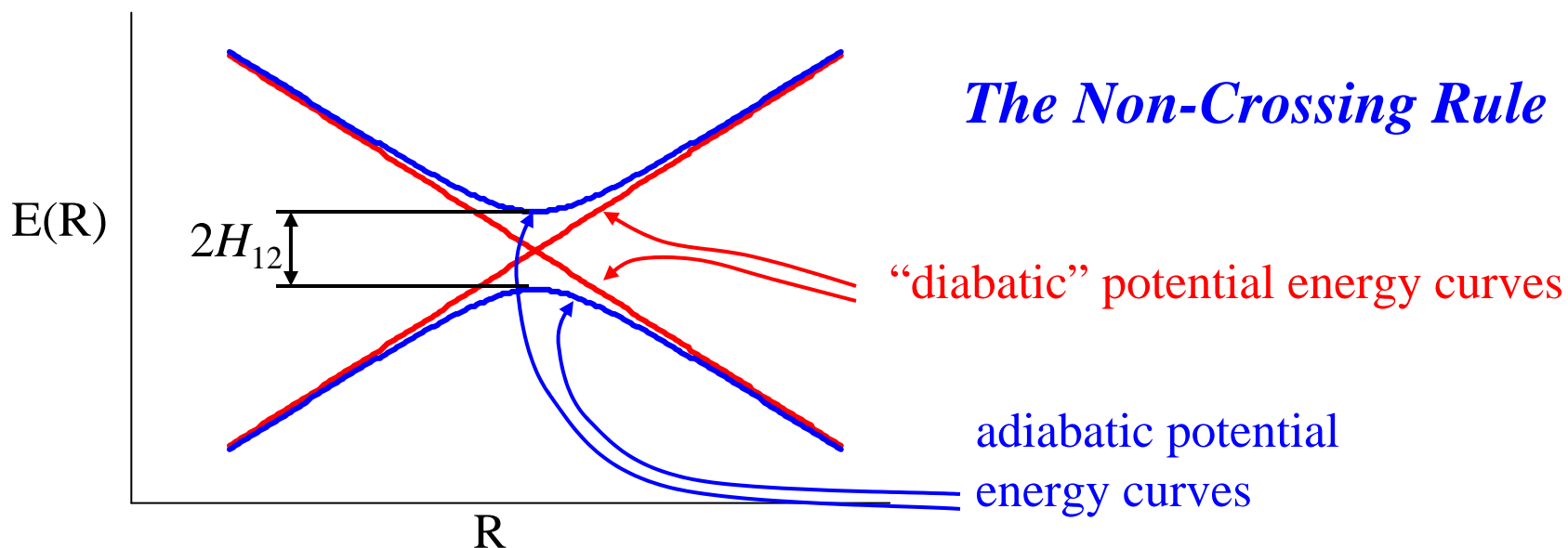
$$\mathcal{E}_{\pm}(R) = \frac{H_{11}(R) + H_{22}(R)}{2} \pm \frac{1}{2} \sqrt{[H_{11}(R) - H_{22}(R)]^2 + 4[H_{12}(R)]^2}$$



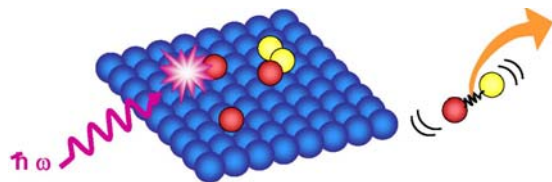
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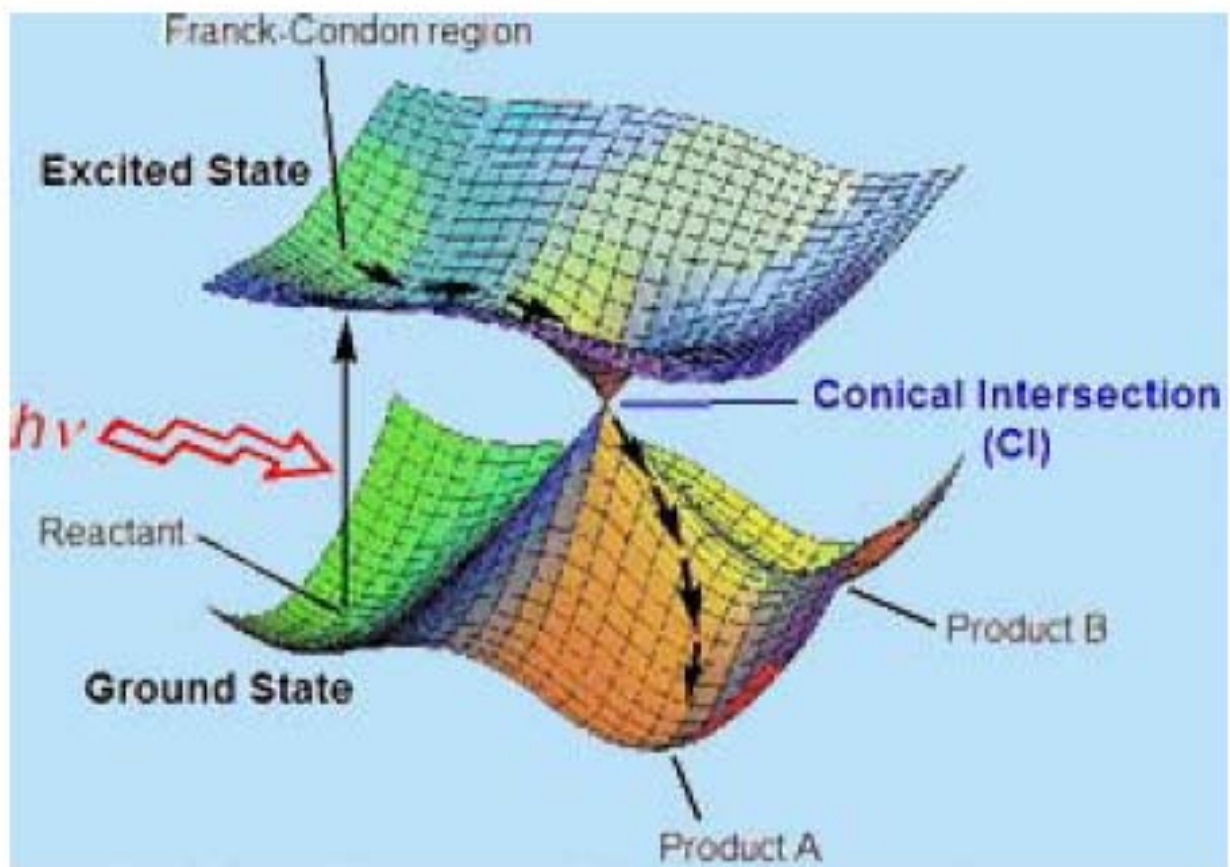


$$\mathcal{E}_{\pm}(R) = \frac{H_{11}(R) + H_{22}(R)}{2} \pm \frac{1}{2} \sqrt{[H_{11}(R) - H_{22}(R)]^2 + 4[H_{12}(R)]^2}$$



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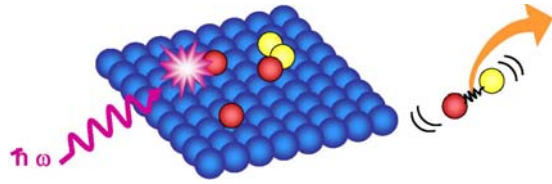


The non-crossing rule
for more than 1 degree
of freedom:

“*Conical Intersection*”

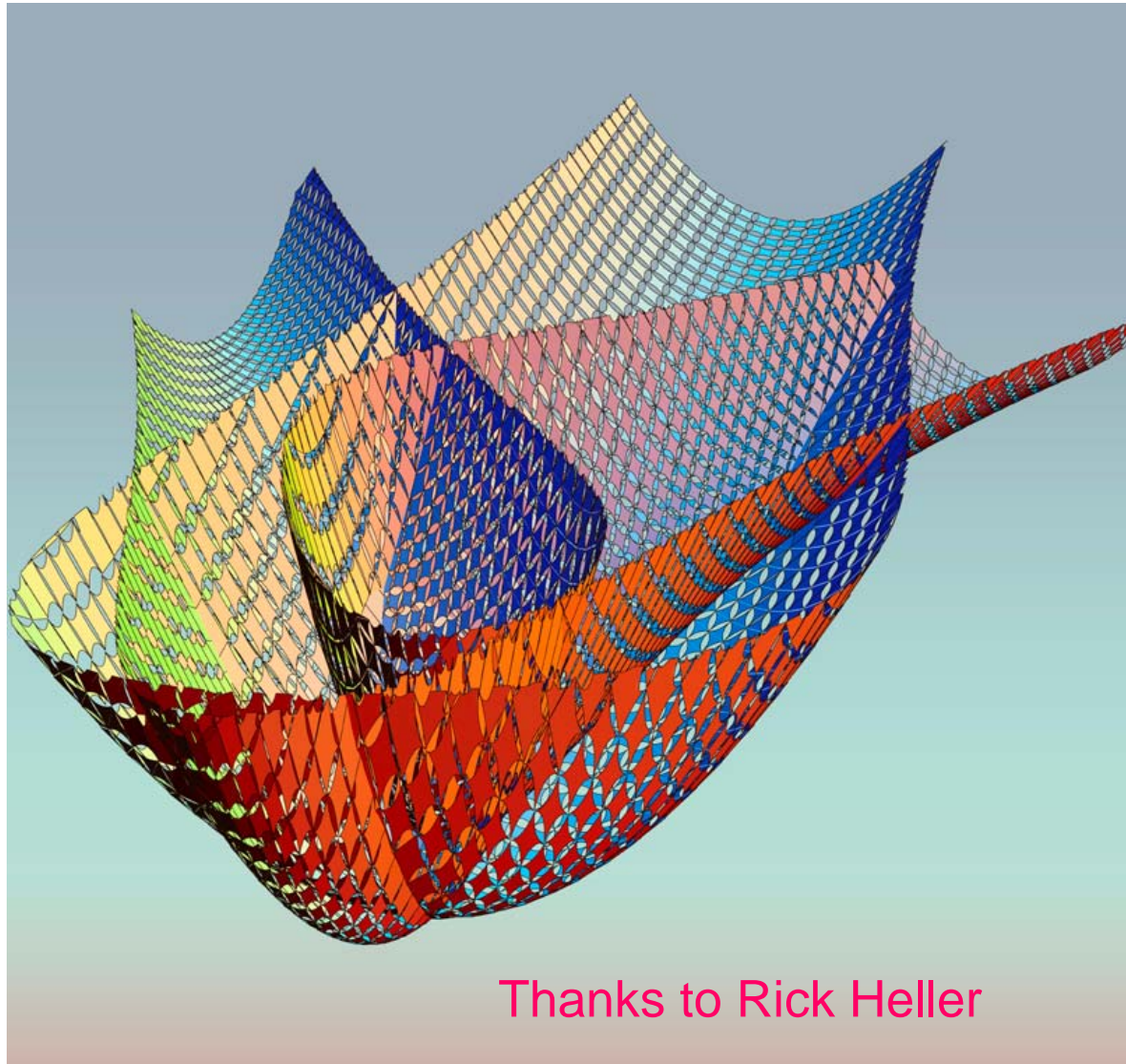
N degrees of freedom:
 $N-2$ dimensional “seam”

$$E_{\pm}(R_1, R_2) = \frac{H_{11}(R_1, R_2) + H_{22}(R_1, R_2)}{2} \pm \frac{1}{2} \sqrt{[H_{11}(R_1, R_2) - H_{22}(R_1, R_2)]^2 + 4[H_{12}(R_1, R_2)]^2}$$

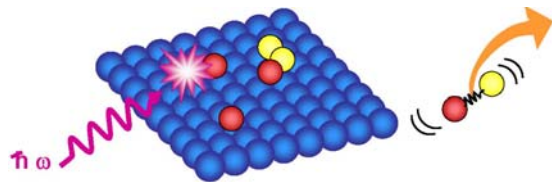


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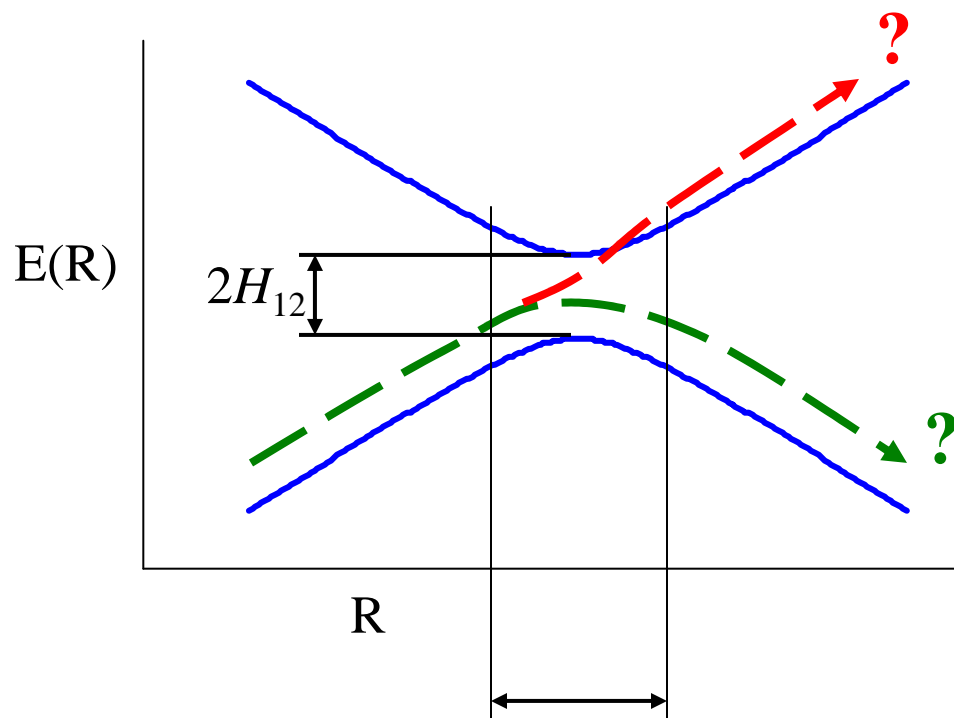


Thanks to Rick Heller



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$$\approx 2H_{12} / |\partial(H_{11} - H_{22}) / \partial R|$$

The Massey Criterion:

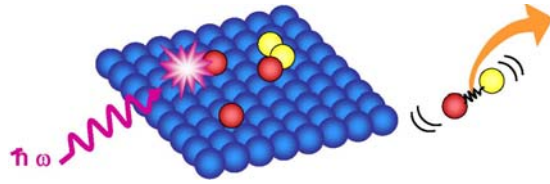
$$\Delta E \Delta t \gg \hbar$$

$$\Delta E \approx 2H_{12}$$

$$\Delta t \approx \text{distance/velocity}$$

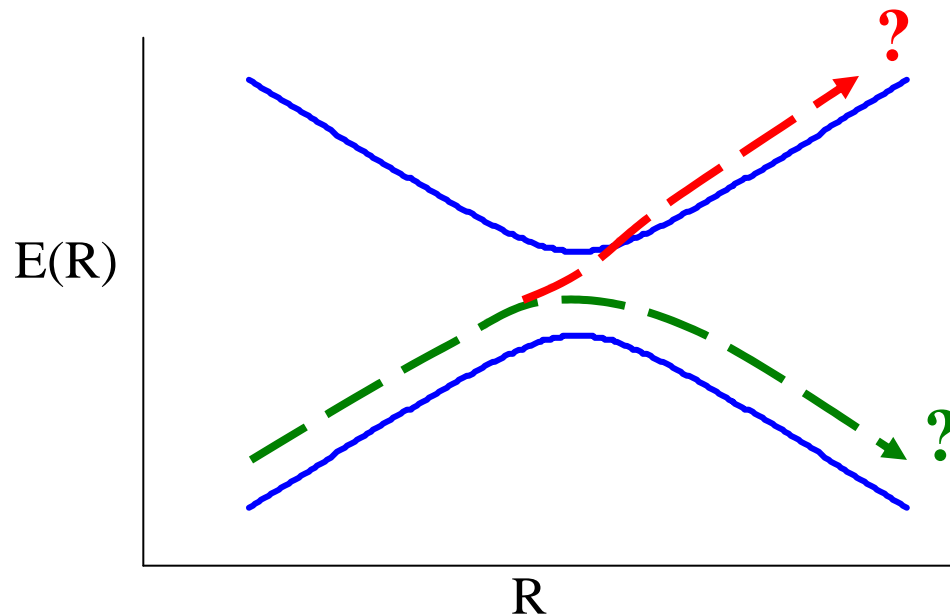
$$\approx 2H_{12} / |\partial(H_{11} - H_{22}) / \partial R| / \dot{R}$$

$$\longrightarrow \frac{\hbar \dot{R} |\partial(H_{11} - H_{22}) / \partial R|}{4H_{12}^2} \ll 1 \longrightarrow \text{adiabatic}$$



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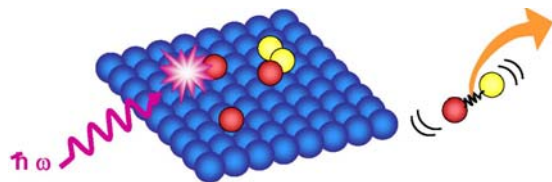


Landau-Zener Approximation

Assumptions:

