

Chapter 20

Many physical properties of a molecule can be calculated as expectation values of a corresponding quantum mechanical operator. The evaluation of other properties can be formulated in terms of the "response" (i.e., derivative) of the electronic energy with respect to the application of an external field perturbation.

I. Calculations of Properties Other Than the Energy

There are, of course, properties other than the energy that are of interest to the practicing chemist. Dipole moments, polarizabilities, transition probabilities among states, and vibrational frequencies all come to mind. Other properties that are of importance involve operators whose quantum numbers or symmetry indices label the state of interest. Angular momentum and point group symmetries are examples of the latter properties; for these quantities the properties are precisely specified once the quantum number or symmetry label is given (e.g., for a 3P state, the average value of L^2 is $\langle ^3P | L^2 | ^3P \rangle = \hbar^2 1(1+1) = 2\hbar^2$).

Although it may be straightforward to specify what property is to be evaluated, often computational difficulties arise in carrying out the calculation. For some ab initio methods, these difficulties are less severe than for others. For example, to compute the electric dipole transition matrix element $\langle \psi_2 | \mathbf{r} | \psi_1 \rangle$ between two states ψ_1 and ψ_2 , one must evaluate the integral involving the one-electron dipole operator $\mathbf{r} = \sum_j e \mathbf{r}_j - \sum_a Z_a \mathbf{R}_a$; here the first sum runs over the N electrons and the second sum runs over the nuclei whose charges are denoted Z_a . To evaluate such transition matrix elements in terms of the Slater-Condon rules is relatively straightforward as long as ψ_1 and ψ_2 are expressed in terms of Slater determinants involving a single set of orthonormal spin-orbitals. If ψ_1 and ψ_2 have been obtained, for example, by carrying out separate MCSCF calculations on the two states in question, the energy optimized spin-orbitals for one state will not be the same as the optimal spin-orbitals for the second state. As a result, the determinants in ψ_1 and those in ψ_2 will involve spin-orbitals that are not orthonormal to one another. Thus, the SC rules can not immediately be applied. Instead, a transformation of the spin-orbitals of ψ_1 and ψ_2 to a single set of orthonormal functions must be carried out. This then expresses ψ_1 and ψ_2 in terms of new Slater determinants over this new set of orthonormal spin-orbitals, after which the SC rules can be exploited.

In contrast, if ψ_1 and ψ_2 are obtained by carrying out a CI calculation using a single set of orthonormal spin-orbitals (e.g., with ψ_1 and ψ_2 formed from two different

eigenvectors of the resulting secular matrix), the SC rules can immediately be used to evaluate the transition dipole integral.

A. Formulation of Property Calculations as Responses

Essentially all experimentally measured properties can be thought of as arising through the response of the system to some externally applied perturbation or disturbance. In turn, the calculation of such properties can be formulated in terms of the response of the energy E or wavefunction to a perturbation. For example, molecular dipole moments μ are measured, via electric-field deflection, in terms of the change in energy

$$E = \mu \cdot \mathbf{E} + 1/2 \mathbf{E} \cdot \alpha \cdot \mathbf{E} + 1/6 \mathbf{E} \cdot \beta \cdot \mathbf{E} \cdot \mathbf{E} + \dots$$

caused by the application of an external electric field \mathbf{E} which is spatially inhomogeneous, and thus exerts a force

$$\mathbf{F} = - \nabla E$$

on the molecule proportional to the dipole moment (good treatments of response properties for a wide variety of wavefunction types (i.e., SCF, MCSCF, MPPT/MBPT, etc.) are given in Second Quantization Based Methods in Quantum Chemistry, P. Jørgensen and J. Simons, Academic Press, New York (1981) and in Geometrical Derivatives of Energy Surfaces and Molecular Properties, P. Jørgensen and J. Simons, Eds., NATO ASI Series, Vol. 166, D. Reidel, Dordrecht (1985)).

To obtain expressions that permit properties other than the energy to be evaluated in terms of the state wavefunction, the following strategy is used:

1. The perturbation $V = H - H^0$ appropriate to the particular property is identified. For dipole moments (μ), polarizabilities (α), and hyperpolarizabilities (β), V is the interaction of the nuclei and electrons with the external electric field

$$V = - \sum_a Z_a e \mathbf{R}_a \cdot \mathbf{E} - \sum_j e \mathbf{r}_j \cdot \mathbf{E}.$$

For vibrational frequencies, one needs the derivatives of the energy E with respect to deformation of the bond lengths and angles of the molecule, so V is the sum of all changes in the electronic Hamiltonian that arise from displacements \mathbf{R}_a of the atomic centers

$$V = \sum_a (\mathbf{R}_a H) \cdot \mathbf{R}_a.$$

2. A power series expansion of the state energy E , computed in a manner consistent with how \mathbf{R} is determined (i.e., as an expectation value for SCF, MCSCF, and CI wavefunctions or as $\langle \mathbf{R} | H | \mathbf{R} \rangle$ for MPPT/MBPT or as $\langle \exp(-T) H \exp(T) | \mathbf{R} \rangle$ for CC wavefunctions), is carried out in powers of the perturbation V :

$$E = E^0 + E^{(1)} + E^{(2)} + E^{(3)} + \dots$$

In evaluating the terms in this expansion, the dependence of $H = H^0 + V$ and of \mathbf{R} (which is expressed as a solution of the SCF, MCSCF, ..., or CC equations for H not for H^0) must be included.

3. The desired physical property must be extracted from the power series expansion of E in powers of V .

B. The MCSCF Response Case

1. The Dipole Moment

To illustrate how the above developments are carried out and to demonstrate how the results express the desired quantities in terms of the original wavefunction, let us consider, for an MCSCF wavefunction, the response to an external electric field. In this case, the Hamiltonian is given as the conventional one- and two-electron operators H^0 to which the above one-electron electric dipole perturbation V is added. The MCSCF wavefunction \mathbf{R} and energy E are assumed to have been obtained via the MCSCF procedure with $H = H^0 + V$, where \mathbf{R} can be thought of as a measure of the strength of the applied electric field.

The terms in the expansion of $E(\mathbf{R})$ in powers of \mathbf{R} :

$$E = E(\mathbf{R} = 0) + \left(\frac{dE}{d\mathbf{R}} \right)_0 + \frac{1}{2} \left(\frac{d^2E}{d\mathbf{R}^2} \right)_0 + \dots$$

are obtained by writing the total derivatives of the MCSCF energy functional with respect to \mathbf{R} and evaluating these derivatives at $\mathbf{R} = 0$

(which is indicated by the subscript $(\cdot)_0$ on the above derivatives):

$$E(\lambda=0) = \langle \psi(\lambda=0) | H^0 | \psi(\lambda=0) \rangle = E^0,$$

$$\left(\frac{dE}{d\lambda}\right)_0 = \langle \psi(\lambda=0) | V | \psi(\lambda=0) \rangle + 2 \sum_J \left(\frac{C_J}{\lambda}\right)_0 \langle \psi(\lambda=0) | C_J H^0 | \psi(\lambda=0) \rangle$$

$$+ 2 \sum_{i,a} \left(\frac{C_{a,i}}{\lambda}\right)_0 \langle \psi(\lambda=0) | C_{a,i} H^0 | \psi(\lambda=0) \rangle$$

$$+ 2 \sum_{i,a} \left(\frac{C_{a,i}}{\lambda}\right)_0 \langle \psi(\lambda=0) | H^0 | C_{a,i} \psi(\lambda=0) \rangle,$$

and so on for higher order terms. The factors of 2 in the last three terms come through using the hermiticity of H^0 to combine terms in which derivatives of ψ occur.