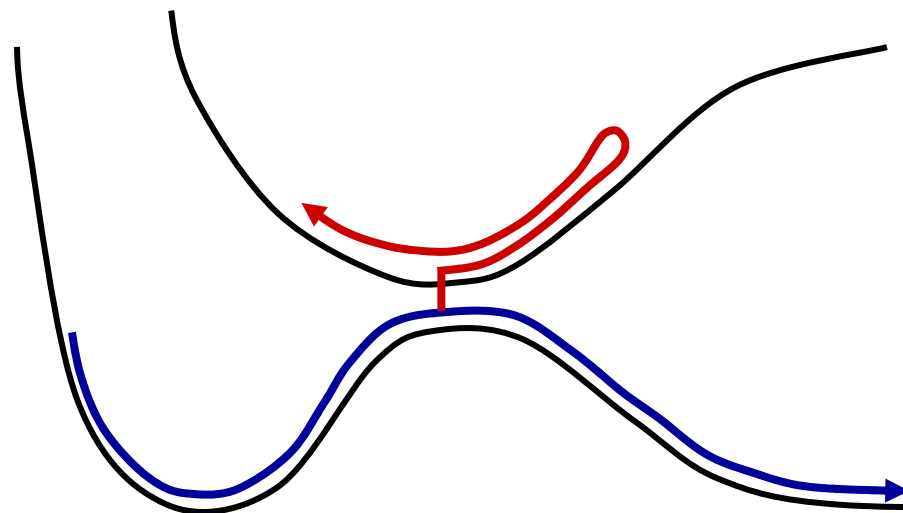
1. H_{11} and H_{22} linear
2. H_{12} constant
3. Velocity constant

$$P_{nonad} \approx \exp \left[\frac{-2\pi H_{12}^2}{\hbar \dot{R} \left| \partial(H_{11} - H_{22}) / \partial R \right|} \right]$$

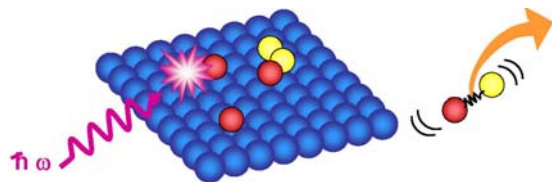


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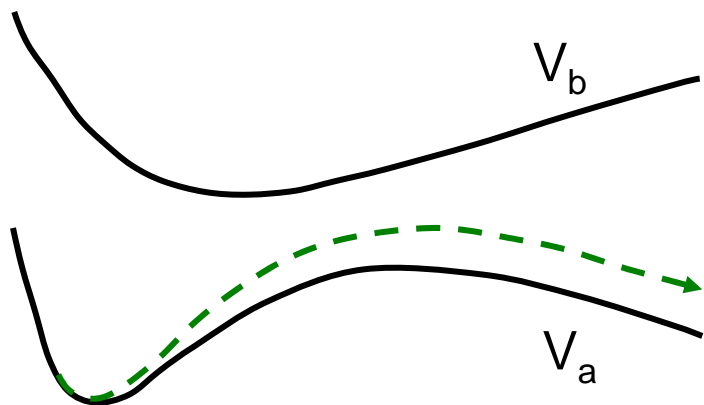
- Classical motion induces electronic transitions
 - Quantum state determines classical forces
- Quantum – Classical Feedback: Self-Consistency



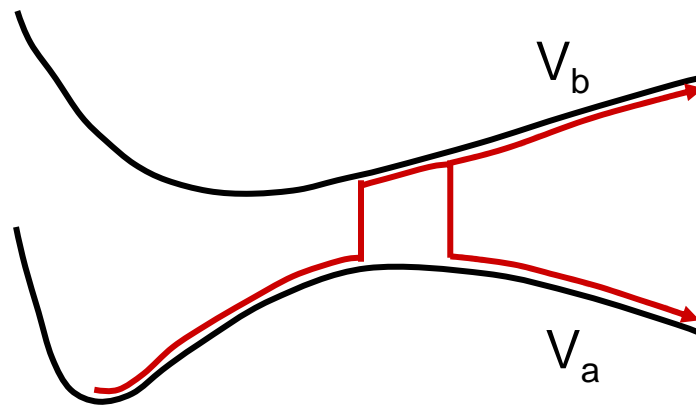
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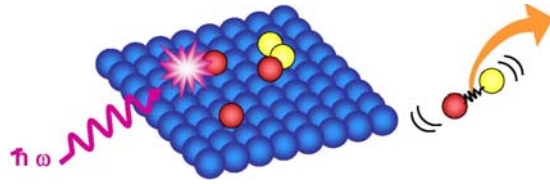
TWO GENERAL MIXED QUANTUM-CLASSICAL APPROACHES FOR INCLUDING FEEDBACK



Ehrenfest (SCF)



Surface-Hopping



V. Beyond Born-Oppenheimer

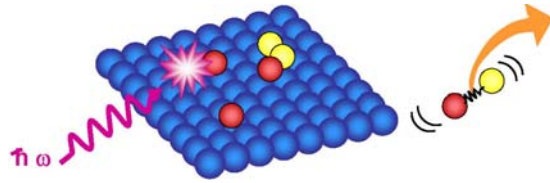
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$$i\hbar \frac{\partial \Psi(r,t)}{\partial t} = \mathcal{H}_{el} \Psi(r,t)$$

A diagram showing a blue solid curve representing a potential energy surface. A red dashed curve represents a wave packet moving along the potential. A red arrow points from the equation $\mathcal{H}_{el} \Psi(r,t)$ to the wave packet. A blue arrow labeled $R(t)$ points to the right, indicating the reaction coordinate.

$$\Psi(r,t) = \sum_i c_i(t) \Phi_i(r;R) \quad (\text{adiabatic states})$$

$$dc_j/dt = -\frac{i}{\hbar} V_{jj} c_j - \dot{R} \cdot \sum_i \langle \Phi_j(r;R) | \nabla_R \Phi_i(r;R) \rangle c_i$$



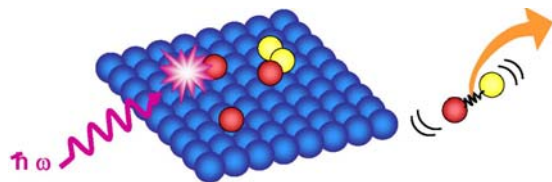
V. Beyond Born-Oppenheimer

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$$i\hbar \frac{\partial \Psi(r,t)}{\partial t} = \mathcal{H}_{el} \Psi(r,t)$$
A diagram showing a solid blue curve representing a classical path $R(t)$ with an arrow at its end. A dashed red curve represents a quantum wavefunction, oscillating around the blue curve. A red arrow points from the text $\mathcal{H}_{el} \Psi(r,t)$ to the dashed red curve.

Classical path must respond **self-consistently** to quantum transitions: “quantum back-reaction”

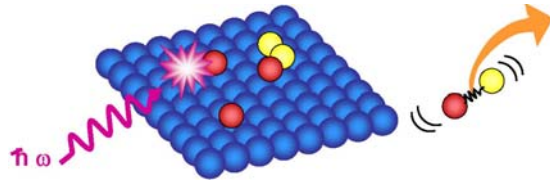
Ehrenfest and Surface Hopping differ only in how classical path is defined



Molecular Dynamics

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- I. The Potential Energy Surface
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- III. Adiabatic “on-the-fly” Dynamics
- IV. Car-Parrinello Dynamics
- V. Beyond Born Oppenheimer
- VI. Ehrenfest Dynamics**
- VII. Surface Hopping
- VIII. Equilibrium in Mixed Quantum-Classical Dynamics
- IX. Mixed Quantum-Classical Nuclear Motion



VI. Ehrenfest Dynamics

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$$i \hbar \frac{\partial}{\partial t} \Psi(\mathbf{r}, \mathbf{R}, t) = \mathcal{H}(\mathbf{r}, \mathbf{R}) \Psi(\mathbf{r}, \mathbf{R}, t) \quad (1)$$

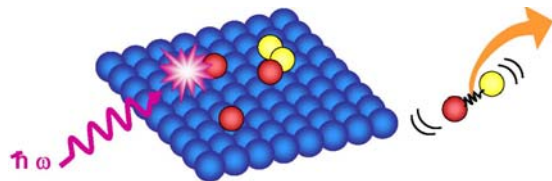
Self-consistent Field Approximation (fully quantum):

$$\Psi(\mathbf{r}, \mathbf{R}, t) = \Xi(\mathbf{r}, t) \Omega(\mathbf{R}, t) \exp \left[\frac{i}{\hbar} \int^t E_r(t') dt' \right] \quad (2)$$

Substituting (2) into (1), multiplying on the left by $\Omega(\mathbf{R}, t)$ and integrating over \mathbf{R} gives the SCF equation for the electronic wave function $\Xi(\mathbf{r}, t)$:

$$i \hbar \frac{\partial \Xi(\mathbf{r}, t)}{\partial t} = - \frac{\hbar^2}{2m_e} \sum_i \nabla_i^2 \Xi(\mathbf{r}, t) + \tilde{V}_{rR}(\mathbf{r}, \mathbf{R}) \Xi(\mathbf{r}, t) \quad (3)$$

$$\text{where } \tilde{V}_{rR}(\mathbf{r}, \mathbf{R}) = \int \Omega^*(\mathbf{R}, t) V_{rR}(\mathbf{r}, \mathbf{R}) \Omega(\mathbf{R}, t) d\mathbf{R} \quad (4)$$



VI. Ehrenfest Dynamics

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Substituting (2) into (1), multiplying on the left by $\Xi(\mathbf{r}, t)$ and integrating over \mathbf{r} gives the equivalent SCF equation for the nuclear wave function $\Omega(\mathbf{R}, t)$:

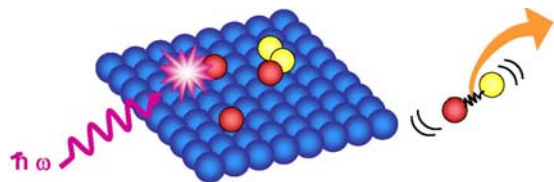
$$i\hbar \frac{\partial \Omega(\mathbf{R}, t)}{\partial t} = -\frac{\hbar^2}{2} \sum_{\alpha} M_{\alpha}^{-1} \nabla_{R_{\alpha}}^2 \Omega(\mathbf{R}, t) + \int \Xi^*(\mathbf{r}, t) \mathcal{H}_{el}(\mathbf{r}, \mathbf{R}) \Xi(\mathbf{r}, t) d\mathbf{r} \Omega(\mathbf{R}, t) \quad (5)$$

The classical (Ehrenfest) limit requires 2 steps:

1. Replace $\Omega(\mathbf{R}, t)$ with a delta function in Eq. (4)
2. Take the classical limit of Eq. (5) (eg. using the Bohm formulation as above).

Thus, the potential energy function governing the nuclei becomes

$$\int \Xi^*(\mathbf{r}, t) \mathcal{H}_{el}(\mathbf{r}, \mathbf{R}) \Xi(\mathbf{r}, t) d\mathbf{r} \quad \text{instead of the adiabatic energy } \mathcal{E}_j(\mathbf{R}).$$



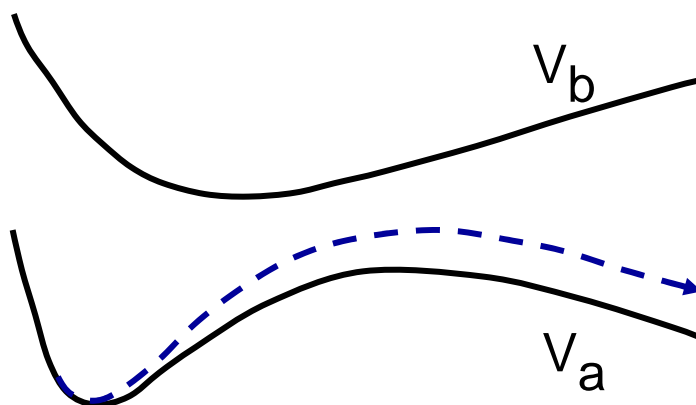
VI. Ehrenfest Dynamics

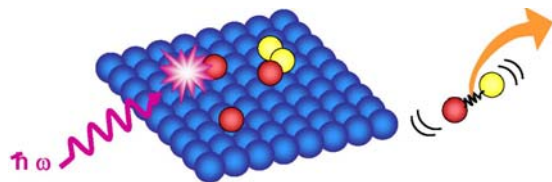
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$$i\hbar \frac{\partial \Psi(r,t)}{\partial t} = \mathcal{H}_{el} \Psi(r,t)$$

A diagram showing a blue solid curve representing a potential energy well. A red dashed curve represents a wave packet moving from left to right, as indicated by a red arrow. The horizontal axis is labeled $R(t)$.

$$M \ddot{R}(t) = -\nabla_R \langle \Psi(t) | \mathcal{H}_{el} | \Psi(t) \rangle$$



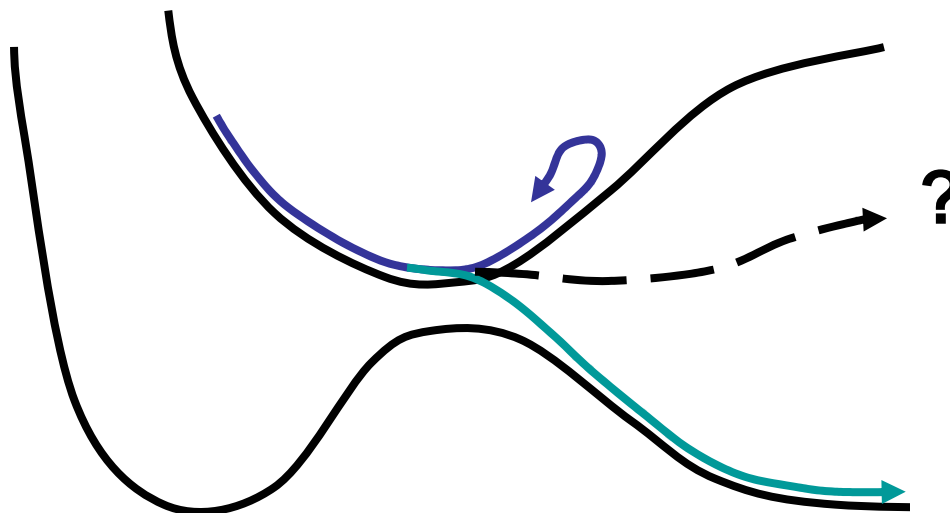


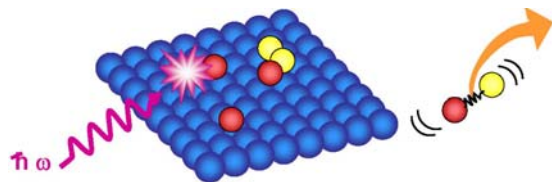
VI. Ehrenfest Dynamics

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$$M \ddot{R}(t) = -\nabla_R \langle \Psi(t) | \mathcal{H}_{el} | \Psi(t) \rangle$$

Problem:
single configuration
→ average path

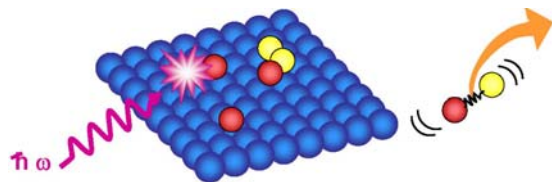




Molecular Dynamics

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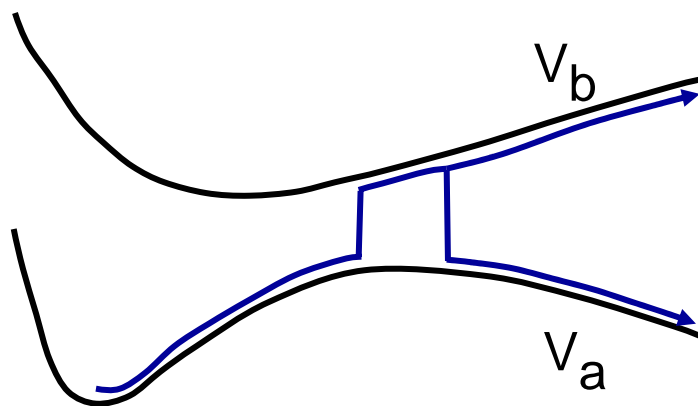
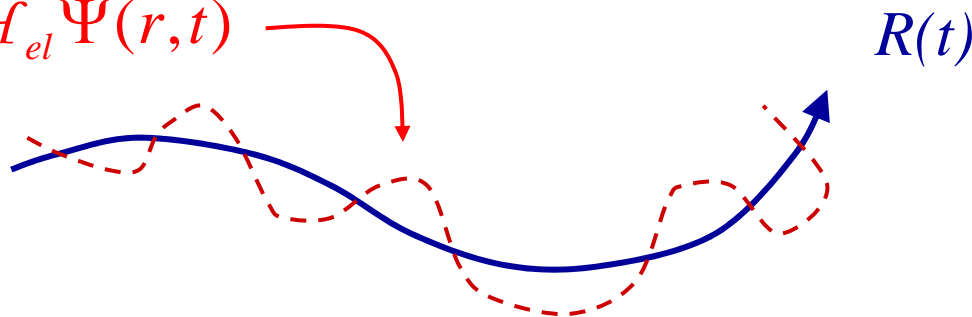
- I. The Potential Energy Surface
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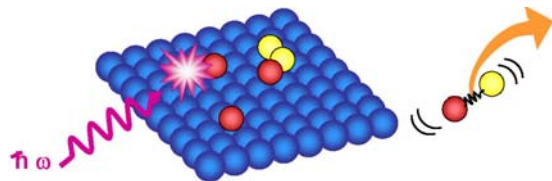


VII. Surface Hopping

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$$i\hbar \frac{\partial \Psi(r,t)}{\partial t} = \mathcal{H}_{el} \Psi(r,t)$$





VII. Surface Hopping

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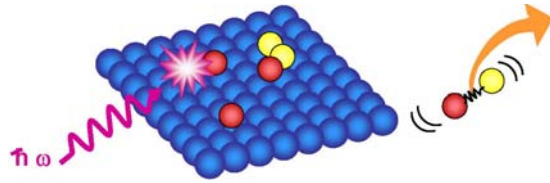
Multi-Configuration Wave Function:

$$\Psi(\mathbf{r}, \mathbf{R}, t) = \sum_j \Phi_j(\mathbf{r}, \mathbf{R}) \Omega_j(\mathbf{R}, t)$$

Substitute into Schrodinger Eq and take classical limit:

→ Surface Hopping

However, a rigorous classical limit has not been achieved !