The first-order correction can be thought of as arising from the response of the wavefunction (as contained in its LCAO-MO and CI amplitudes and basis functions ψ) plus the response of the Hamiltonian to the external field. Because the MCSCF energy functional has been made stationary with respect to variations in the C_J and $C_{i,a}$ amplitudes, the second and third terms above vanish:

$$E/C_J = 2 \langle \psi(\lambda=0) | C_J H^0 | \psi(\lambda=0) \rangle = 0,$$

$$E/C_{a,i} = 2 \langle \psi(\lambda=0) | C_{a,i} H^0 | \psi(\lambda=0) \rangle = 0.$$

If, as is common, the atomic orbital bases used to carry out the MCSCF energy optimization are not explicitly dependent on the external field, the third term also vanishes because $(\psi(\lambda=0) | C_{a,i} H^0 | \psi(\lambda=0)) = 0$. Thus for the MCSCF case, the first-order response is given as the average value of the perturbation over the wavefunction with $\lambda=0$:

$$\left(\frac{dE}{d\lambda}\right)_0 = \langle \psi(\lambda=0) | V | \psi(\lambda=0) \rangle.$$

For the external electric field case at hand, this result says that the field-dependence of the state energy will have a linear term equal to

$$\langle \psi(\lambda=0) | V | \psi(\lambda=0) \rangle = \langle \psi(\lambda=0) | \sum_a Z_a \mathbf{e} \cdot \mathbf{R}_a - \sum_j e \mathbf{r}_j \cdot \mathbf{e} | \psi(\lambda=0) \rangle,$$

where \mathbf{e} is a unit vector in the direction of the applied electric field (the magnitude of the field λ having already been removed in the power series expansion). Since the dipole

moment is determined experimentally as the energy's slope with respect to field strength, this means that the dipole moment is given as:

$$\mu = \langle \psi | \sum_a Z_a e \mathbf{R}_a - \sum_j e \mathbf{r}_j | \psi \rangle.$$

2. The Geometrical Force

These same techniques can be used to determine the response of the energy to displacements \mathbf{R}_a of the atomic centers. In such a case, the perturbation is

$$V = - \sum_a \mathbf{R}_a \cdot \sum_i Z_a e^2 \frac{\mathbf{r}_i - \mathbf{R}_a}{|\mathbf{r}_i - \mathbf{R}_a|^3}$$

$$= - \sum_a Z_a e^2 \mathbf{R}_a \cdot \sum_i (\mathbf{r}_i - \mathbf{R}_a) / |\mathbf{r}_i - \mathbf{R}_a|^3.$$

Here, the one-electron operator $\sum_i (\mathbf{r}_i - \mathbf{R}_a) / |\mathbf{r}_i - \mathbf{R}_a|^3$ is referred to as 'the Hellmann-Feynman' force operator; it is the derivative of the Hamiltonian with respect to displacement of center-a in the x, y, or z direction.

The expressions given above for $E(\lambda=0)$ and $(dE/d\lambda)_0$ can once again be used, but with the Hellmann-Feynman form for V . Once again, for the MCSCF wavefunction, the variational optimization of the energy gives

$$\langle \psi | \sum_j C_j | H^0 | \psi(\lambda=0) \rangle = \langle \psi | \sum_{a,i} C_{a,i} | H^0 | \psi(\lambda=0) \rangle = 0.$$

However, because the atomic basis orbitals are attached to the centers, and because these centers are displaced in forming V , it is no longer true that $(d\psi/d\lambda)_0 = 0$; the variation in the wavefunction caused by movement of the basis functions now contributes to the first-order energy response. As a result, one obtains

$$(dE/d\lambda)_0 = - \sum_a Z_a e^2 \mathbf{R}_a \cdot \langle \psi | \sum_i (\mathbf{r}_i - \mathbf{R}_a) / |\mathbf{r}_i - \mathbf{R}_a|^3 | \psi \rangle$$

$$+ 2 \sum_a \mathbf{R}_a \cdot \left(\frac{d\psi}{d\lambda} \right)_0 \langle \psi | \sum_i | H^0 | \psi(\lambda=0) \rangle.$$

The first contribution to the force

$$\mathbf{F}_a = - Z_a e^2 \langle \psi | \sum_i (\mathbf{r}_i - \mathbf{R}_a) / |\mathbf{r}_i - \mathbf{R}_a|^3 | \psi \rangle$$

$$+ 2 \langle \mathbf{R}_a \rangle_0 \langle \cdot \cdot | \mathbf{H}^0 | \cdot \cdot = 0 \rangle$$

along the x, y, and z directions for center-a involves the expectation value, with respect to the MCSCF wavefunction with $\rho = 0$, of the Hellmann-Feynman force operator. The second contribution gives the forces due to infinitesimal displacements of the basis functions on center-a.

The evaluation of the latter contributions can be carried out by first realizing that

$$= \sum_j C_j \langle \cdot \cdot | J_1 J_2 J_3 \dots J_n \dots J_N | \cdot \cdot \rangle$$

with

$$J_j = \sum_\mu C_{\mu j} \frac{\partial}{\partial \mathbf{R}_a}$$

involves the basis orbitals through the LCAO-MO expansion of the ψ_j s. So the derivatives of the basis orbitals contribute as follows:

$$\langle \mathbf{R}_a \rangle_0 \langle \cdot \cdot | = \sum_j \psi_j \langle \cdot \cdot | C_j C_{\cdot j} \langle \cdot \cdot | J_1 J_2 J_3 \dots \mathbf{R}_a \dots J_N | \cdot \cdot \rangle$$

Each of these factors can be viewed as combinations of CSFs with the same C_j and $C_{\cdot j}$ coefficients as in ψ_j but with the j^{th} spin-orbital involving basis functions that have been differentiated with respect to displacement of center-a. It turns out that such derivatives of Gaussian basis orbitals can be carried out analytically (giving rise to new Gaussians with one higher and one lower l-quantum number).

When substituted into $\langle \mathbf{R}_a \rangle_0 \langle \cdot \cdot | \mathbf{H}^0 | \cdot \cdot = 0 \rangle$, these basis derivative terms yield

$$\langle \mathbf{R}_a \rangle_0 \langle \cdot \cdot | \mathbf{H}^0 | \cdot \cdot = 0 \rangle = \sum_j \psi_j \langle \cdot \cdot | C_j C_{\cdot j} \langle \cdot \cdot | J_1 J_2 J_3 \dots \mathbf{R}_a \dots J_N | \mathbf{H}^0 | \cdot \cdot \rangle,$$

whose evaluation via the Slater-Condon rules is straightforward. It is simply the expectation value of \mathbf{H}^0 with respect to ψ_j (with the same density matrix elements that arise

in the evaluation of ψ_a 's energy) but with the one- and two-electron integrals over the atomic basis orbitals involving one of these differentiated functions:

$$\begin{aligned} & \langle \mu | g | \psi_a \rangle = \langle \mathbf{R}_a | \mu | g | \psi_a \rangle = \langle \mathbf{R}_a | \mu | g | \psi_a \rangle \\ & + \langle \mu | \mathbf{R}_a | g | \psi_a \rangle + \langle \mu | g | \mathbf{R}_a | \psi_a \rangle + \langle \mu | g | \mathbf{R}_a | \psi_a \rangle. \end{aligned}$$

In summary, the force \mathbf{F}_a felt by the nuclear framework due to a displacement of center-a along the x, y, or z axis is given as

$$\mathbf{F}_a = -Z_a e^2 \langle \psi_a | \sum_i (\mathbf{r}_i - \mathbf{R}_a) / |\mathbf{r}_i - \mathbf{R}_a|^3 | \psi_a \rangle + \langle \mathbf{R}_a | \langle \psi_a | \mathbf{H}^0 | \psi_a \rangle \rangle,$$

where the second term is the energy of ψ_a but with all atomic integrals replaced by integral derivatives: $\langle \mu | g | \psi_a \rangle = \langle \mathbf{R}_a | \mu | g | \psi_a \rangle$.

C. Responses for Other Types of Wavefunctions

It should be stressed that the MCSCF wavefunction yields especially compact expressions for responses of E with respect to an external perturbation because of the variational conditions

$$\langle \psi_a | \sum_J C_J | \mathbf{H}^0 | \psi_a \rangle = 0 \quad \langle \psi_a | \sum_{a,i} C_{a,i} | \mathbf{H}^0 | \psi_a \rangle = 0$$

that apply. The SCF case, which can be viewed as a special case of the MCSCF situation, also admits these simplifications. However, the CI, CC, and MPPT/MBPT cases involve additional factors that arise because the above variational conditions do not apply (in the CI case, $\langle \psi_a | \sum_J C_J | \mathbf{H}^0 | \psi_a \rangle = 0$ still applies, but the orbital condition $\langle \psi_a | \sum_{a,i} C_{a,i} | \mathbf{H}^0 | \psi_a \rangle = 0$ does not because the orbitals are not varied to make the CI energy functional stationary).