VII. Surface Hopping

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One Approach: Multi-Configuration Bohm Equations:

$$\Psi(\mathbf{r}, \mathbf{R}, t) = \sum_j \Phi_j(\mathbf{r}, \mathbf{R}) \Omega_j(\mathbf{R}, t)$$

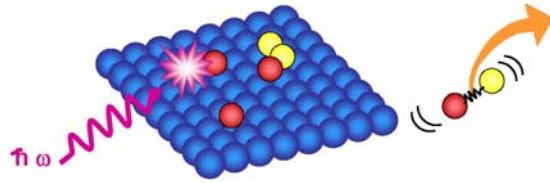
$$\Omega_j(\mathbf{R}, t) = A_j(\mathbf{R}, t) \exp\left[\frac{i}{\hbar} S_j(\mathbf{R}, t)\right]$$

$$\dot{S}_j = -\frac{1}{2M} (\nabla_R S_j)^2 - E_j(\mathbf{R}) - \frac{\hbar^2}{2M} \frac{\nabla_R^2 A_j}{A_j}$$

small $\hbar \rightarrow$

$$\dot{S}_j = -\frac{1}{2M} (\nabla_R S_j)^2 - E_j(\mathbf{R})$$

\rightarrow motion on potential energy surface j



VII. Surface Hopping

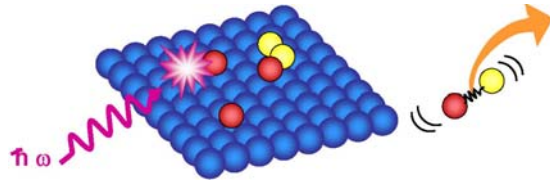
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$$\dot{A}_n = \nabla_R A_n \cdot \dot{\mathbf{R}} - \frac{1}{2M} A_n \nabla_R^2 S_n - \sum_m A_m \langle \Phi_n \nabla_R \Phi_m \rangle \cdot \dot{\mathbf{R}} \exp\left[\frac{i}{\hbar}(S_m - S_n)\right]$$

Surface Hopping:

Evaluate all quantities along a single path

Sum over many stochastic paths



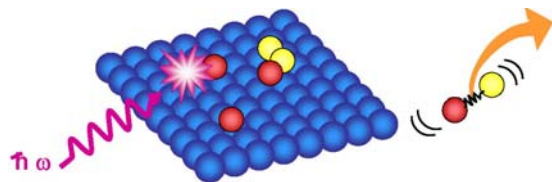
VII. Surface Hopping

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Multi-Configuration Theory: *Surface Hopping*

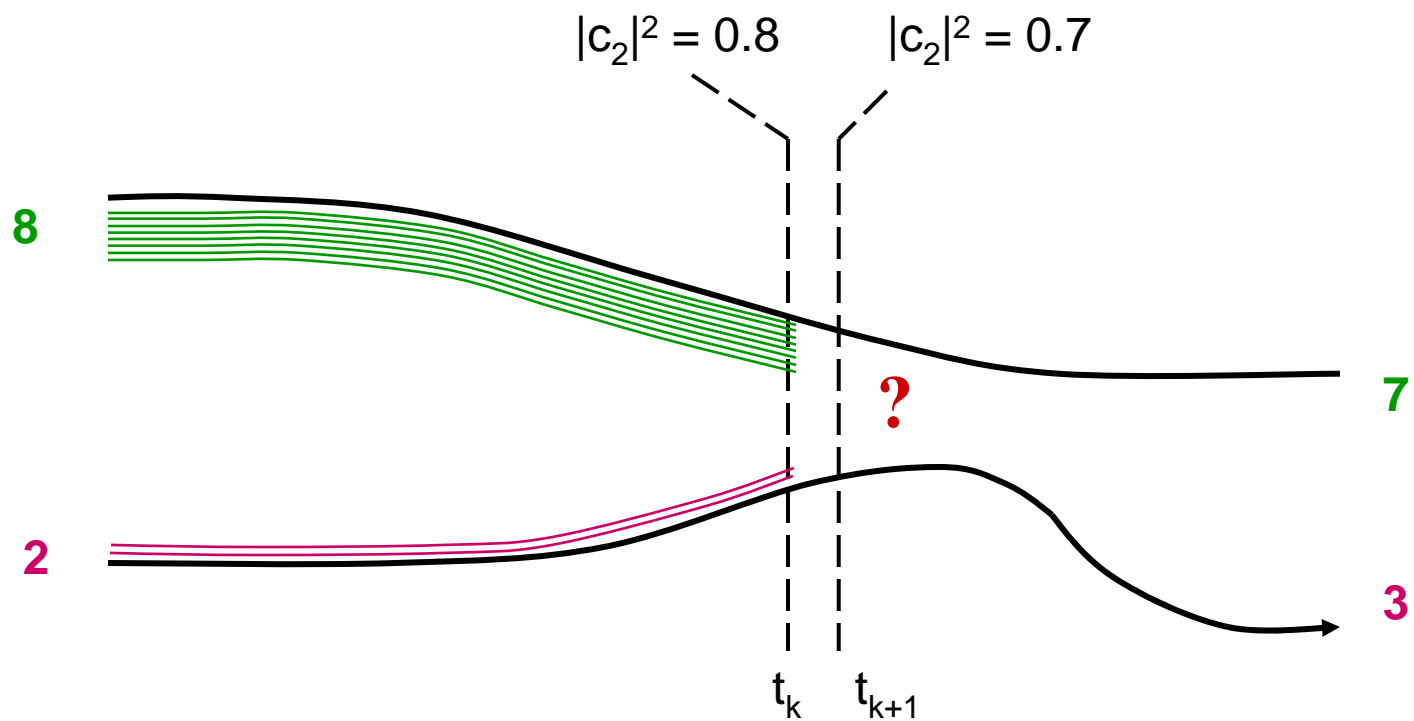
$$i\hbar \frac{\partial \Psi(t)}{\partial t} = \mathcal{H}_{el} \Psi(t)$$

- 1] $M \ddot{R}(t) = -\nabla_R \mathcal{E}_k$, i.e., motion on single p.e.s.
- 2] Stochastic “hops” between states so that probability = $|c_k|^2$
- 3] Apply instantaneous “Pechukas Force” to conserve energy
- 4] “Fewest Switches”: achieve [2] with fewest possible hops:

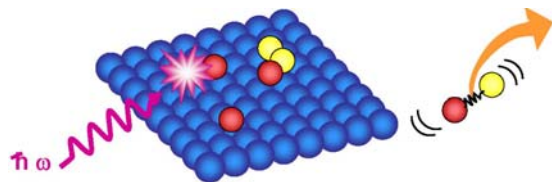


VII. Surface Hopping

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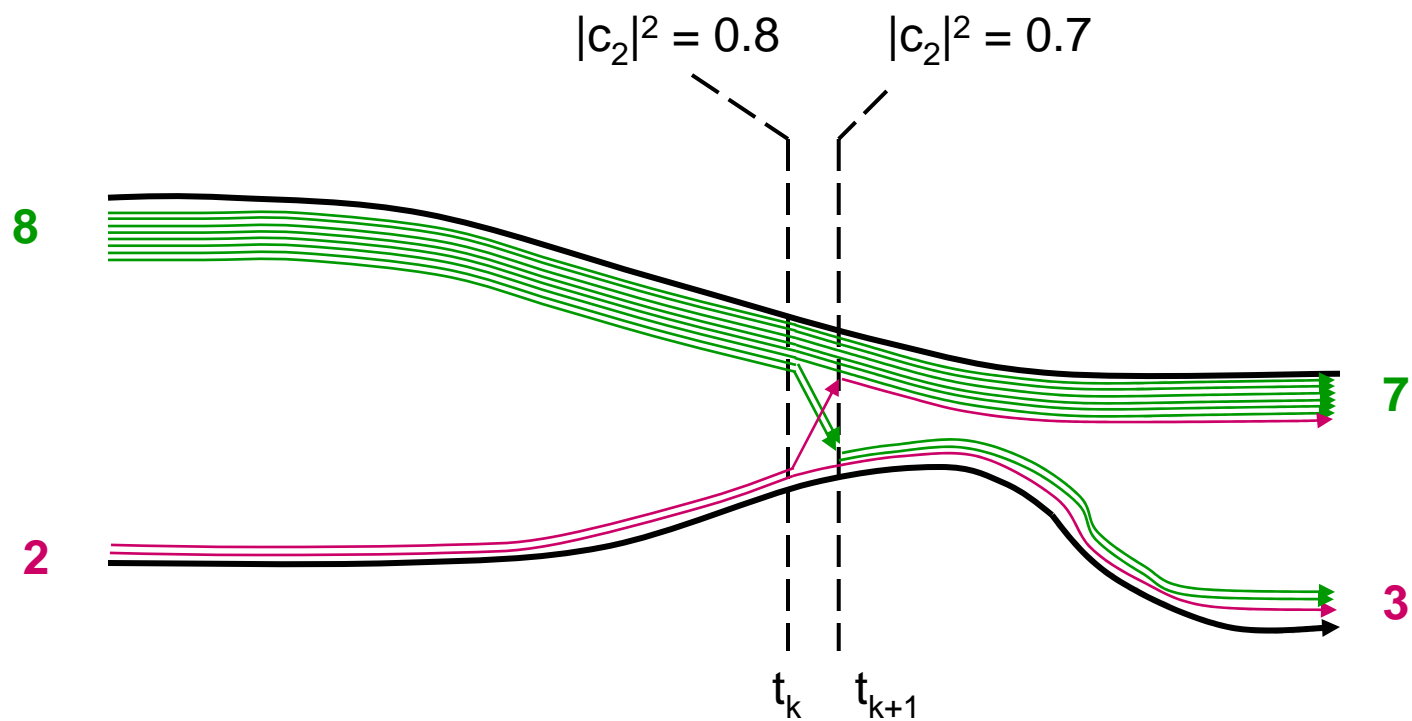


Stochastic *Fewest Switches* algorithm (2-state):

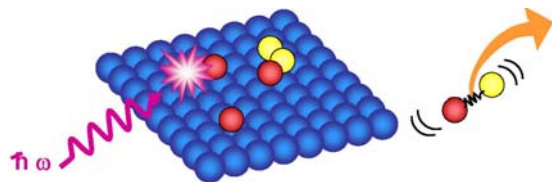


VII. Surface Hopping

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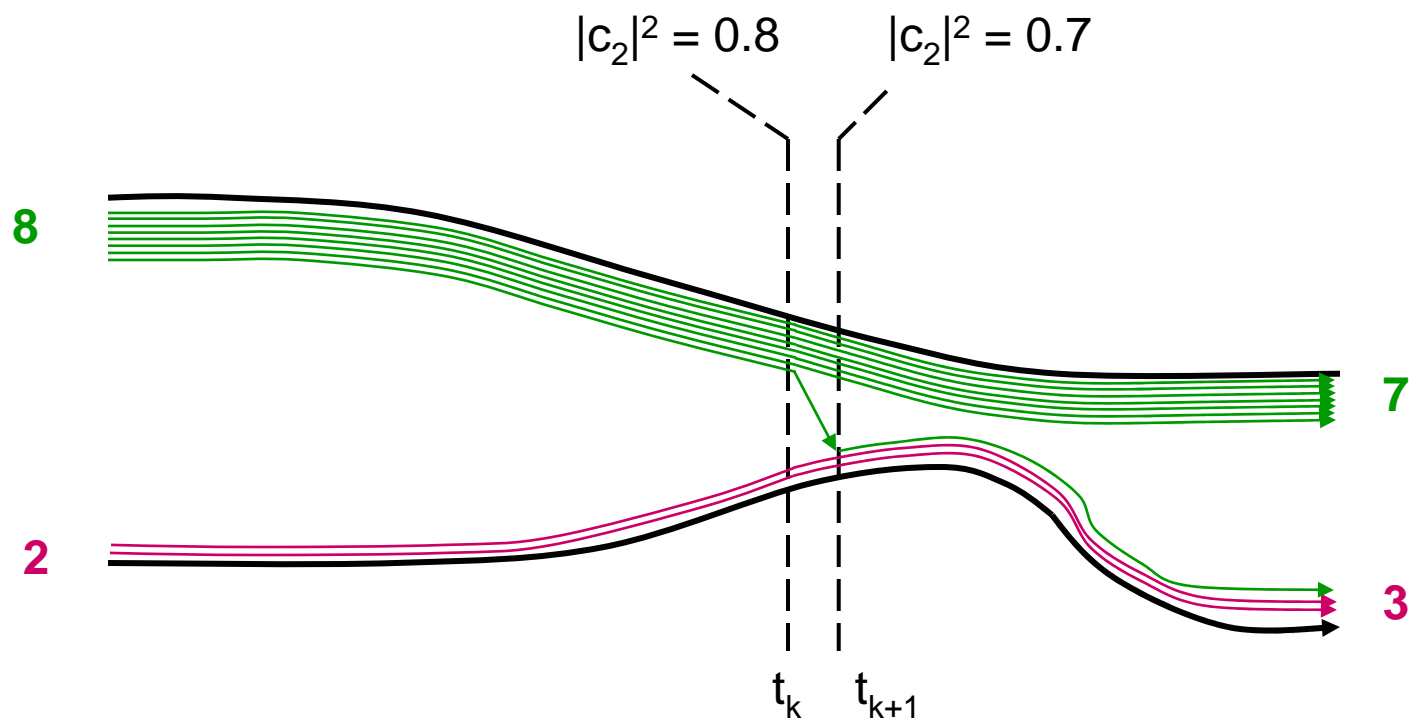


Stochastic *Fewest Switches* algorithm (2-state):



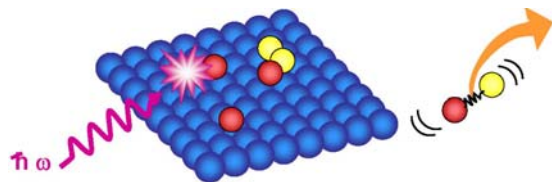
VII. Surface Hopping

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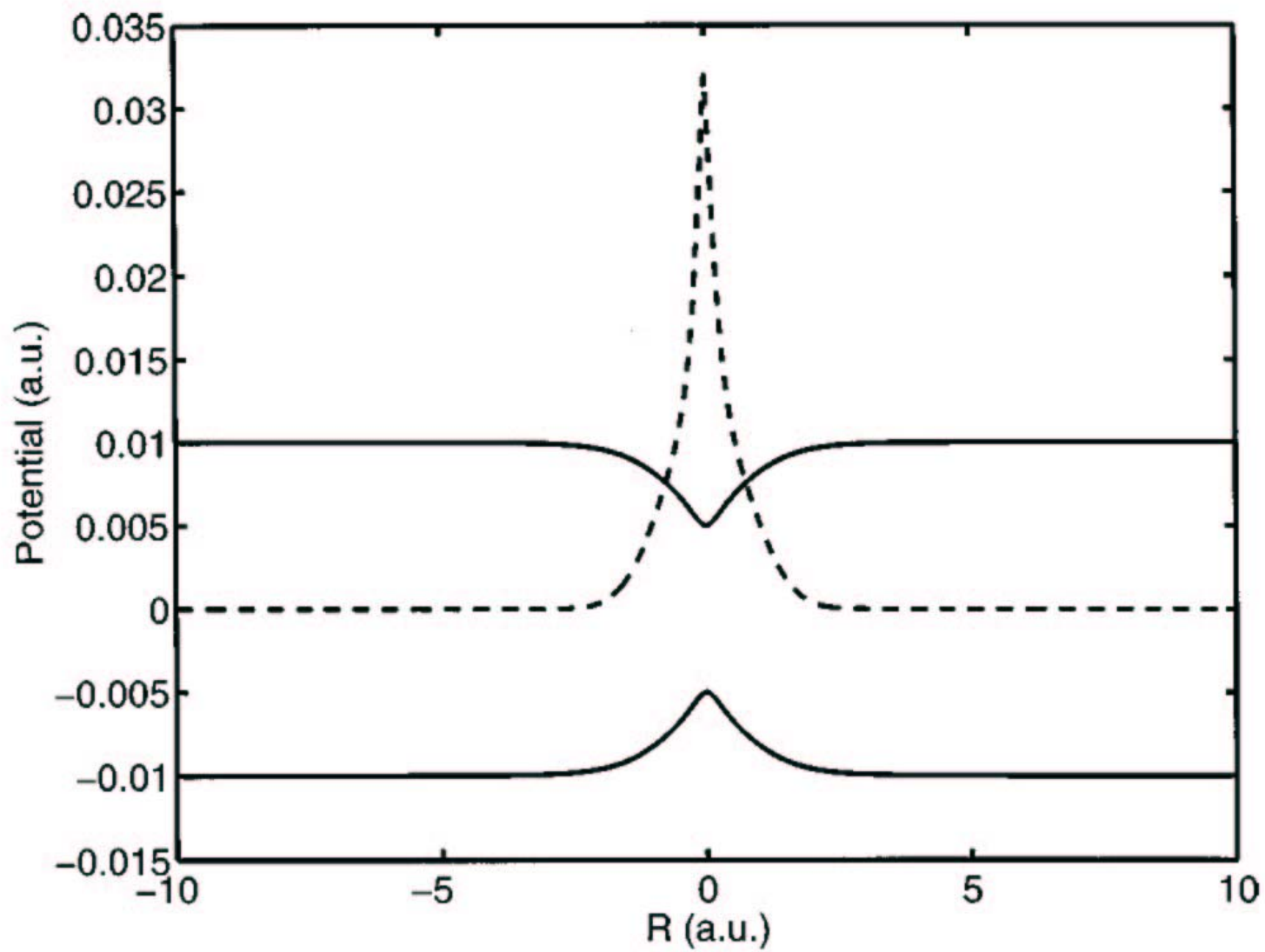
Stochastic *Fewest Switches* algorithm (2-state):

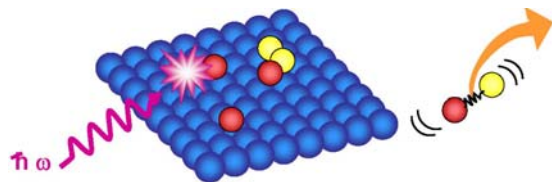
$$P_{2 \rightarrow 1} = \begin{cases} \frac{|c_2(k)|^2 - |c_2(k+1)|^2}{|c_2(k)|^2}, & |c_2(k)|^2 > |c_2(k+1)|^2 \\ 0, & |c_2(k)|^2 \leq |c_2(k+1)|^2 \end{cases}$$