Within the CC, CI, and MPPT/MBPT methods, one must evaluate the so-called responses of the C_J and $C_{a,i}$ coefficients $(C_J / \lambda)_0$ and $(C_{a,i} / \lambda)_0$ that appear in the full energy response as (see above)

$$2 \sum_J (C_J / \lambda)_0 \langle \psi_a | \sum_J C_J | \mathbf{H}^0 | \psi_a \rangle + 2 \sum_{i,a} (C_{a,i} / \lambda)_0 \langle \psi_a | \sum_{a,i} C_{a,i} | \mathbf{H}^0 | \psi_a \rangle.$$

To do so requires solving a set of response equations that are obtained by differentiating whatever

equations govern the C_I and $C_{a,i}$ coefficients in the particular method (e.g., CI, CC, or MPPT/MBPT) with respect to the external perturbation. In the geometrical derivative case, this amounts to differentiating with respect to x, y, and z displacements of the atomic centers. These response equations are discussed in Geometrical Derivatives of Energy Surfaces and Molecular Properties, P. Jørgensen and J. Simons, Eds., NATO ASI Series, Vol. 166, D. Reidel, Dordrecht (1985). Their treatment is somewhat beyond the scope of this text, so they will not be dealt with further here.

D. The Use of Geometrical Energy Derivatives

1. Gradients as Newtonian Forces

The first energy derivative is called the gradient \mathbf{g} and is the negative of the force \mathbf{F} (with components along the a^{th} center denoted \mathbf{F}_a) experienced by the atomic centers $\mathbf{F} = -\mathbf{g}$. These forces, as discussed in Chapter 16, can be used to carry out classical trajectory simulations of molecular collisions or other motions of large organic and biological molecules for which a quantum treatment of the nuclear motion is prohibitive.

The second energy derivatives with respect to the x, y, and z directions of centers a and b (for example, the x, y component for centers a and b is $H_{ax,by} = (\partial^2 E / \partial x_a \partial y_b)_0$) form the Hessian matrix \mathbf{H} . The elements of \mathbf{H} give the local curvatures of the energy surface along the 3N cartesian directions.

The gradient and Hessian can be used to systematically locate local minima (i.e., stable geometries) and transition states that connect one local minimum to another. At each of these stationary points, all forces and thus all elements of the gradient \mathbf{g} vanish. At a local minimum, the \mathbf{H} matrix has 5 or 6 zero eigenvalues corresponding to translational and rotational displacements of the molecule (5 for linear molecules; 6 for non-linear species) and 3N-5 or 3N-6 positive eigenvalues. At a transition state, \mathbf{H} has one negative eigenvalue, 5 or 6 zero eigenvalues, and 3N-6 or 3N-7 positive eigenvalues.

2. Transition State Rate Coefficients

The transition state theory of Eyring or its extensions due to Truhlar and co-workers (see, for example, D. G. Truhlar and B. C. Garrett, *Ann. Rev. Phys. Chem.* **35**, 159 (1984)) allow knowledge of the Hessian matrix at a transition state to be used to compute a rate coefficient k_{rate} appropriate to the chemical reaction for which the transition state applies.

More specifically, the geometry of the molecule at the transition state is used to compute a rotational partition function Q_{rot}^\ddagger in which the principal moments of inertia I_a , I_b , and I_c (see Chapter 13) are those of the transition state (the \ddagger symbol is, by convention, used to label the transition state):

$$Q_{\text{rot}}^\ddagger = \prod_{n=a,b,c} \sqrt{\frac{8\pi^2 I_n kT}{h^2}},$$

where k is the Boltzmann constant and T is the temperature in $^\circ\text{K}$.

The eigenvalues $\{\nu_i\}$ of the mass weighted Hessian matrix (see below) are used to compute, for each of the $3N-7$ vibrations with real and positive ν_i values, a vibrational partition function that is combined to produce a transition-state vibrational partition function:

$$Q_{\text{vib}}^\ddagger = \prod_{i=1,3-7} \frac{\exp(-h\nu_i/2kT)}{1 - \exp(-h\nu_i/kT)}.$$

The electronic partition function of the transition state is expressed in terms of the activation energy (the energy of the transition state relative to the electronic energy of the reactants) E^\ddagger as:

$$Q_{\text{electronic}}^\ddagger = g^\ddagger \exp(-E^\ddagger/kT)$$

where g^\ddagger is the degeneracy of the electronic state at the transition state geometry.

In the original Eyring version of transition state theory (TST), the rate coefficient k_{rate} is then given by:

$$k_{\text{rate}} = \frac{kT}{h} g^\ddagger \exp(-E^\ddagger/kT) \frac{Q_{\text{rot}}^\ddagger Q_{\text{vib}}^\ddagger}{Q_{\text{reactants}}},$$

where $Q_{\text{reactants}}$ is the conventional partition function for the reactant materials.

For example, in a bimolecular reaction such as:



the reactant partition function

$$Q_{\text{reactants}} = Q_{\text{F}} Q_{\text{H}_2}$$

is written in terms of the translational and electronic (the degeneracy of the ^2P state produces the 2 (3) overall degeneracy factor) partition functions of the F atom

$$Q_{\text{F}} = \frac{2}{h^2} \frac{m_{\text{F}} k T}{h^2}^{3/2} 2 \quad (3)$$

and the translational, electronic, rotational, and vibrational partition functions of the H_2 molecule

$$Q_{\text{H}_2} = \frac{2}{h^2} \frac{m_{\text{H}_2} k T}{h^2}^{3/2} \frac{8}{2h^2} \frac{\exp(-\frac{h}{2} \frac{H_2}{2kT})}{1 - \exp(-\frac{h}{H_2} / kT)}$$

The factor of 2 in the denominator of the H_2 molecule's rotational partition function is the "symmetry number" that must be inserted because of the identity of the two H nuclei.

The overall rate coefficient k_{rate} (with units sec^{-1} because this is a rate per collision pair) can thus be expressed entirely in terms of energetic, geometrical, and vibrational information about the reactants and the transition state. Even within the extensions to Eyring's original model, such is the case. The primary difference in the more modern theories is that the transition state is identified not as the point on the potential energy surface at which the gradient vanishes and there is one negative Hessian eigenvalue. Instead, a so-called variational transition state (see the above reference by Truhlar and Garrett) is identified. The geometry, energy, and local vibrational frequencies of this transition state are then used to compute, must like outlined above, k_{rate} .

3. Harmonic Vibrational Frequencies

It is possible (see, for example, J. Nichols, H. L. Taylor, P. Schmidt, and J. Simons, *J. Chem. Phys.* 92, 340 (1990) and references therein) to remove from \mathbf{H} the zero eigenvalues that correspond to rotation and translation and to thereby produce a Hessian matrix whose eigenvalues correspond only to internal motions of the system. After doing so, the number of negative eigenvalues of \mathbf{H} can be used to characterize the nature of the

stationary point (local minimum or transition state), and \mathbf{H} can be used to evaluate the local harmonic vibrational frequencies of the system.

The relationship between \mathbf{H} and vibrational frequencies can be made clear by recalling the classical equations of motion in the Lagrangian formulation:

$$d/dt(L/ \dot{q}_j) - (L/ q_j) = 0,$$

where q_j denotes, in our case, the $3N$ cartesian coordinates of the N atoms, and \dot{q}_j is the velocity of the corresponding coordinate. Expressing the Lagrangian L as kinetic energy minus potential energy and writing the potential energy as a local quadratic expansion about a point where \mathbf{g} vanishes, gives

$$L = 1/2 \sum_j m_j \dot{q}_j^2 - E(0) - 1/2 \sum_{j,k} q_j H_{j,k} q_k .$$

Here, $E(0)$ is the energy at the stationary point, m_j is the mass of the atom to which q_j applies, and the $H_{j,k}$ are the elements of \mathbf{H} along the x , y , and z directions of the various atomic centers.