VII. Surface Hopping

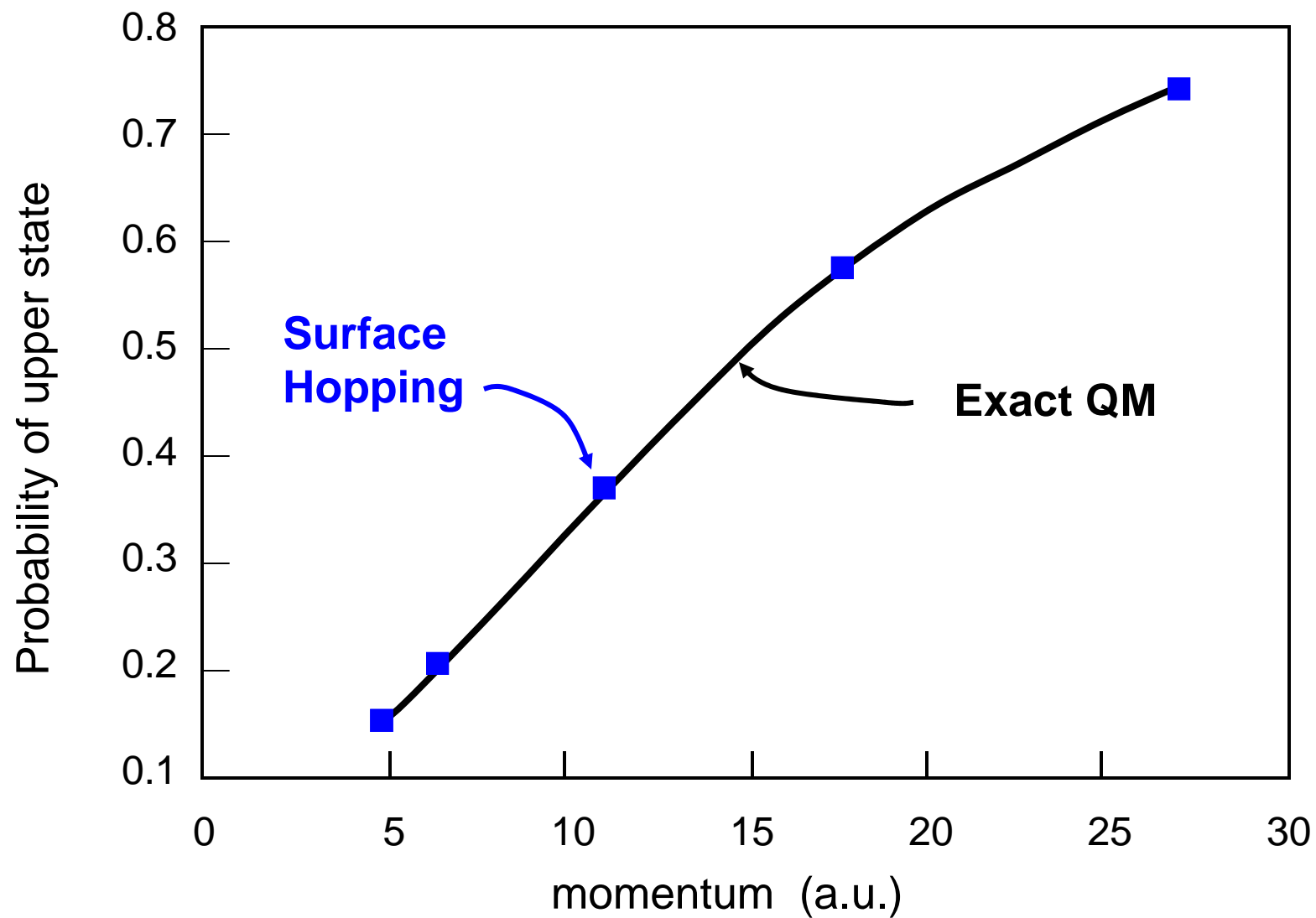
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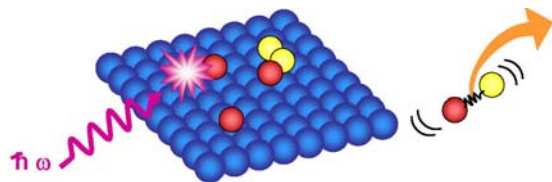




VII. Surface Hopping

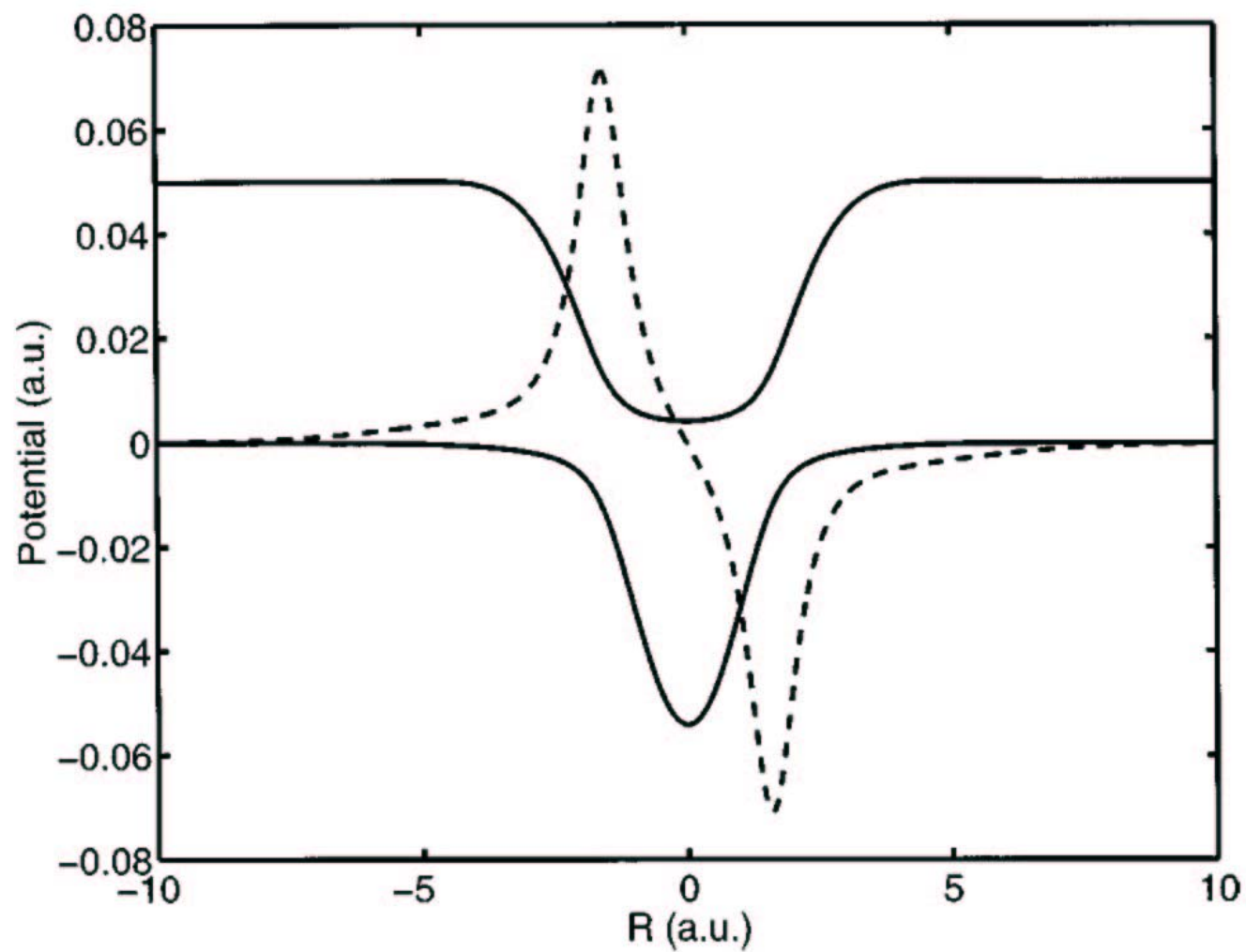
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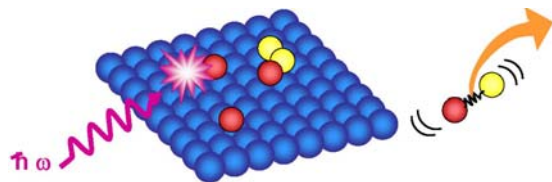




VII. Surface Hopping

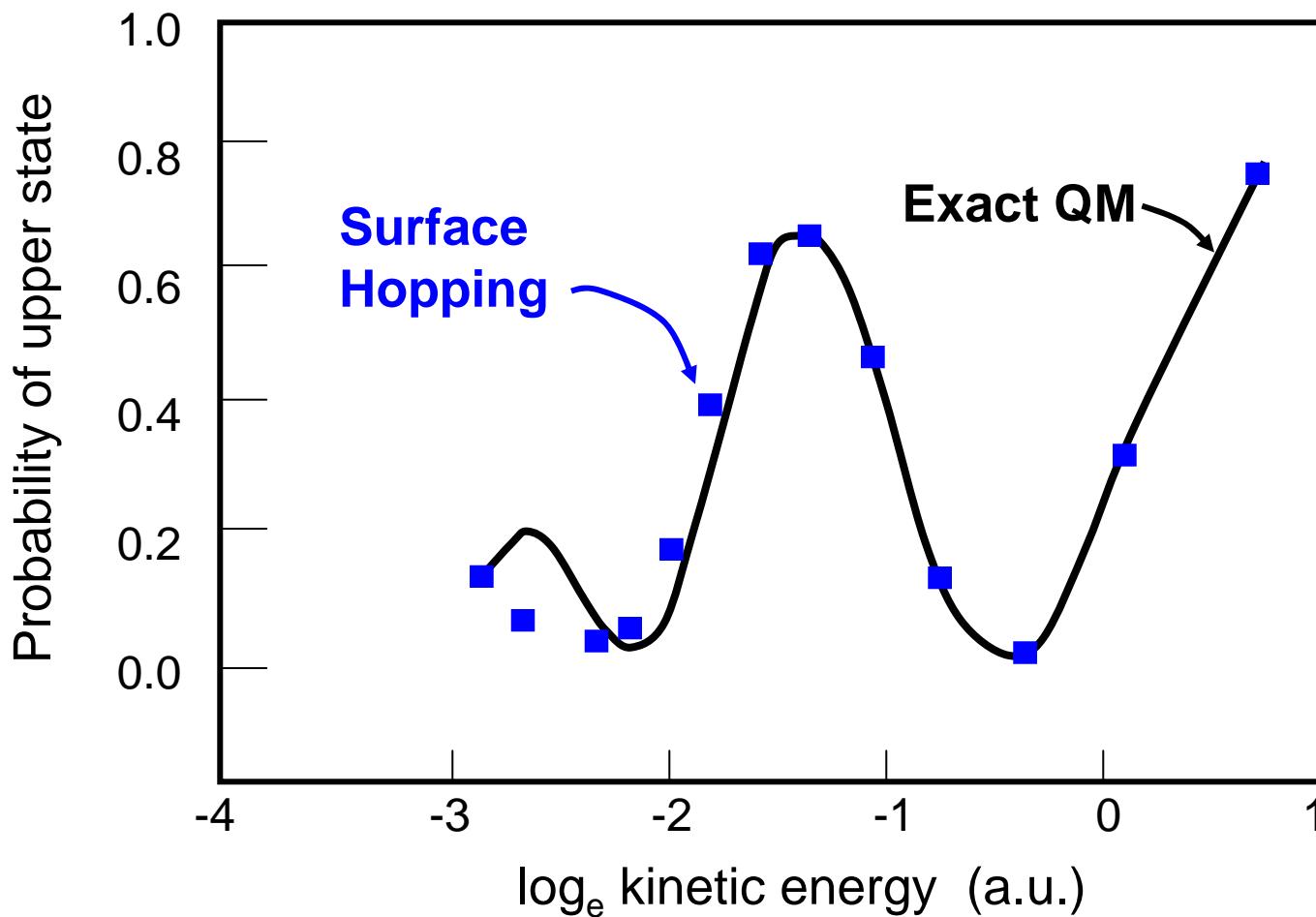
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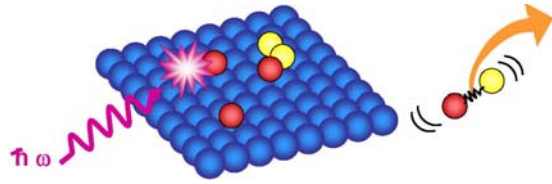




VII. Surface Hopping

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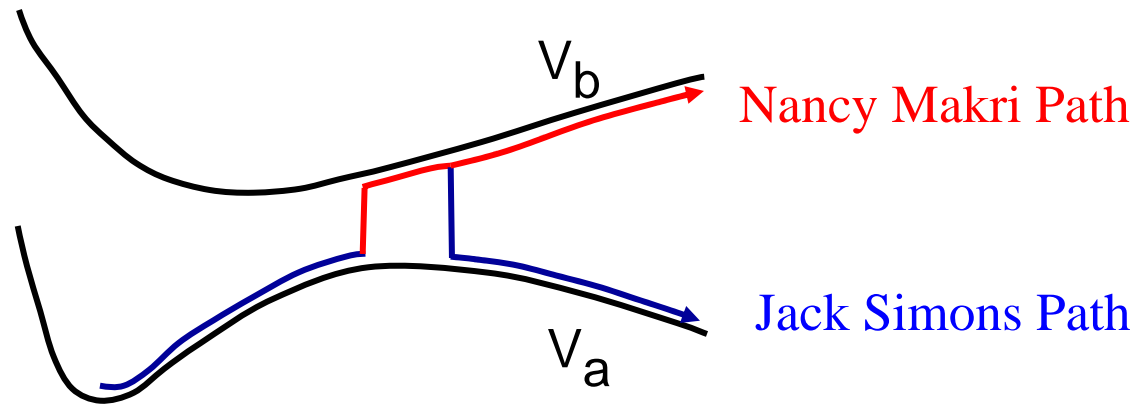
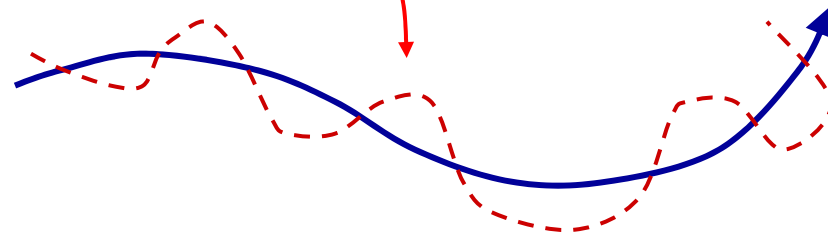


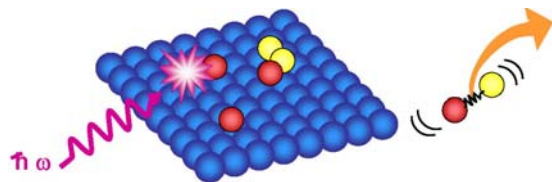


VII. Surface Hopping

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$$i\hbar \frac{\partial \Psi(r,t)}{\partial t} = \mathcal{H}_{el} \Psi(r,t) \quad R(t)$$





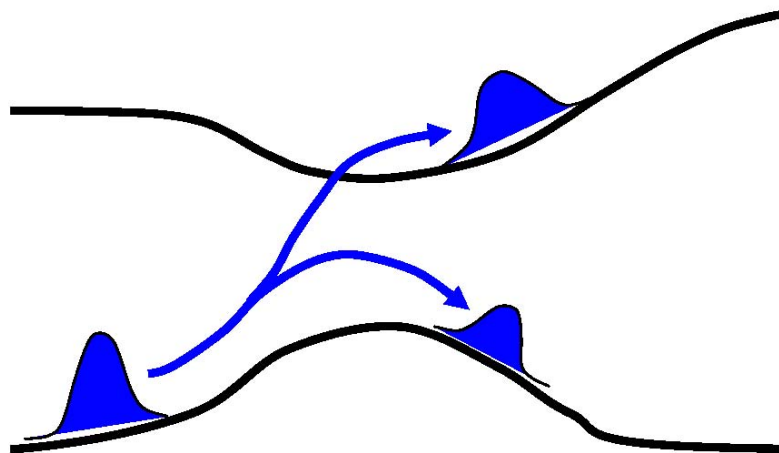
VII. Surface Hopping

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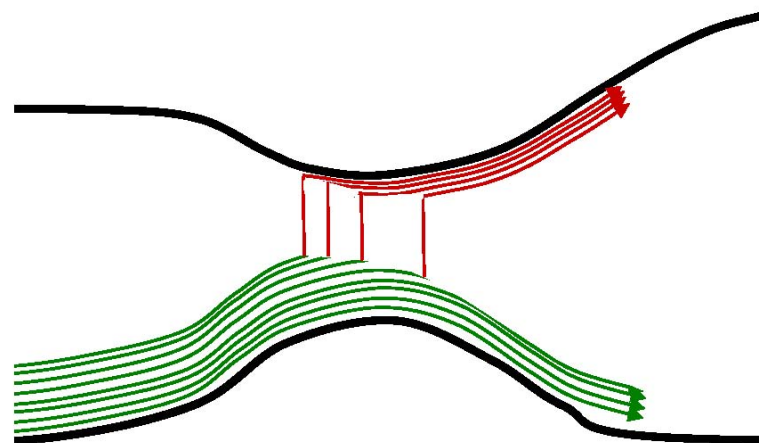
SHORTCOMINGS OF SURFACE HOPPING