Applying the Lagrangian equations to this form for L gives the equations of motion of the q_j coordinates:

$$m_j \ddot{q}_j = - \sum_k H_{j,k} q_k .$$

To find solutions that correspond to local harmonic motion, one assumes that the coordinates q_j oscillate in time according to

$$q_j(t) = q_j \cos(\omega_j t) .$$

Substituting this form for $q_j(t)$ into the equations of motion gives

$$m_j \omega_j^2 q_j = - \sum_k H_{j,k} q_k .$$

Defining

$$q_j' = q_j (m_j)^{1/2}$$

and introducing this into the above equation of motion yields

$$\ddot{q}_j' = - \sum_k H'_{j,k} q_k' ,$$

where

$$H'_{j,k} = H_{j,k} (m_j m_k)^{-1/2}$$

is the so-called mass-weighted Hessian matrix.

The squares of the desired harmonic vibrational frequencies ω_j^2 are thus given as eigenvalues of the mass-weighted Hessian \mathbf{H}' :

$$\mathbf{H}' \mathbf{q}' = - \omega^2 \mathbf{q}'$$

The corresponding eigenvector, $\{q'_i, j\}$ gives, when multiplied by $m_j^{-1/2}$, the atomic displacements that accompany that particular harmonic vibration. At a transition state, one of the ω^2 will be negative and $3N-6$ or $3N-7$ will be positive.

4. Reaction Path Following

The Hessian and gradient can also be used to trace out 'streambeds' connecting local minima to transition states. In doing so, one utilizes a local harmonic description of the potential energy surface

$$E(\mathbf{x}) = E(\mathbf{0}) + \mathbf{x} \cdot \mathbf{g} + 1/2 \mathbf{x} \cdot \mathbf{H} \cdot \mathbf{x} + \dots,$$

where \mathbf{x} represents the (small) step away from the point $\mathbf{x}=\mathbf{0}$ at which the gradient \mathbf{g} and Hessian \mathbf{H} have been evaluated. By expressing \mathbf{x} and \mathbf{g} in terms of the eigenvectors \mathbf{v} of \mathbf{H}

$$\mathbf{H} \mathbf{v} = - \omega^2 \mathbf{v} ,$$

$$\mathbf{x} = \sum_i \langle \mathbf{v}_i | \mathbf{x} \rangle \mathbf{v}_i = \sum_i x_i \mathbf{v}_i ,$$

$$\mathbf{g} = \sum_i \langle \mathbf{v}_i | \mathbf{g} \rangle \mathbf{v}_i = \sum_i g_i \mathbf{v}_i ,$$

the energy change $E(\mathbf{x}) - E(\mathbf{0})$ can be expressed in terms of a sum of independent changes along the eigendirections:

$$E(\mathbf{x}) - E(\mathbf{0}) = \left[x_1 g_1 + \frac{1}{2} x_1^2 \right] + \dots$$

Depending on the signs of g_i and of ω_i , various choices for the displacements x_i will produce increases or decreases in energy:

1. If ω_i is positive, then a step x_i 'along' g_i (i.e., one with $x_i g_i$ positive) will generate an energy increase. A step 'opposed to' g_i will generate an energy decrease if it is short enough that $x_i g_i$ is larger in magnitude than $\frac{1}{2} x_i^2 \omega_i^2$, otherwise the energy will increase.
2. If ω_i is negative, a step opposed to g_i will generate an energy decrease. A step along g_i will give an energy increase if it is short enough for $x_i g_i$ to be larger in magnitude than $\frac{1}{2} x_i^2 \omega_i^2$, otherwise the energy will decrease.

Thus, to proceed downhill in all directions (such as one wants to do when searching for local minima), one chooses each x_i in opposition to g_i and of small enough length to guarantee that the magnitude of $x_i g_i$ exceeds that of $\frac{1}{2} x_i^2 \omega_i^2$ for those modes with $\omega_i > 0$. To proceed uphill along a mode with $\omega_i < 0$ and downhill along all other modes with $\omega_i > 0$, one chooses x_i 'along' g_i with x_i short enough to guarantee that $x_i g_i$ is larger in magnitude than $\frac{1}{2} x_i^2 \omega_i^2$, and one chooses the other x_j opposed to g_j and short enough that $x_j g_j$ is larger in magnitude than $\frac{1}{2} x_j^2 \omega_j^2$.