1] Trajectories are independent

Trajectories should talk to each other

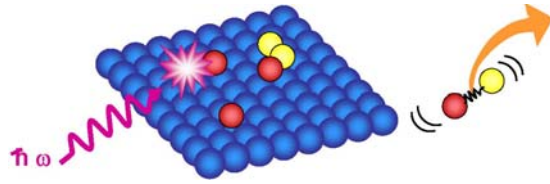


quantum wave packet



surface hopping

Fundamental approximation, but required to make practical



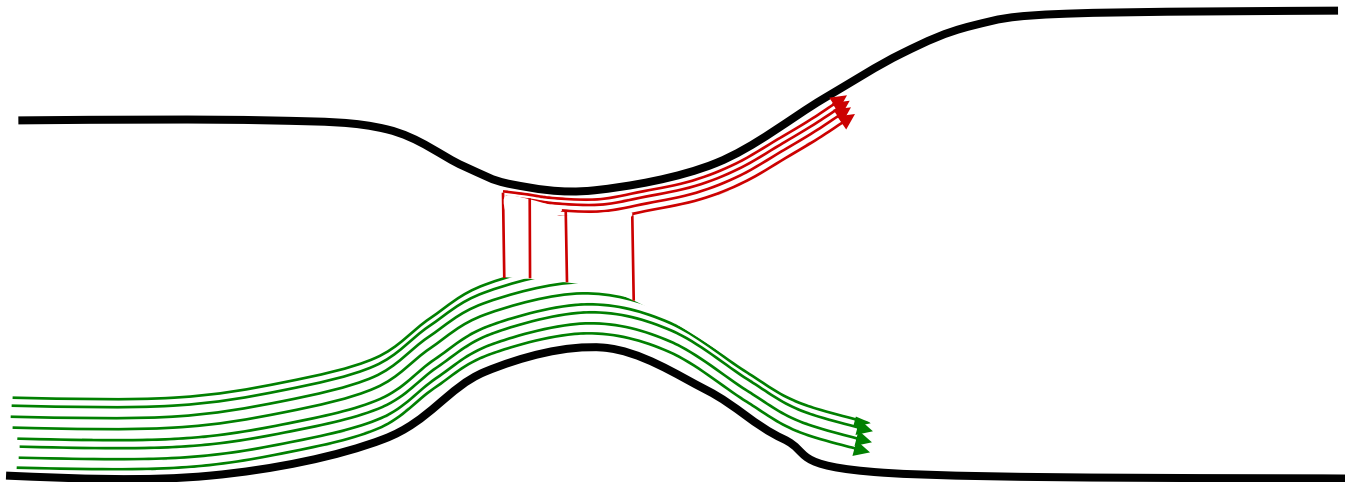
VII. Surface Hopping

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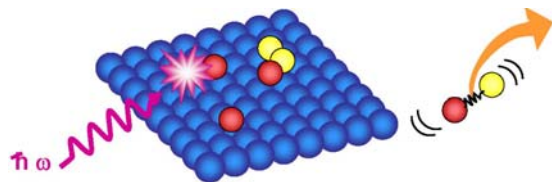
SHORTCOMINGS OF SURFACE HOPPING

2] Too drastic: hops require sudden change of velocity

Consider swarm of trajectories –
trajectories hop stochastically at different times



→ gradual evolution of flux



VII. Surface Hopping

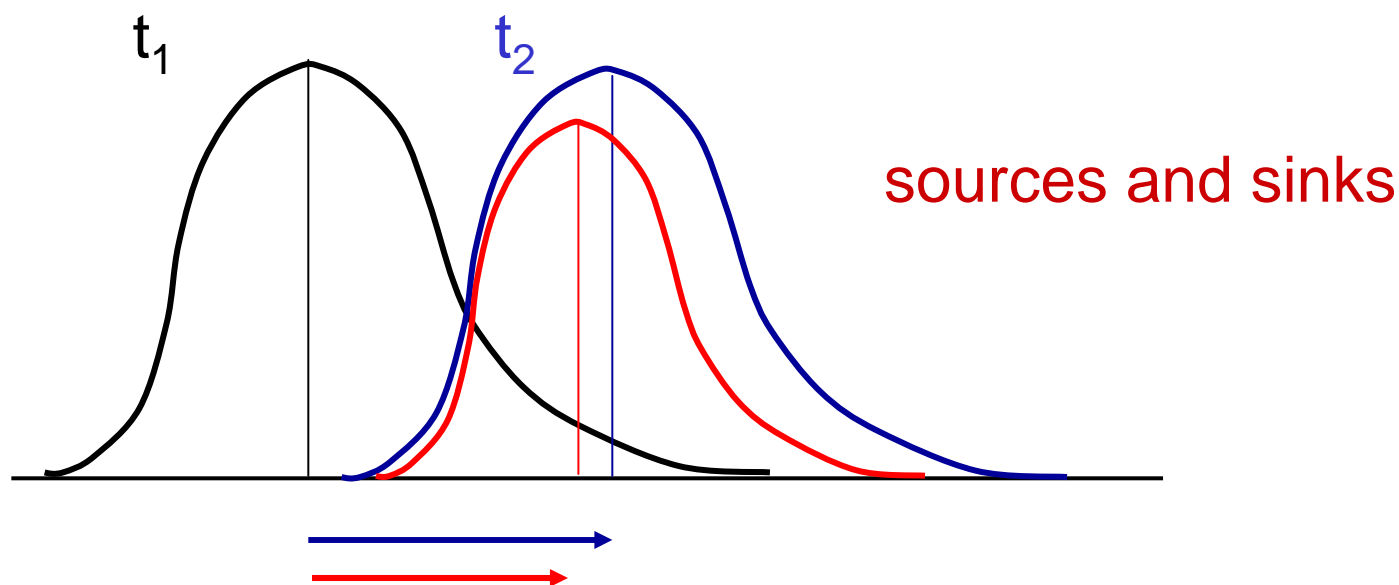
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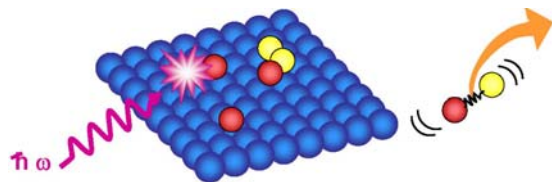
SHORTCOMINGS OF SURFACE HOPPING

- 3] Trajectories should evolve on some effective potential, not on a single adiabatic potential energy surface

Consider swarm of trajectories –

trajectories hop stochastically at different times:



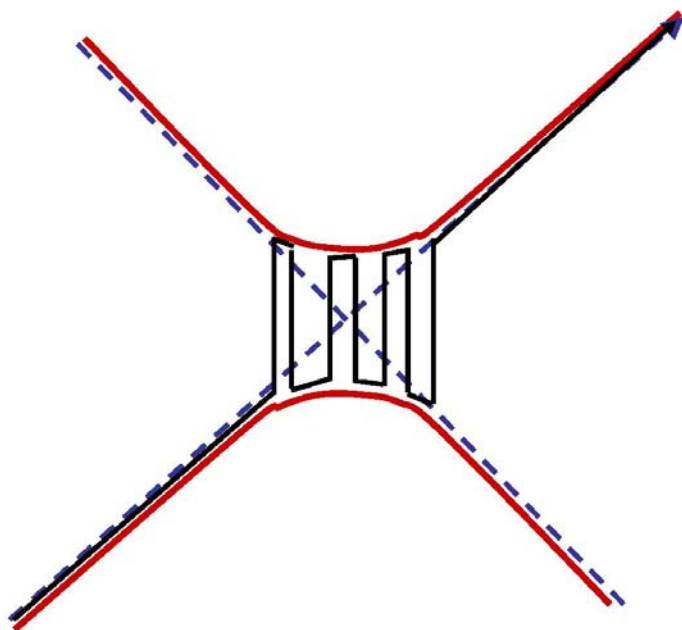


VII. Surface Hopping

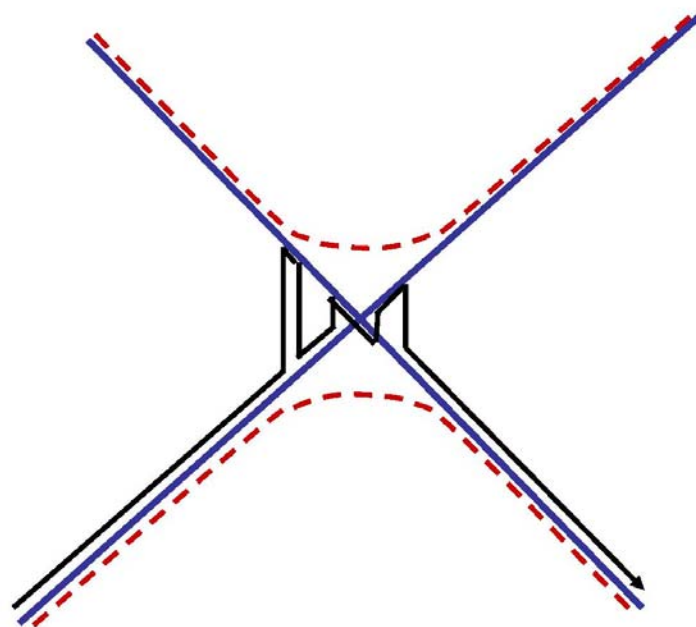
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SHORTCOMINGS OF SURFACE HOPPING

4] Not invariant to representation

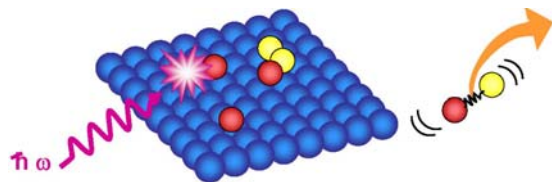


adiabatic representation



diabatic representation

The natural representation for surface hopping is *adiabatic*



VII. Surface Hopping

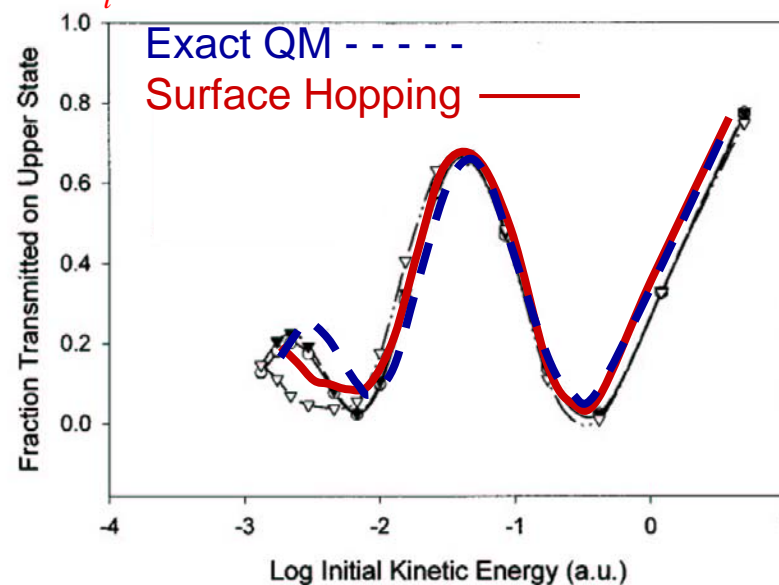
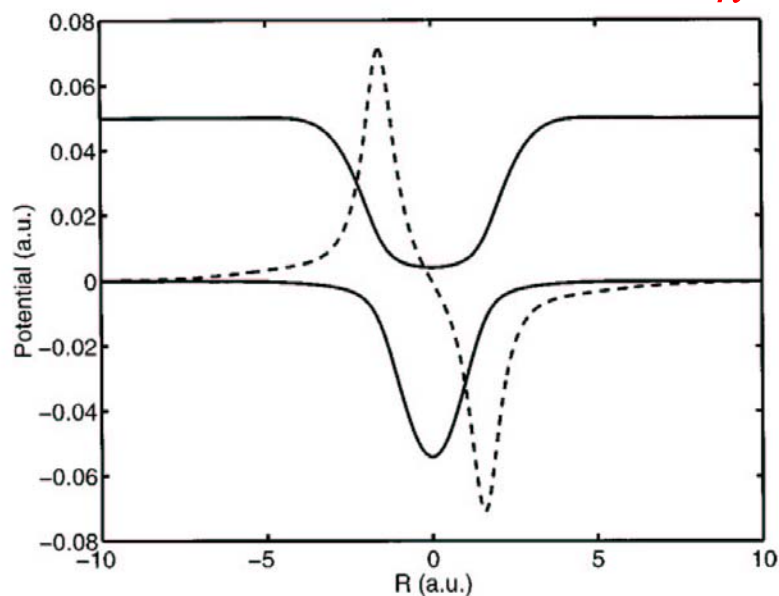
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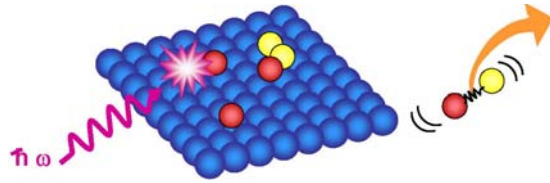
SHORTCOMINGS OF SURFACE HOPPING

5] Quantum Mechanical Coherence Neglected –
~~uses probabilities, not amplitudes~~ FALSE

$$\Psi(t) = \sum_i c_i(t) \varphi_i(R)$$

$$dc_j/dt = -\frac{i}{\hbar} V_{jj} c_j - \dot{R} \cdot \sum_i \langle \varphi_j | \nabla_R \varphi_i \rangle c_i$$





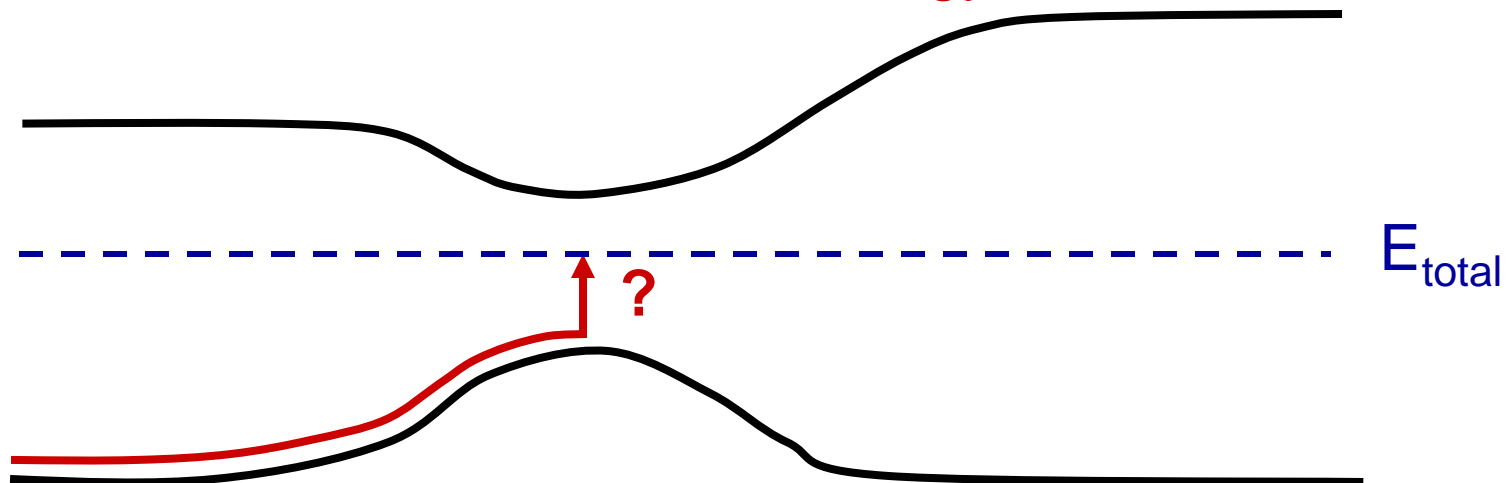
VII. Surface Hopping

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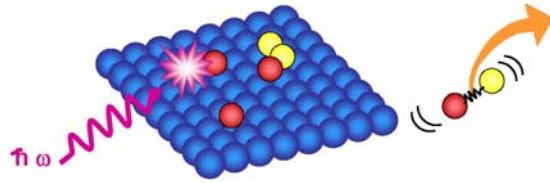
SHORTCOMINGS OF SURFACE HOPPING

6] Forbidden Hops (or *frustrated hops*)

Hopping algorithm calls for a hop but there is insufficient kinetic energy



→ probability on state $k \neq |c_k|^2$



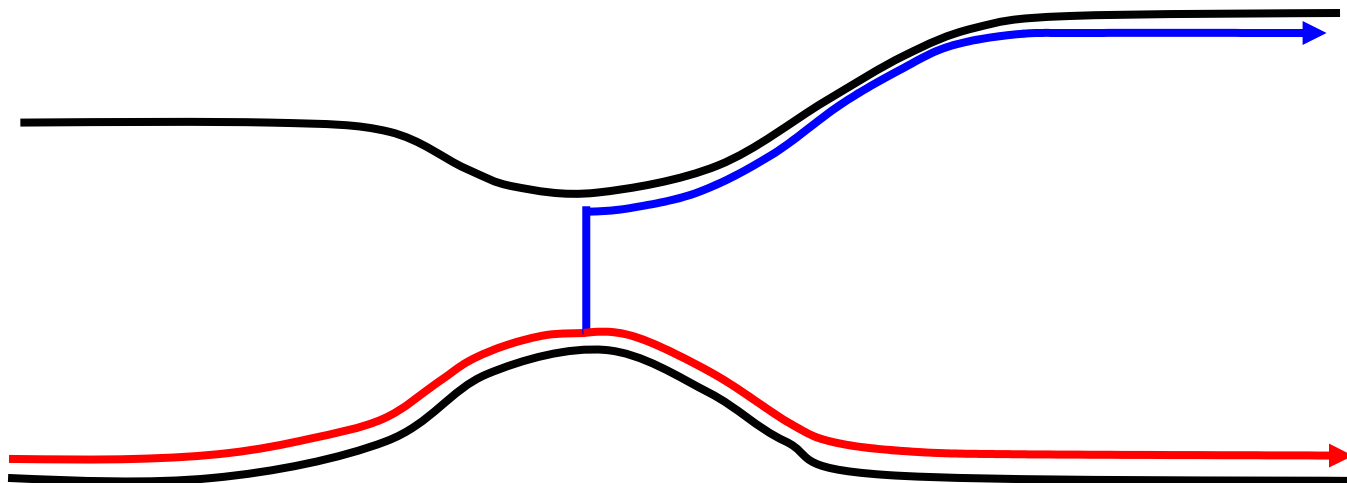
VII. Surface Hopping

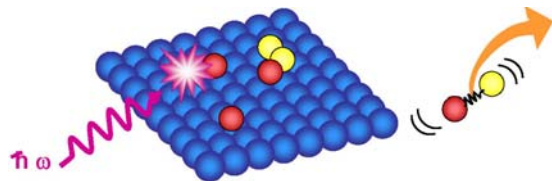
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SHORTCOMINGS OF SURFACE HOPPING

7] Detailed Balance?

What are the populations of the quantum states at equilibrium?

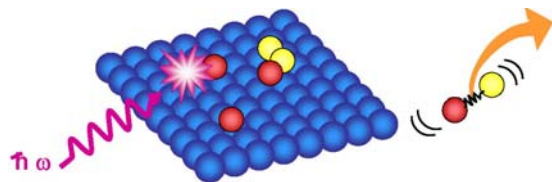




Molecular Dynamics

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- I. The Potential Energy Surface
- II. The Classical Limit via the Bohm Equations
- III. Adiabatic “on-the-fly” Dynamics
- IV. Car-Parrinello Dynamics
- V. Beyond Born Oppenheimer
- VI. Ehrenfest Dynamics
- VII. Surface Hopping
- VIII. Equilibrium in Mixed Quantum-Classical Dynamics**
- IX. Mixed Quantum-Classical Nuclear Motion

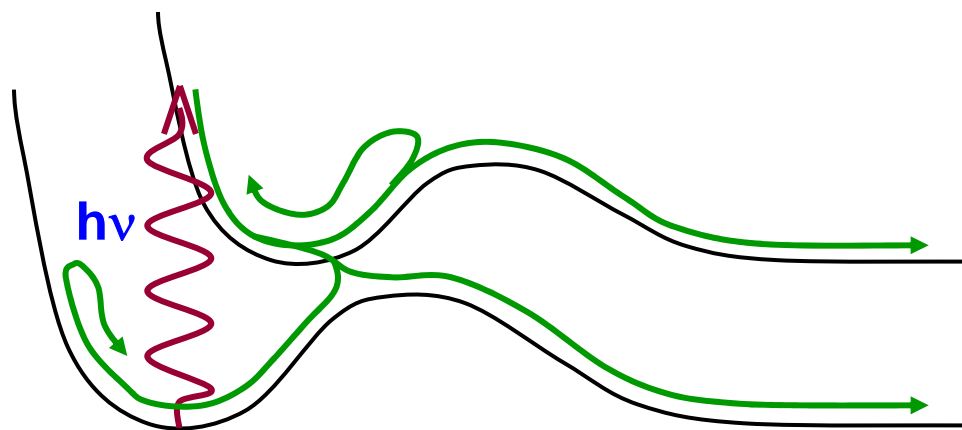


VIII. Equilibrium in Mixed Quantum-Classical Dynamics

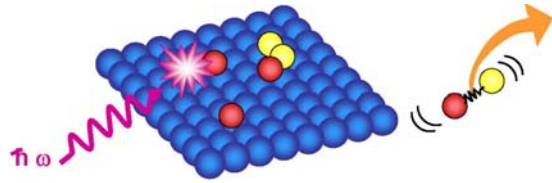
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Detailed Balance: $\mathcal{N}_1 P_{1 \rightarrow 2} = \mathcal{N}_2 P_{2 \rightarrow 1} \rightarrow \text{Equilibrium}$

- Long Timescales
- Multiple Transitions
- Relaxation Processes
- Infrequent events



e.g., nonradiative transition
vs. reaction on excited state



VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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Detailed Balance: $\mathcal{N}_1 P_{1 \rightarrow 2} = \mathcal{N}_2 P_{2 \rightarrow 1}$

$$i\hbar \frac{\partial \Psi(r, t)}{\partial t} = \mathcal{H}_{el} \Psi(r, t)$$

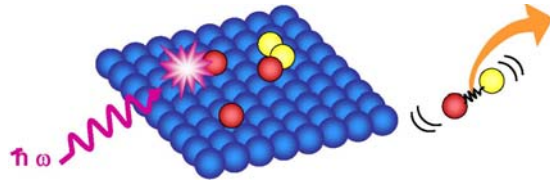
A diagram showing a solid blue curve representing a wave function $\Psi(r, t)$ and a dashed red curve representing a reaction coordinate $R(t)$. A red arrow points from the label $\mathcal{H}_{el} \Psi(r, t)$ in the equation above to the solid blue curve.

$$dc_j/dt = -\frac{i}{\hbar} V_{jj} c_j - \dot{R} \cdot \sum_i \langle \Phi_j(r; R) | \nabla_R \Phi_i(r; R) \rangle c_i$$

time reversible

$$P_{1 \rightarrow 2} = P_{2 \rightarrow 1} \quad \rightarrow \quad |c_1|^2 = |c_2|^2$$

probabilities of each quantum state are equal: **infinite temperature**



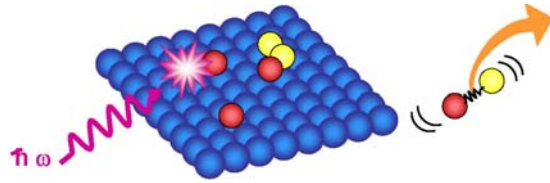
VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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“In theories in which the reservoir is treated classically and its effects on the system described in terms of random functions instead of noncommuting operators, it follows that $W_{mn} = W_{nm}$. This is a serious shortcoming of all semiclassical theories of relaxation.” K. Blum, *Density Matrix Theory and Applications*, 2nd Ed., (Plenum, NY, 1996).

However

This is **not** true for either Ehrenfest or Surface Hopping



VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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Ehrenfest (SCF)

$$\rho(E_{QM}) \propto \int_0^{\infty} H(E_{TOT} - E_{QM}) \exp(-\beta E_{TOT}) dE_{TOT}$$

$$\langle E_{QM} \rangle = \frac{\int_0^{\Delta} E_{QM} \rho(E_{QM}) dE_{QM}}{\int_0^{\Delta} \rho(E_{QM}) dE_{QM}}$$

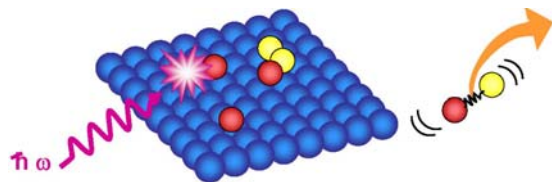
$$\rightarrow \langle E_{QM} \rangle = \frac{1}{\beta} - \frac{\Delta \exp(-\beta \Delta)}{1 - \exp(-\beta \Delta)}$$

$$\langle |c_2|^2 \rangle = \frac{1}{\beta \Delta} - \frac{\exp(-\beta \Delta)}{1 - \exp(-\beta \Delta)}$$

Boltzmann:

$$\langle E_{QM} \rangle = \frac{\Delta \exp(-\beta \Delta)}{1 + \exp(-\beta \Delta)}$$

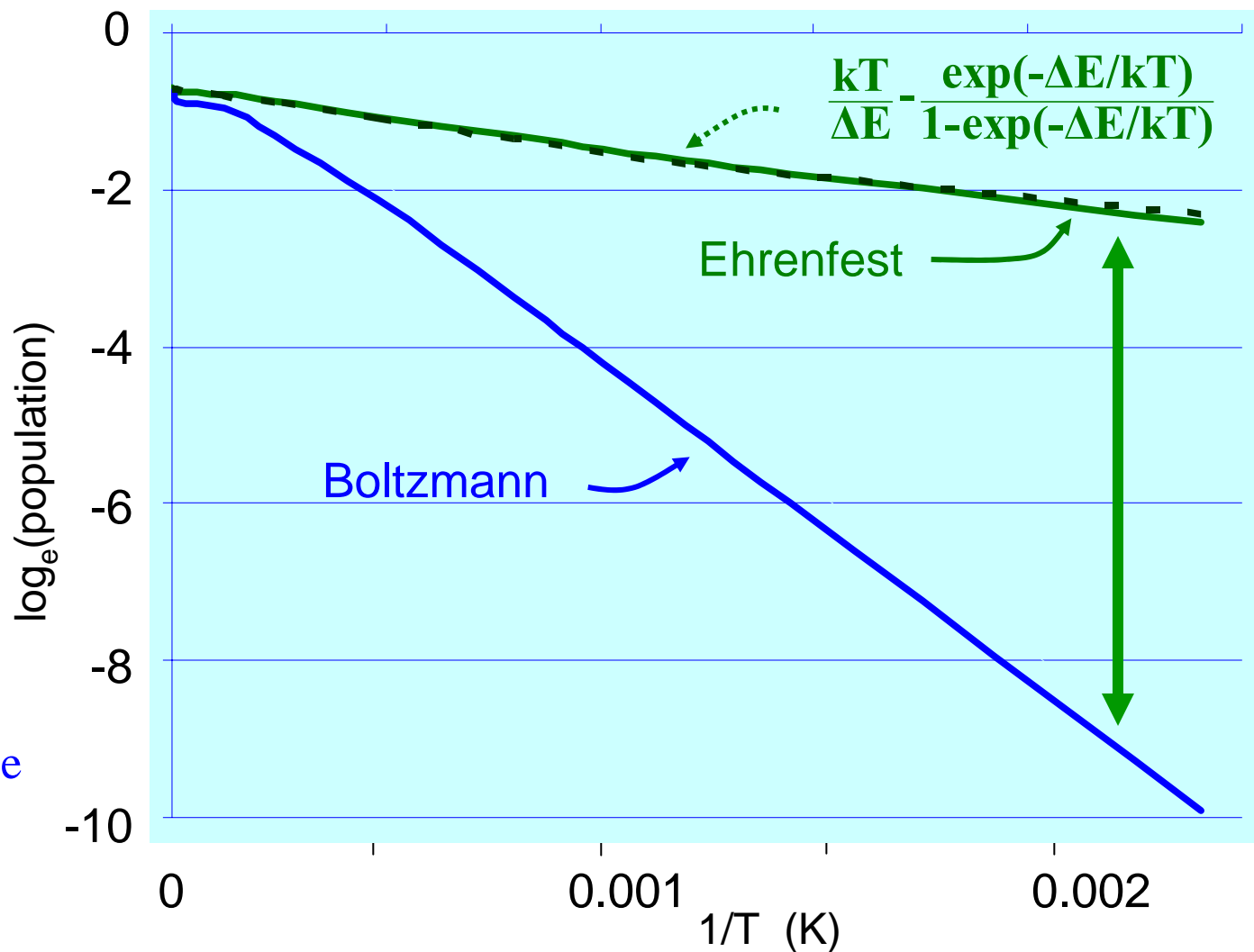
$$\langle |c_2|^2 \rangle = \frac{\exp(-\beta \Delta)}{1 + \exp(-\beta \Delta)}$$



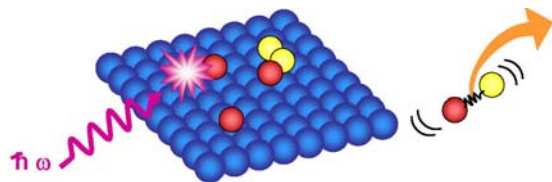
VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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Excited State Population vs. Inverse Temperature



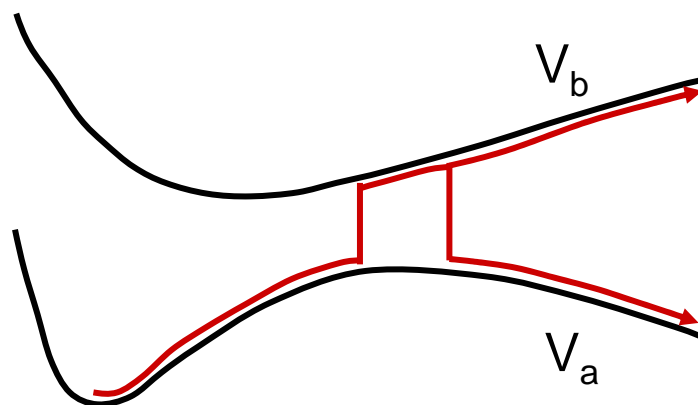
2-state system
 $\Delta E = 34.6 \text{ kJ/mole}$



VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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What about surface hopping ?



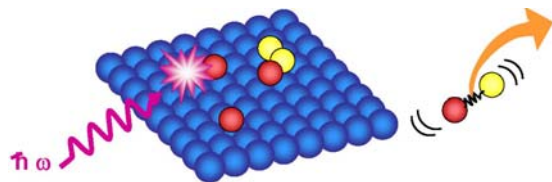
$$dc_j/dt = -\frac{i}{\hbar}V_{jj}c_j - \dot{R} \cdot \sum_i \langle \Phi_j(r;R) | \nabla_R \Phi_i(r;R) \rangle c_i$$

time reversible

$$P_{1 \rightarrow 2} = P_{2 \rightarrow 1} \quad \rightarrow \quad |c_1|^2 = |c_2|^2$$

probabilities of each quantum state are equal: infinite temperature

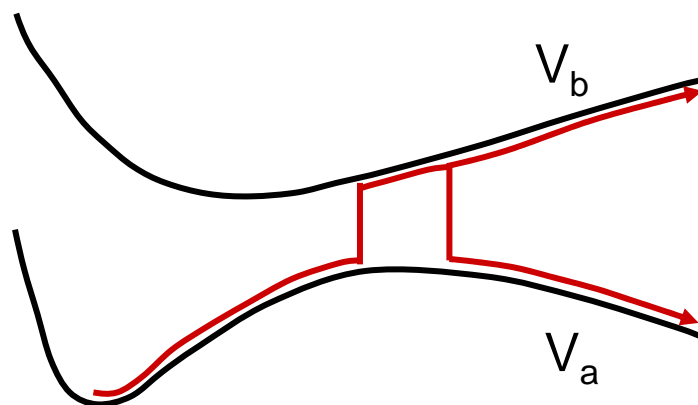
looks bad



VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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What about surface hopping ?

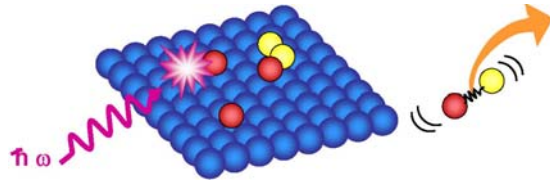


$$dc_j/dt = -\frac{i}{\hbar}V_{jj}c_j - \dot{R} \cdot \sum_i \langle \Phi_j(r;R) | \nabla_R \Phi_i(r;R) \rangle c_i$$

time reversible

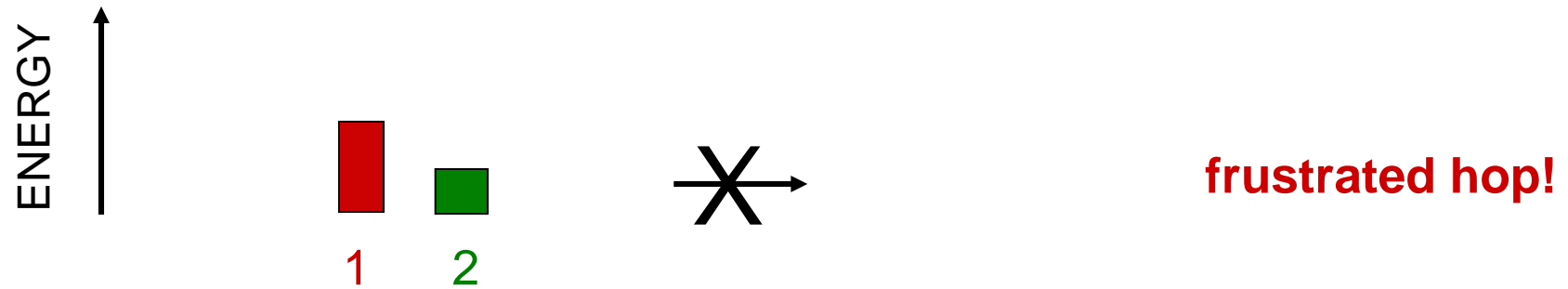
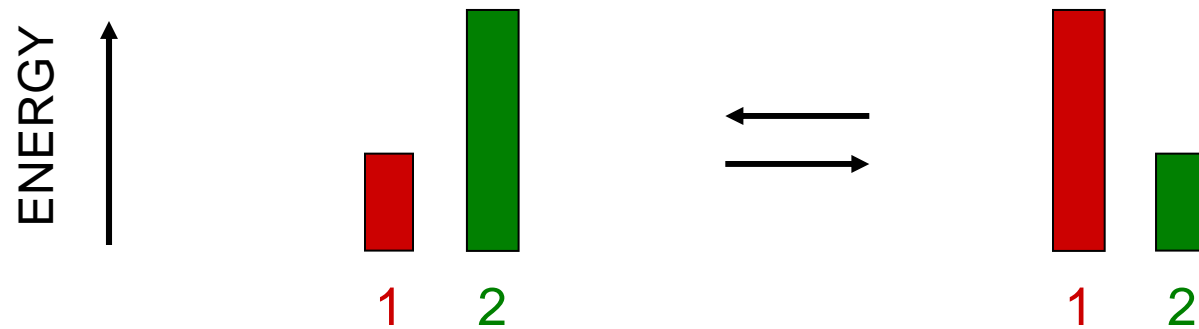
$$P_{1 \rightarrow 2} = P_{2 \rightarrow 1} \quad \rightarrow \quad |c_1|^2 = |c_2|^2$$

but surface hopping probabilities $\neq |c_k|^2$ *frustrated hops*

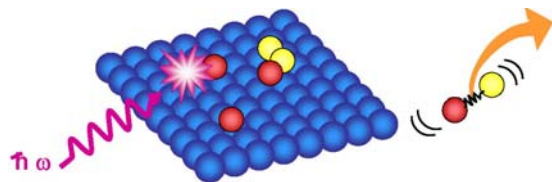


VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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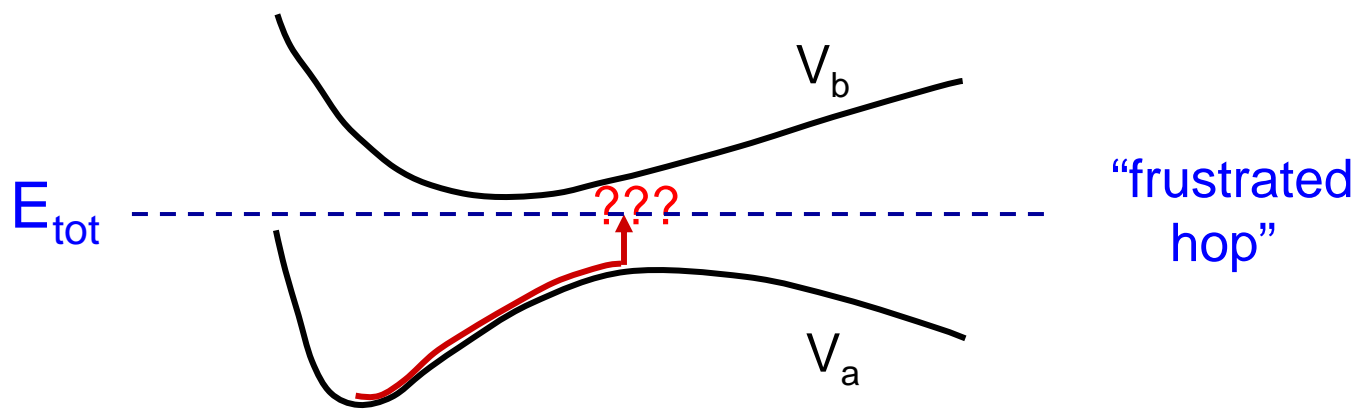