Such considerations have allowed the development of highly efficient potential energy surface 'walking' algorithms (see, for example, J. Nichols, H. L. Taylor, P. Schmidt, and J. Simons, *J. Chem. Phys.* 92, 340 (1990) and references therein) designed to trace out streambeds and to locate and characterize, via the local harmonic frequencies, minima and transition states. These algorithms form essential components of most modern ab initio, semi-empirical, and empirical computational chemistry software packages.

II. Ab Initio, Semi-Empirical and Empirical Force Field Methods

A. *Ab Initio* Methods

Most of the techniques described in this Chapter are of the ab initio type. This means that they attempt to compute electronic state energies and other physical properties, as functions of the positions of the nuclei, from first principles without the use or knowledge of experimental input. Although perturbation theory or the variational method may be used to generate the working equations of a particular method, and although finite atomic orbital basis sets are nearly always utilized, these approximations do not involve 'fitting' to known experimental data. They represent approximations that can be systematically improved as the level of treatment is enhanced.

B. Semi-Empirical and Fully Empirical Methods

Semi-empirical methods, such as those outlined in Appendix F, use experimental data or the results of ab initio calculations to determine some of the matrix elements or integrals needed to carry out their procedures. Totally empirical methods attempt to describe the internal electronic energy of a system as a function of geometrical degrees of freedom (e.g., bond lengths and angles) in terms of analytical 'force fields' whose parameters have been determined to 'fit' known experimental data on some class of compounds. Examples of such parameterized force fields were presented in Section III. A of Chapter 16.

C. Strengths and Weaknesses

Each of these tools has advantages and limitations. Ab initio methods involve intensive computation and therefore tend to be limited, for practical reasons of computer time, to smaller atoms, molecules, radicals, and ions. Their CPU time needs usually vary with basis set size (M) as at least M^4 ; correlated methods require time proportional to at least M^5 because they involve transformation of the atomic-orbital-based two-electron integrals to the molecular orbital basis. As computers continue to advance in power and memory size, and as theoretical methods and algorithms continue to improve, ab initio techniques will be applied to larger and more complex species. When dealing with systems in which qualitatively new electronic environments and/or new bonding types arise, or excited electronic states that are unusual, ab initio methods are essential. Semi-empirical or empirical methods would be of little use on systems whose electronic properties have not been included in the data base used to construct the parameters of such models.

On the other hand, to determine the stable geometries of large molecules that are made of conventional chemical units (e.g., CC, CH, CO, etc. bonds and steric and torsional interactions among same), fully empirical force-field methods are usually quite reliable and computationally very fast. Stable geometries and the relative energetic stabilities of various conformers of large macromolecules and biopolymers can routinely be predicted using such tools if the system contains only conventional bonding and common chemical building blocks. These empirical potentials usually do not contain sufficient flexibility (i.e., their parameters and input data do not include enough knowledge) to address processes that involve rearrangement of the electronic configurations. For example, they can not treat:

1. Electronic transitions, because knowledge of the optical oscillator strengths and of the energies of excited states is absent in most such methods;
2. Concerted chemical reactions involving simultaneous bond breaking and forming, because to do so would require the force-field parameters to evolve from those of the reactant bonding to those for the product bonding as the reaction proceeds;
3. Molecular properties such as dipole moment and polarizability, although in certain fully empirical models, bond dipoles and lone-pair contributions have been incorporated (although again only for conventional chemical bonding situations).

Semi-empirical techniques share some of the strengths and weaknesses of ab initio and of fully empirical methods. They treat at least the valence electrons explicitly, so they are able to address questions that are inherently electronic such as electronic transitions, dipole moments, polarizability, and bond breaking and forming. Some of the integrals involving the Hamiltonian operator and the atomic basis orbitals are performed ab initio; others are obtained by fitting to experimental data. The computational needs of semi-empirical methods lie between those of the ab initio methods and the force-field techniques. As with the empirical methods, they should never be employed when qualitatively new electronic bonding situations are encountered because the data base upon which their parameters were determined contain, by assumption, no similar bonding cases.