→ More configurations with low energy



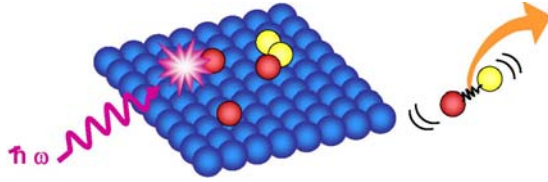
VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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$$P_{1 \rightarrow 2} = \exp(-\Delta E / kT) P_{2 \rightarrow 1}$$

$T_{\text{quantum}} \rightarrow \text{correct}$



VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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$$P_{\text{hop}} = -\Delta|c_k|^2 / |c_k|^2 = \text{fractional decrease of state } k \text{ population}$$

$$= -2\Delta \text{Re}[c_k^* \dot{c}_k] / |c_k|^2$$

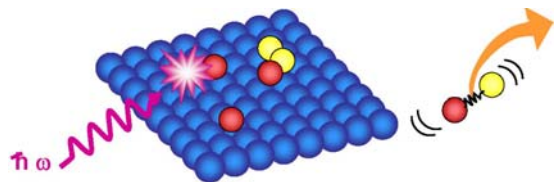
$$dc_j/dt = -\frac{i}{\hbar} V_{jj} c_j - \underbrace{\dot{R} \cdot \sum_i}_{\text{Adiabatic representation}} \langle \Phi_j(r; R) | \nabla_R \Phi_i(r; R) \rangle c_i$$

$$P_{\text{hop}} \propto v$$

$$P_{1 \rightarrow 2} \propto v_1 \rho(v_1) dv_1 \propto v_1 \exp(-\frac{1}{2} m v_1^2 / kT) dv_1$$

$$\propto \exp(-E_1 / kT)$$

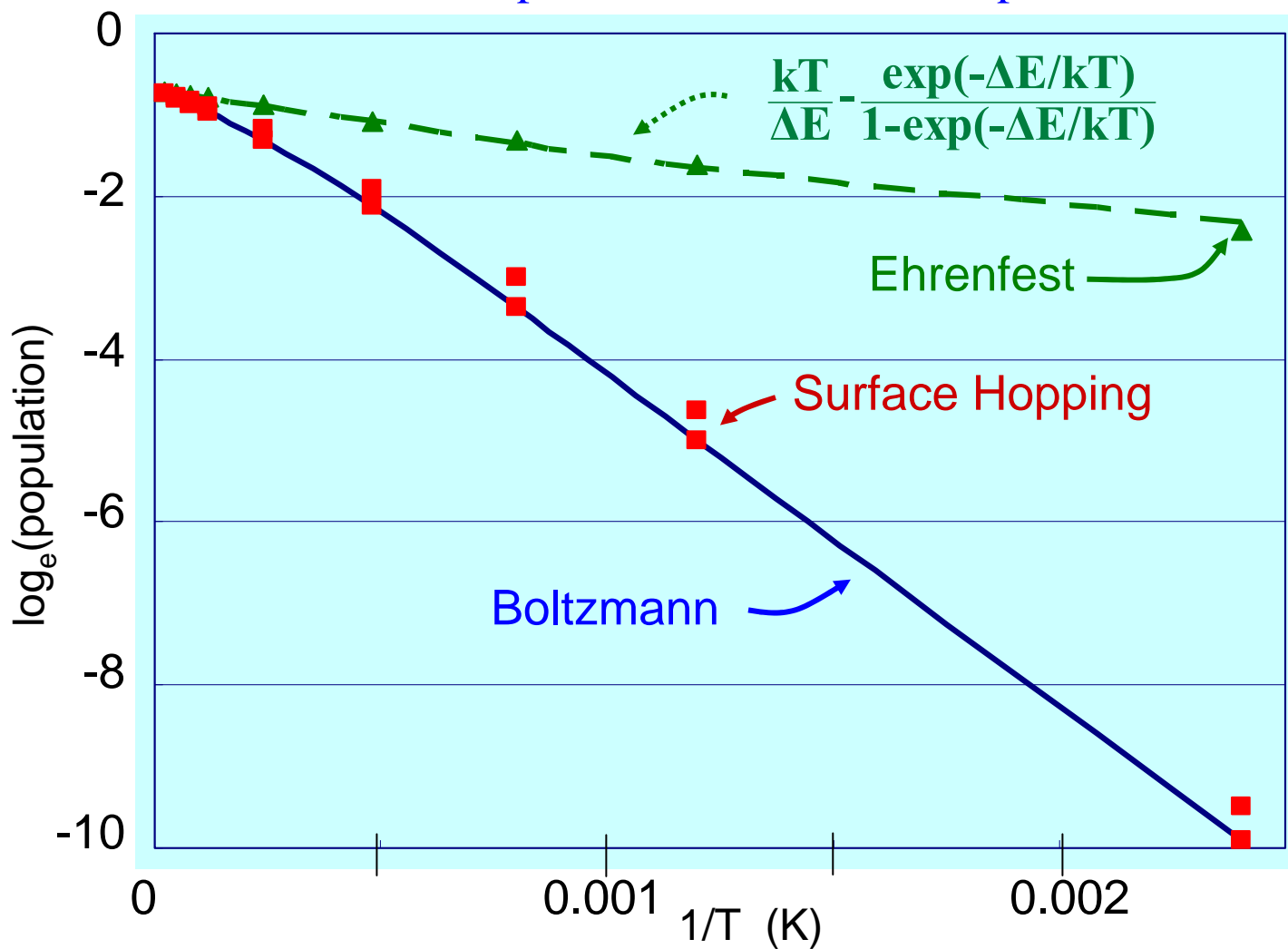
$$\mathcal{N}_1 P_{1 \rightarrow 2} = \mathcal{N}_2 P_{2 \rightarrow 1} \rightarrow \frac{\mathcal{N}_2}{\mathcal{N}_1} = \exp[-(E_2 - E_1) / kT]$$



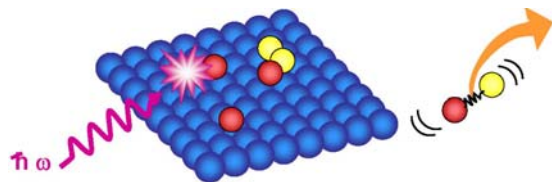
VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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Excited State Population vs. Inverse Temperature



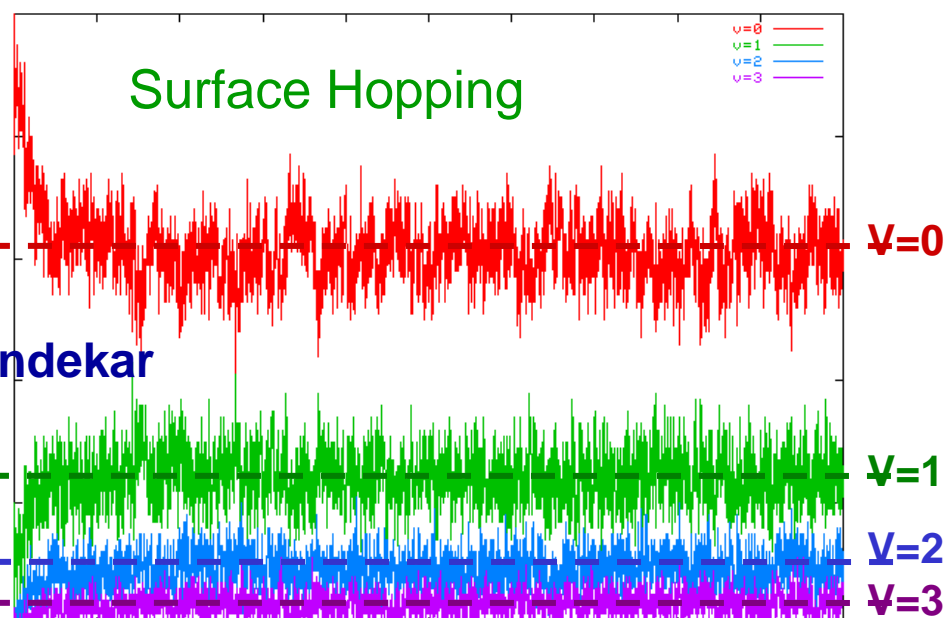
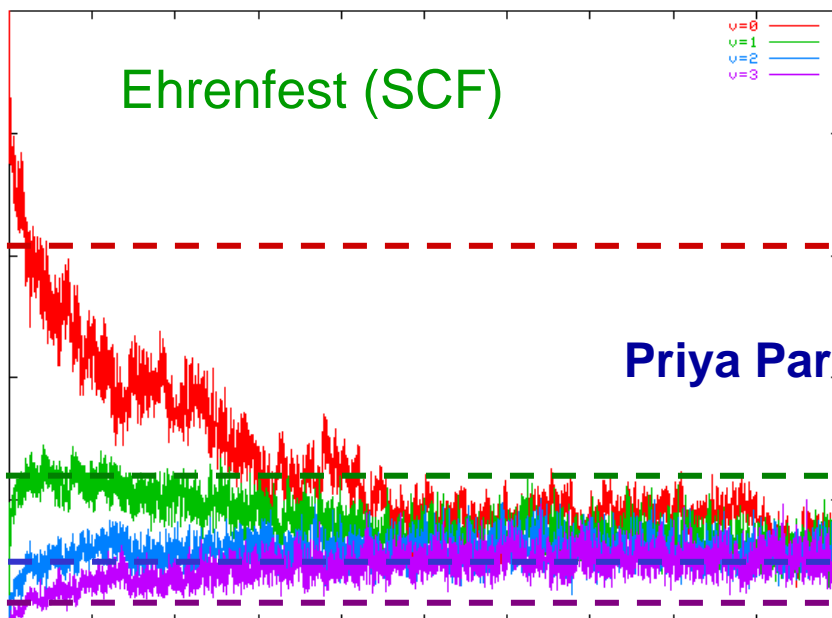
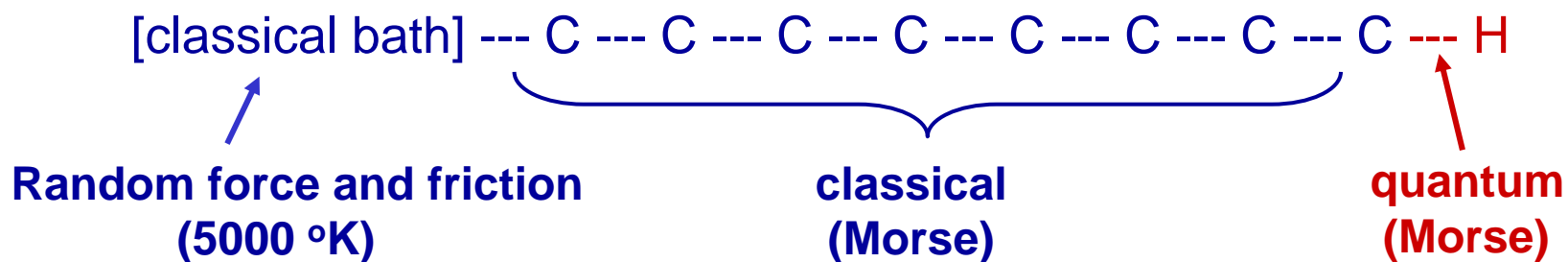
2-State System
 $\Delta E = 34.6$ kJ/mole

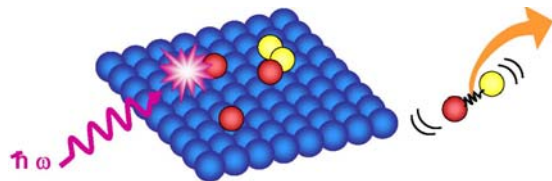


VIII. Equilibrium in Mixed Quantum-Classical Dynamics

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Many Quantum States

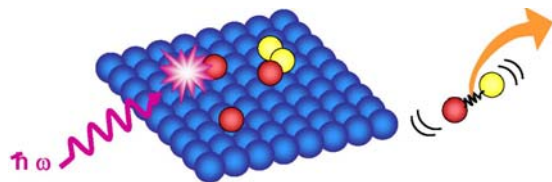




Molecular Dynamics

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- I. The Potential Energy Surface
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- IX. Mixed Quantum-Classical Nuclear Motion**

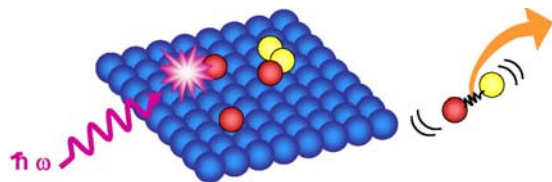


XI. Mixed Quantum-Classical Nuclear Motion

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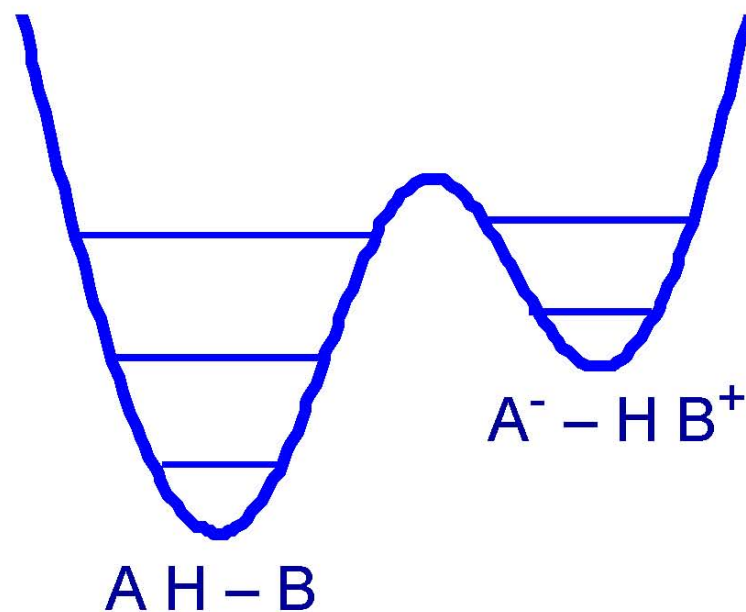
Tenets of Conventional Molecular Dynamics

1. The Born-Oppenheimer Approximation Multiple Electronic States, Metals, ...
2. Classical Mechanical Nuclear Motion Zero Point Motion, Quantized Energy Levels, Tunneling



XI. Mixed Quantum-Classical Nuclear Motion

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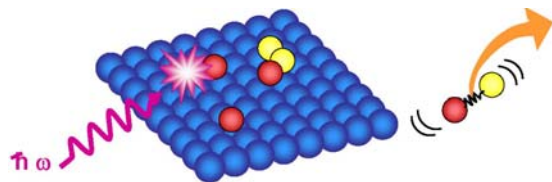


Quantum Effects:

Zero-Point Energy

Quantized Energy Levels

Tunneling



XI. Mixed Quantum-Classical Nuclear Motion

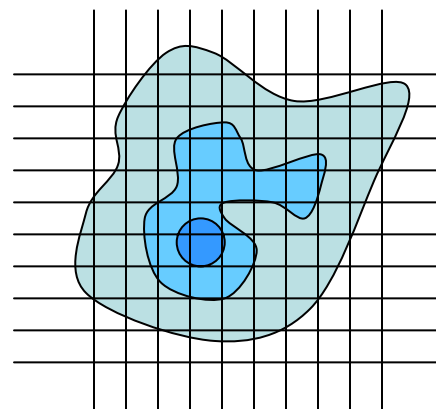
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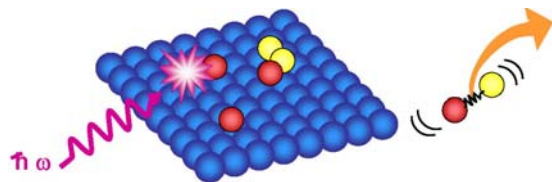
Ultimate Solution:

Treat All Electrons and Nuclei
by Quantum Mechanics

Problem:

Scaling with Size is Prohibitive

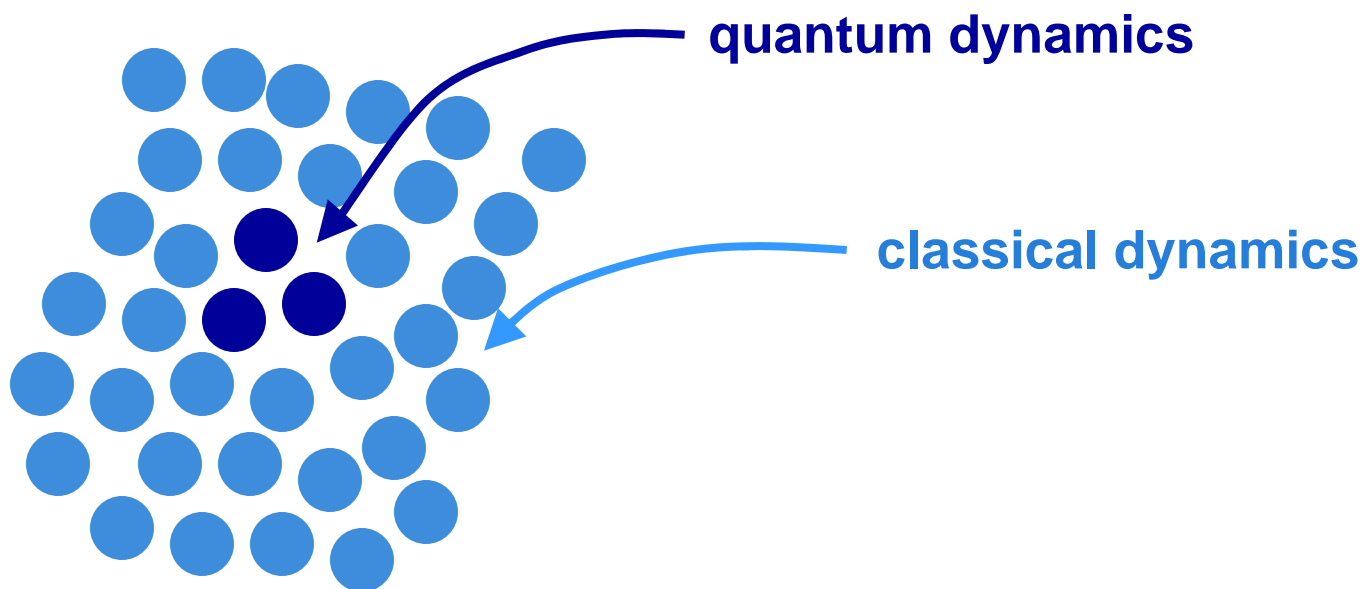


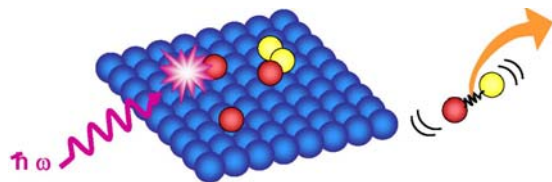


XI. Mixed Quantum-Classical Nuclear Motion

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AN ALTERNATIVE STRATEGY: MIXED QUANTUM-CLASSICAL DYNAMICS





XI. Mixed Quantum-Classical Nuclear Motion

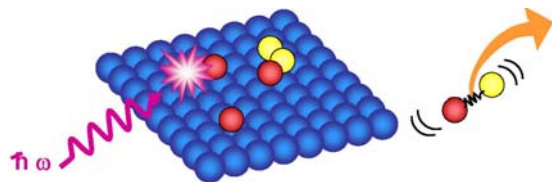
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AN ALTERNATIVE STRATEGY: MIXED QUANTUM-CLASSICAL DYNAMICS

Treat crucial electronic or nuclear degrees of freedom by quantum mechanics, and the remaining nuclei by classical mechanics.

self-consistency

quantum “back-reaction” on classical particles

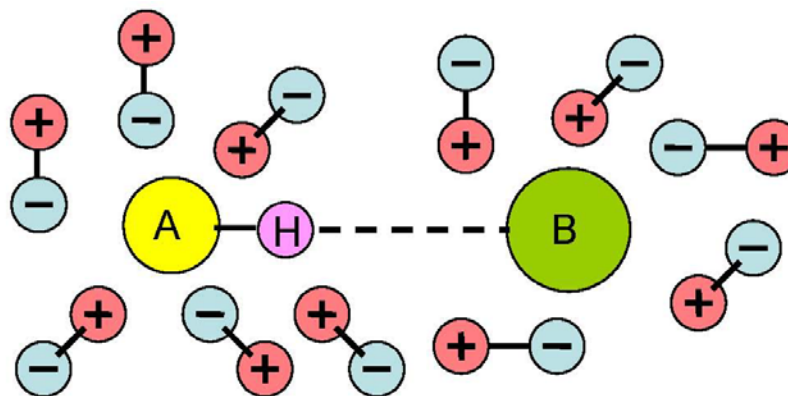


XI. Mixed Quantum-Classical Nuclear Motion

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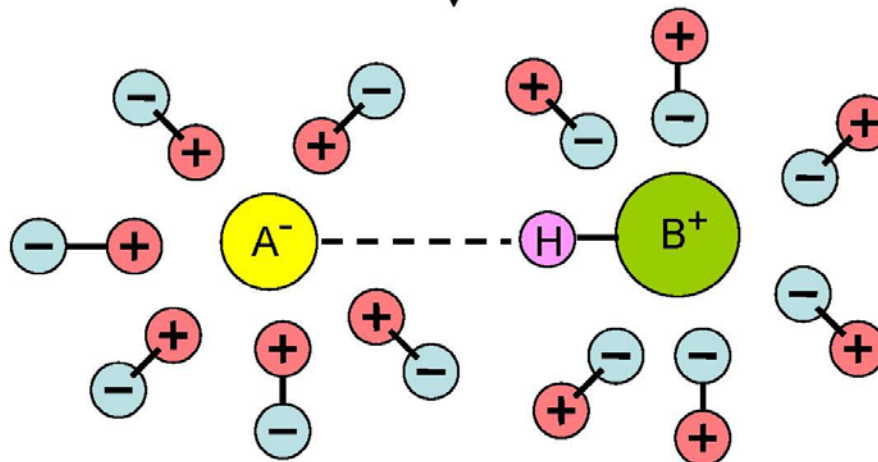
PROTON TRANSFER IN SOLUTION

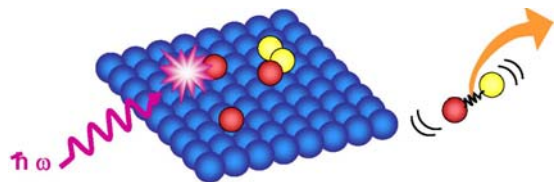
covalent reactant:



self consistency !

ionic product:

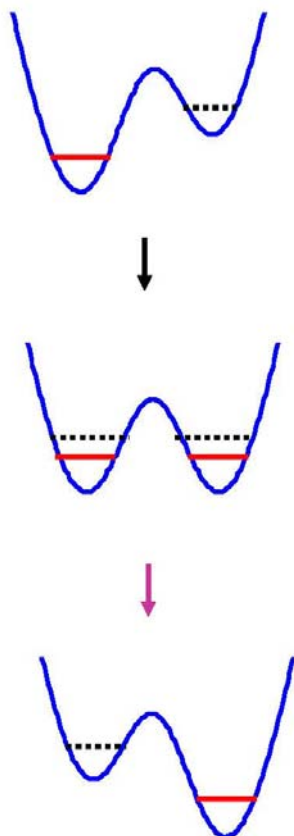




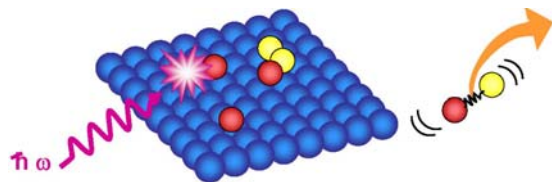
XI. Mixed Quantum-Classical Nuclear Motion

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ADIABATIC vs. NON-ADIABATIC



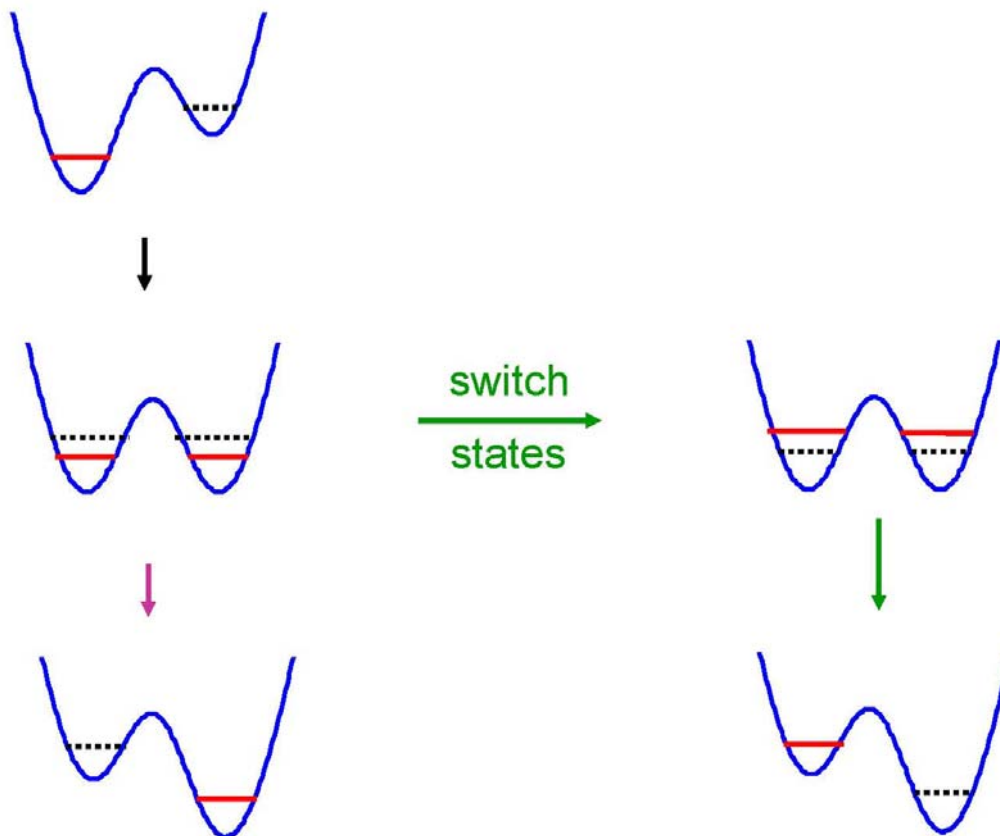
ADIABATIC → REACTION



XI. Mixed Quantum-Classical Nuclear Motion

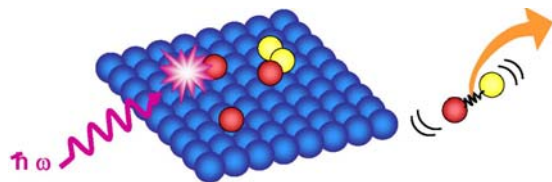
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ADIABATIC vs. NON-ADIABATIC



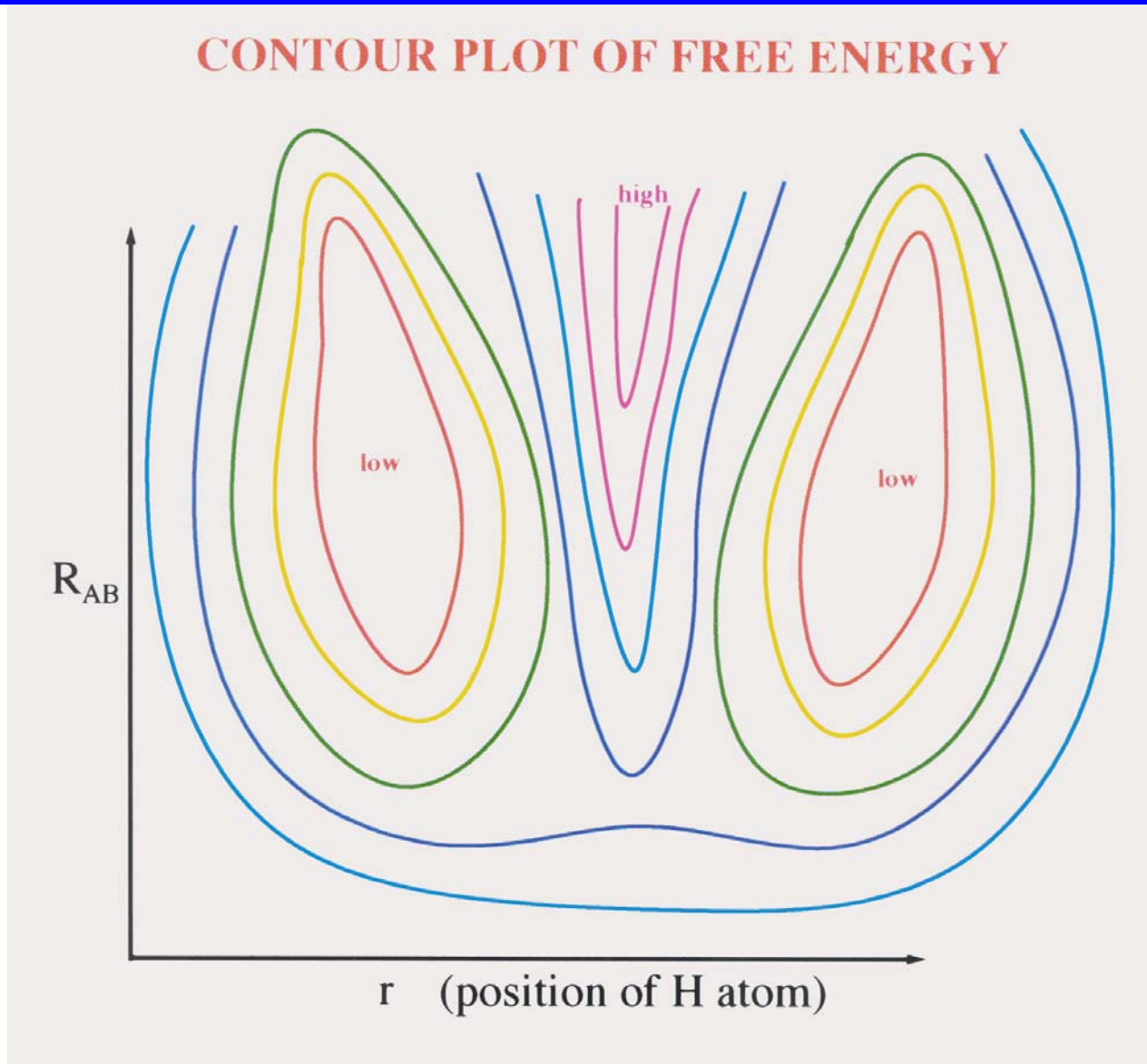
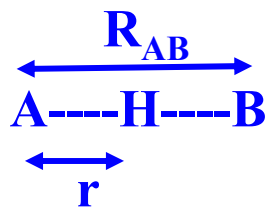
ADIABATIC → REACTION

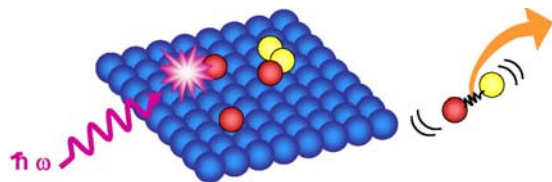
NONADIABATIC → NO REACTION



XI. Mixed Quantum-Classical Nuclear Motion

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XI. Mixed Quantum-Classical Nuclear Motion

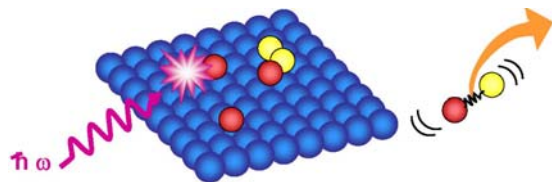
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Proton and Hydride
Transfer in Enzymes:

Sharon Hammes-Schiffer
Penn State University

liver alcohol dehydrogenase
(LADH)

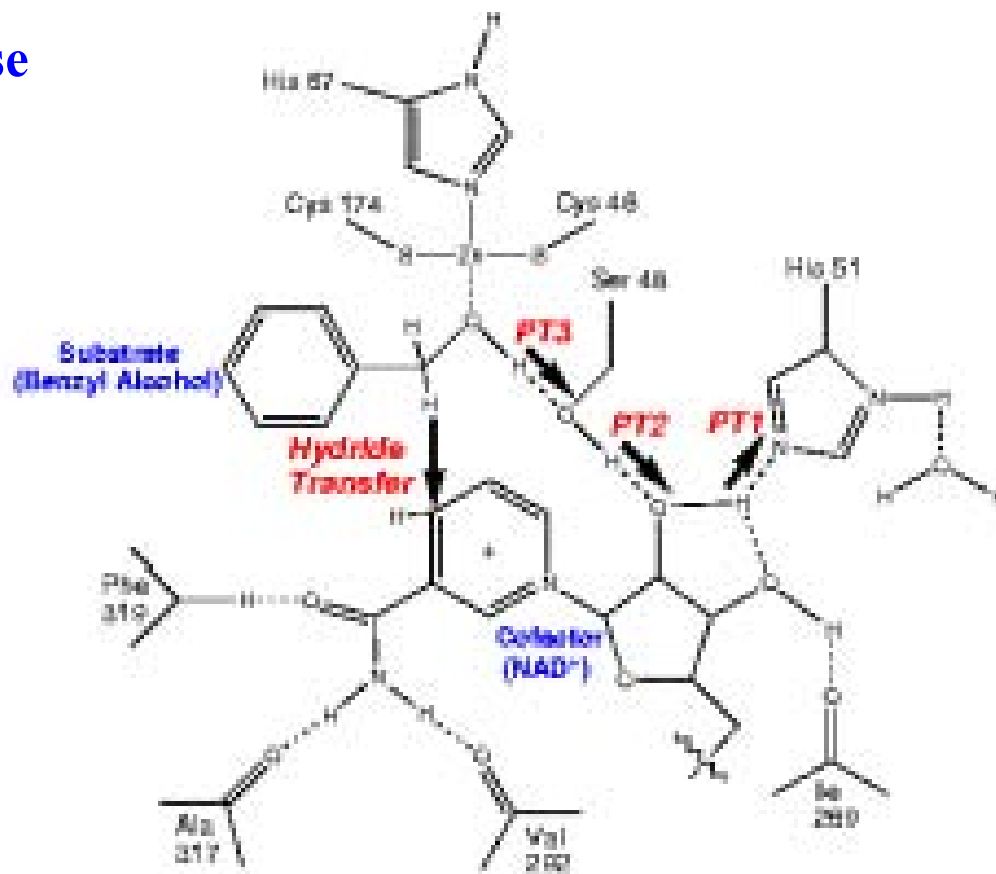




XI. Mixed Quantum-Classical Nuclear Motion

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liver alcohol dehydrogenase



Sharon Hammes-Schiffer